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1 **Concepts for efficient preparation of particulate polymer carrier systems by droplet-**
2 **based microfluidics**

3

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9

10 **Abstract.**

11 Droplet-based microfluidics has grown out of its infancy as technical solutions became
12 available for a broad community of researchers aiming at highly defined structures of polymer-
13 based drug carrier systems. While the beauty of obtained particles and the precision of their
14 (continuous) production may be very fascinating from a scientific perspective, microfluidics is
15 further developing towards the use in production processes. This review summarizes recent
16 concepts and developments in droplet-based microfluidics covering theoretical aspects of the
17 operation principle as well as approaches to increased throughput and thus to enable efficient
18 production. The application of microfluidic templating for preparing functional polymer particles
19 including dispersions of preformed polymers, multicompartment particles and the use of
20 template droplets as microreactors for carrier synthesis are also included. When operated at
21 high-throughput, in a continuous process and with excellent control over particle properties,
22 microfluidics may become a preparation technique for particulate carriers competitive to batch
23 emulsification not only in research but also for commercial fabrication, e.g., of individualized,
24 patient-specific formulations.

25

26 **Keywords:** droplet-based microfluidics, high-throughput, polymer particles, controlled drug
27 **release**

28

29 **1. Introduction**

30 The preparation of particulate carrier systems by conventional batch emulsification of
31 polymer/drug combinations has been the basis for injectable sustained release formulations,
32 which are successfully used in the clinics for local and systemic treatment of diseases
33 (Wischke and Schwendeman, 2008) (Ramazani et al., 2016) (Lee et al., 2016). As part of these
34 batch production processes, the dispersion of polymer solutions into nascent particles is
35 performed by application of high (non-uniform) shear stresses and/or by phase separation. As
36 a consequence of non-uniform/statistic events, the sizes and morphologies of individual
37 particles within a batch can substantially vary from each other, particularly when
38 multicomponent carriers are to be prepared (Qi et al., 2019). In fact, particle sizes and drug
39 distribution within polymeric carriers can strongly affect the drug release features (Park et al.,
40 2019). This calls for technologies that would allow better homogeneity of particles within a
41 produced batch, a reproducible tailoring of particle sizes, and the design of well-defined internal
42 structures in case of multicomponent carrier matrices.

43 Microfluidic techniques opened up a multitude of opportunities to enhance and accelerate the
44 research in pharmaceutical sciences, including drug discovery, drug synthesis and in vitro
45 assays (Nys and Fillet, 2018). For drug formulation into particulate carrier systems, both phase
46 separation/self-assembly processes and dispersion processes can be realized in microfluidic
47 channels, which allow the embedding of drug molecules into polymeric matrices (Liu et al.,
48 2017) (Damiati et al., 2018).

49 In this review, the focus will be on microfluidic dispersion of different phases into droplets.
50 These droplets incorporating either polymerizable monomers, functionalized telechelics, or
51 preformed polymers can subsequently be transferred by crosslinking and/or solvent removal
52 into polymer particles of highly defined structure and size, typically in the micrometer range (Li
53 et al., 2008a). Given the fact that some materials are challenging to disperse, customized
54 devices and emulsification concepts may allow preparing particles also from such materials
55 (Abate et al., 2011).

56 Aside from the experimental beauty and opportunities arising for research, the commercial
57 applicability of droplet-based microfluidics for preparing polymeric carrier systems may depend
58 on a number of additional aspects. This includes technical boundaries of the technology,
59 including the question of throughput, as well as financial aspects regarding investments and
60 production costs even for high-value products. In this respect, the rate of droplet formation
61 remains key to productivity of microfluidic particle preparation. It is linked to process
62 parameters, device design, and, importantly, the properties of the materials to be dispersed.
63 Still, as literally monodisperse droplets of two-phase or multiple phase emulsions can be
64 continuously templated in microfluidic channels (Datta et al., 2014) (Wang et al., 2011) and a
65 toolbox of concepts to increase throughput has become available, droplet-based microfluidics
66 can be expected to be increasingly used both in academic research and, as currently ongoing,
67 in pharmaceutical development.

68 In this review as part of the Special Issue '*Europe*' of *International Journal of Pharmaceutics*,
69 microfluidic concepts for preparing particulate carrier systems will be summarized, highlighting
70 their significant development towards applications in the last 5 years. While a substantial set
71 of work in the field of droplet-based microfluidics including many ground breaking findings
72 originate from outside of Europe, a research community as well as industrial contributing
73 players have established in Europe over the last one or two decades. At current, start-up
74 incubators on microfluidics are growing. European companies have made various
75 instrumentation commercially available for research and development, including microfluidic
76 chips, pressure driven pumps, and even full-pack microfluidic production stations for polymer
77 particles. By acknowledging the research conducted by an increasing number of European
78 and international research groups, this review will concentrate on recent concepts and
79 developments in droplet-based microfluidics covering some theoretical aspects of the
80 operation principle, strategies to increased throughput, and the application of microfluidic
81 templating for preparing particulate drug carriers.

82

83 2. Operation principle of common microfluidic droplet generators

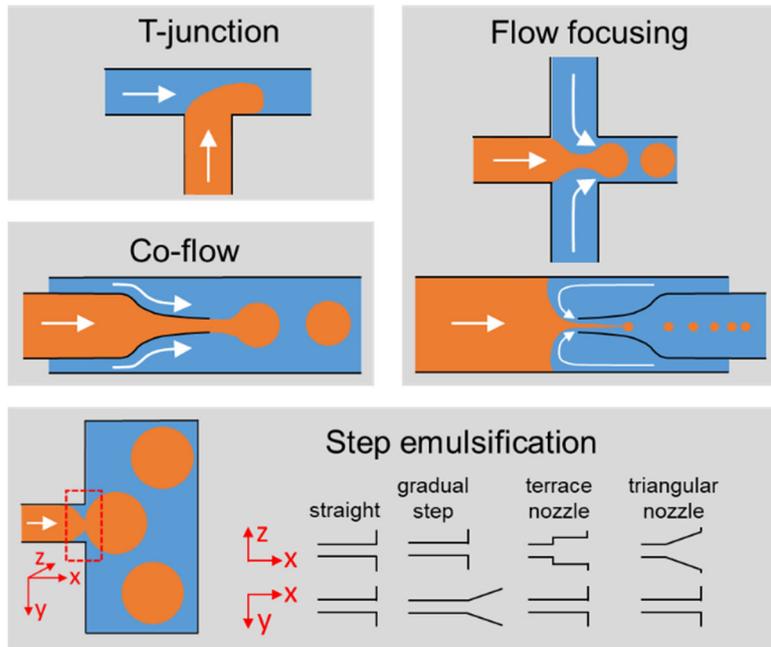
84 In microfluidic droplet generators, droplets are produced one after the other. This principle is
85 different from classical mixers, e.g., rotor-stator homogenizers, where numerous droplets are
86 generated at the same time in different positions of the vessel. In droplet-based microfluidics,
87 the dispersion of two or more immiscible phases is conducted in channels of well-defined
88 geometries under continuous flow. This flow condition is characterized by balanced viscous
89 forces, inertial forces, interfacial tensions and, in many cases, contributions by buoyancy. The
90 flow conditions of two phase systems in microfluidic capillaries can be described by a number
91 of dimensionless numbers such as the capillary number Ca and the Weber number We , as will
92 be illustrated in later sections of this review. Those numbers can be determined from the
93 interfacial tension σ between the two phases, the channel diameter d , the viscosity η , the
94 velocity v , and the density ρ of the respective fluid.

$$95 \quad Ca = \frac{\eta \cdot v}{\sigma} \quad (1)$$

$$96 \quad We = \frac{\rho \cdot v^2 \cdot d}{\sigma} \quad (2)$$

97 The Ca represents the viscous shear forces in relation to surface tension. Ca is of interest for
98 the outer fluid, also named continuous phase, as it acts on the phase that should be dispersed
99 into droplets (inner fluid, dispersed phase). The We of the inner fluid represents the inertial
100 forces in relation to surface tension. A more extensive description of physical parameters,
101 dimensionless numbers and their contribution to droplet formation and droplet sizing in relation
102 to capillary dimensions can be found in previous reviews (Zhu and Wang, 2017).

103



104

105 **Fig. 1:** Schemes of droplet formation at exemplary channel geometries. Blue: Continuous phase; Orange:
 106 Dispersed phase.

107

108 Depending on the channel geometry and operation mode, different droplet formation
 109 mechanisms/regimes are involved, of which a few will be briefly summarized here (Fig. 1). In
 110 microchannels with T-shaped junctions, the dispersed phase is typically fed perpendicularly
 111 into a straight channel that is well-wetted by the continuous phase. At high v and thus high Ca ,
 112 droplet formation mainly bases on shear stress by the cross flow of the continuous phase.
 113 However, more commonly, T-junctions are employed in a slow and well controlled operation
 114 mode below a critical value of Ca , where the thread of the dispersed phase introduced at T-
 115 junctions results in a pressure drop along this channel section. Under these conditions, the
 116 droplet break-up is eventually dominated by a squeezing of the neck of the thread by the
 117 continuous phase (Garstecki et al., 2006).

118 Other widely employed channel geometries are axisymmetric co-flow as well as focused-flow
 119 geometries, which can be constructed, e.g., from glass capillaries (Shah et al., 2008) (Martino
 120 et al., 2014) (Benson et al., 2013). Additionally, cross junctions or ψ -shaped junctions with a
 121 flow focusing effect can be obtained, e.g., by soft lithography in polydimethylsiloxane (PDMS)
 122 (Tang and Whitesides, 2010). In focused flow devices, the thread of disperse phase is

123 enveloped by the continuous phase, which practically has the same flow direction at least at
124 the site of dispersion, even if feeding channels for the continuous phase may be arranged in
125 different angles for practical reasons. Depending on the employed flow rates, different
126 operation principles can be distinguished: At slow flow, channel plugging and squeezing of the
127 dispersed phase thread by the surrounding continuous phase can account for emulsification
128 (Romero and Abate, 2012). When operated in the dripping regime, droplet break-up occurs by
129 shear forces acting on the dispersed phase thread at the junction or orifice, where the
130 immiscible phases meet. This dripping mode, which is characterized by low Ca and We , is
131 most commonly used to produce practically monodisperse particles. They can be size-
132 controlled within certain boundaries by flow rates and channel dimensions. At high flow rates
133 in the jetting regime, a long thread of the disperse phase is formed and droplet formation occurs
134 further downstream, which is typically less controlled as mainly driven by Rayleigh-Plateau
135 instabilities. In any case, narrow orifices in focused-flow devices can assist in emulsification by
136 increased shear and thread elongation. Furthermore, focused flow set-ups allow for droplet
137 sizes smaller than the orifice diameter in both coaxial glass capillary assemblies (Shah et al.,
138 2008) and PDMS devices (Tran et al., 2014).

139 In step emulsification, droplets can be formed at low frequencies without relevant contributions
140 of shear forces by the continuous phase. Here, the disperse phase enters from a feeding
141 channel through an opening of various possible geometries (e.g., straight channel (Mittal et al.,
142 2014), terrace nozzle (Kobayashi et al., 2012), triangular nozzle (Amstad et al., 2016)) into a
143 much wider channel, representing a step towards reduced spatial confinement. As droplet size
144 can rapidly increase at the orifice, a jump in Laplace pressure of the liquid thread occurs,
145 eventually causing its upstream necking and break-up through Rayleigh-Plateau instabilities
146 (Dangla et al., 2013).

147 The concepts introduced above can be categorized as 'passive' droplet formation techniques,
148 where physical principles such as interfacial tension result in fluid instability and drive the
149 break-up of thin threads of solvents into droplets. It should be noted that in addition to those

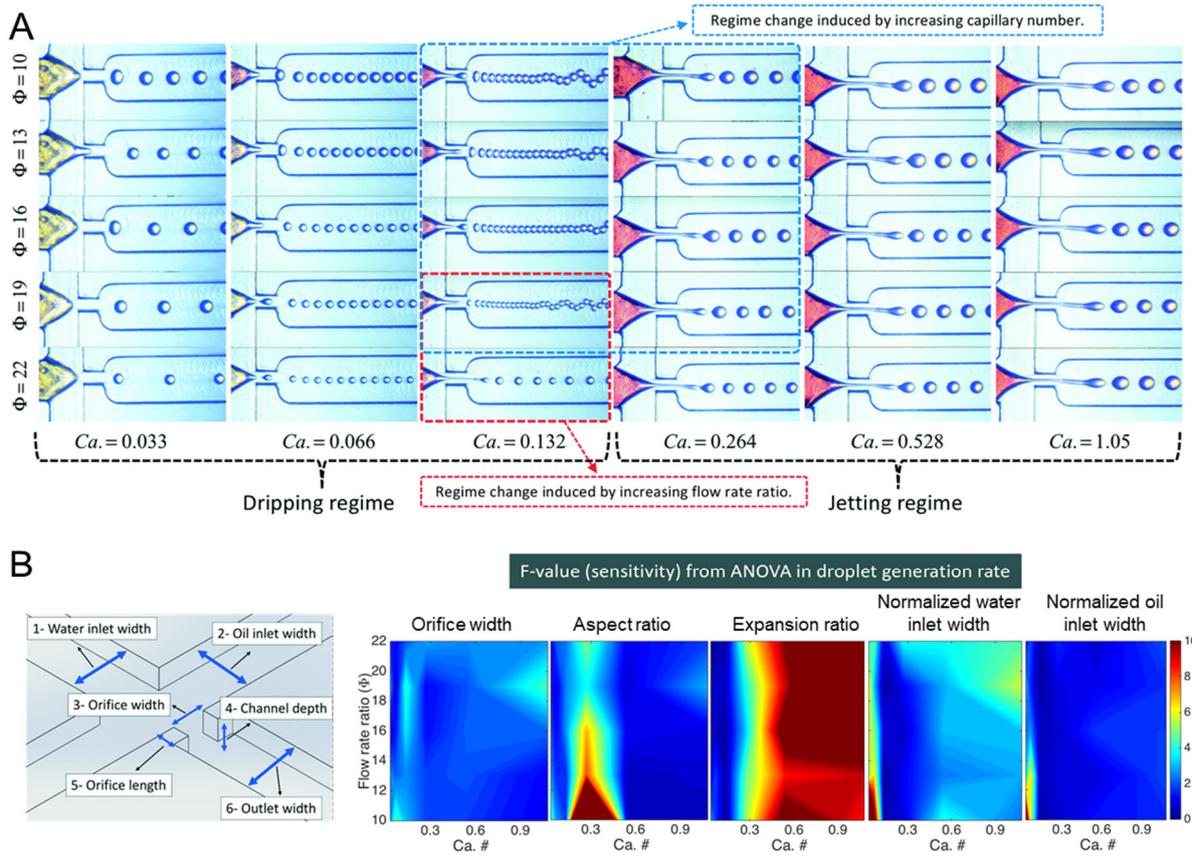
150 concepts, also 'active' principles based on additional energy triggers (e.g. electric, magnetic,
151 thermal) are explored for, e.g., on-demand droplet formation (Zhu and Wang, 2017).

152

153 **3. Concepts to increase throughput**

154 The capacity to obtain monodisperse droplets of predefined sizes is a main benefit of passively
155 driven microfluidic emulsification. Most microfluidic devices can disperse fluids at rates that
156 require high-speed cameras for detailed examination of the droplet formation. Still, as drops
157 are formed one after the other, the overall throughput of a single channel does not compare
158 with the quantities obtained in batch emulsification processes applied, e.g., for pharmaceutical
159 sustained release carrier products (Takechi et al., 2002).

160 Changing one parameter directly linked to throughput in microfluidics, i.e., simply increasing
161 the flow rates, may not necessarily result in higher output of the desired product. The size of
162 droplets is determined by the device geometry in combination with process parameters (e.g.
163 flow rates, flow rate ratios) and fluid properties (e.g. viscosities, interfacial tension). Enhancing
164 the flow rate of one phase (i.e., altered flow rate ratio and/or capillary number) may be
165 associated with an alteration in droplet size or transition in droplet formation modes (Utada et
166 al., 2007) (Loizou et al., 2018). For instance, increasing capillary numbers of the continuous
167 phase and the flow rate ratios caused a substantial reduction of droplet size in polycarbonate
168 flow focusing devices as long as operated within the dripping regime. However, eventually
169 jetting occurred and much larger, less homogeneous particles were obtained (Lashkaripour et
170 al., 2019) (Fig. 2A). Increased back-pressure at high flow rates and, in consequence, bulging
171 of some polymer materials like PDMS used to make channels by soft lithography may set
172 additional limitations (Matthew et al., 2003). Furthermore, as it is true for most processes upon
173 increasing speed, slight variation of experimental parameters like flow rate fluctuations or
174 environmental vibrations can disturb a continuous operation (Li et al., 2014). Keeping this in
175 mind, in a real life situation, faster production may be at the risk of an enhanced or unstable
176 width of particle size distributions.



177

178 **Fig. 2:** Effect of increasing flow rates on droplet formation in a focused flow polycarbonate device. (A) Increasing
 179 flow rate of the continuous oil phase (increasing Ca) and flow rate ratio Φ of the water and oil phases
 180 result in altered particle sizes and regime changes from dripping to jetting conditions. (B) Identification
 181 of geometrical parameters affecting droplet generation rate to a high (red), medium (yellow and green)
 182 or limited (blue) extent at various flow conditions in the given device geometry: Orifice width (3); Aspect
 183 ratio of orifice width (3) and channel depth (4); Expansion ratio as the ratio of orifice width (3) and outlet
 184 channel width (6); Normalized widths of the water (1) and oil (2) inlets each determined in relation to the
 185 orifice width (3). Republished with permission of The Royal Society of Chemistry from (Lashkaripour et
 186 al., 2019); permission conveyed through Copyright Clearance Center, Inc.

187

188 Still, with most of the existing concepts to increase throughput as presented below, which focus
 189 either on enhancing the productivity per channel or on increasing the number of employed
 190 emulsification junctions (DFU, droplet-forming units), very homogeneous droplets can be
 191 obtained based on tailored design of droplet generator arrangements and well-balanced
 192 operation modes.

193

194

195 **3.1. Per-channel productivity – balancing throughput versus precision**

196 ***Altered junction geometries***

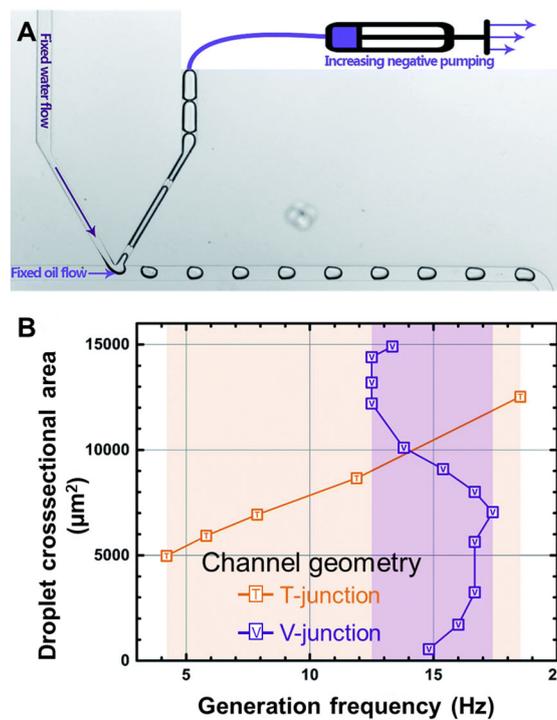
197 Modifying the geometry of the microfluidic junctions such as the channel shape, height, width
198 and cross-sectional area, respectively, are some of the main principles to affect the droplet
199 formation. As illustrated in Figure 2B, geometrical parameters can either strongly or less
200 extensively affect droplet generation rates, which allowed to cover a range of 10 – 800 formed
201 droplets per second (Hz) in this specific flow focusing device (Lashkaripour et al., 2019).

202 In addition to channels with symmetric cross sections, single channels with triple- or
203 multihelical cross-sectional designs were evaluated. In these channels, the primary axial flow
204 has been overlaid with a secondary circumferential flow allowing for symmetry breaking and
205 droplet formation from jets of immiscible fluids. The time of droplet formation decreased and
206 thus the frequency increased systematically when the angle of the helix was increased
207 (Ganneboyina and Ghatak, 2013).

208 Beside tailoring channel dimensions, alternative geometric arrangements may allow to
209 increase throughput. One of those concepts deviates from using individual DFU with defined
210 flow paths, but uses an array with several rows of pillars that break crude emulsion into fairly
211 homogeneous droplets depending on pillar geometry (coefficient of variation $CV \geq 13\%$). This
212 approach may be beneficial particularly for very viscous droplets, i.e., a ratio of viscosities of
213 the dispersed and continuous phase $\eta_d/\eta_c \gg 1$, which are often hard to emulsify with standard
214 DFU. The applicability of such a device containing 40 rows of square pillars was exemplarily
215 shown for methacrylate-based UV-crosslinkable monomers by preparing 20 μm polymer
216 microparticles at high production rates of $15 \text{ g}\cdot\text{h}^{-1}$ (CV 20%) (Amstad et al., 2014).

217 A precise control of droplet size and size distributions can be obtained by a modification of the
218 T-junction set-up, in which the straight main channel is connected with two side channels in a
219 V-shape arrangement (Fig. 3). The main channel is supplied with continuous phase and one
220 of the side channels located, e.g., in an angle of 60° to the main channel, provides the disperse
221 phase. The second side channel acts as control channel and is supported with a valve and/or

222 pump suitable to inject or withdraw fluids. Depending on the operation mode of the control
 223 channel, at a fixed rate of continuous phase flow, the size of droplets can be tuned by droplet
 224 splitting at the V-junction. This setup allowed tuning particle sizes formed without major effect
 225 on generation frequencies of ~ 15 Hz. Additionally, also relatively higher frequencies of droplets
 226 formation for small particles were possible at V-junctions compared to a T-junction of similar
 227 dimensions (Ding et al., 2015). It should be noted that unsymmetrical droplet splitting, as used
 228 here, may be linked to a substantial quantity of dispersed phase being discarded, which could
 229 be critical in economic production.



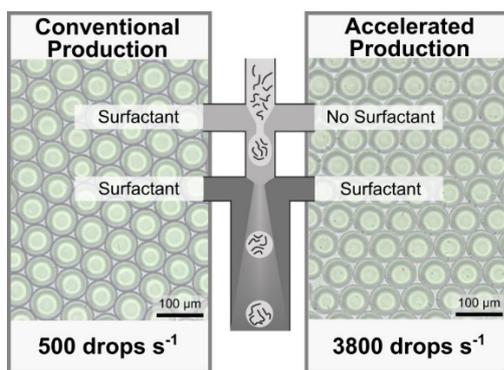
230
 231 **Fig. 3:** V-channel geometry allowing manipulation of droplet size. (A) Image of V-channel set-up: A pump is
 232 connected to the control channel (upper right) applies suction (“negative pumping”) for droplet splitting.
 233 By increasing suction rates, the size of droplets that are collected through the main channel are
 234 systematically reduced. (B) Dependency of droplet size and generation frequency for V-channel
 235 geometry compared to T-junction channels. Republished with permission of The Royal Society of
 236 Chemistry from (Ding et al., 2015); permission conveyed through Copyright Clearance Center, Inc.

237
 238 **Delayed surfactant addition**

239 As indicated above, enhancing the flow rates to increase throughput will face limitations as a
 240 transition occurs from the well-controlled dripping regime towards the jetting regime with
 241 typically broader size distributions. The dripping-to-jetting transition was assigned to a state,

242 when either Ca of the outer fluid or We of the disperse phase or their sum is roughly in the
 243 range of 1 on a logarithmic scale (Utada et al., 2007). As has been shown in Equations 1 and
 244 2, these numbers present the balance of viscous forces of the continuous phase (Ca) and the
 245 inertial force of the disperse phase (We), each normalized to the interfacial tension σ between
 246 the two phases. Based on this inverse relationship of Ca and We to σ , a high value of σ might
 247 in principle allow increasing the throughput (v) while keeping Ca and We low, i.e. remaining in
 248 the dripping regime. However, for kinetic stabilization of emulsion droplets and to avoid
 249 coalescence leading to broad droplet size distributions, typically a coverage of the interface of
 250 the immiscible phases with surfactants or steric stabilizers is needed, leading to reduced σ . By
 251 delayed surfactant addition in PDMS devices, a temporal and spatial separation of droplet
 252 formation at the nozzle and droplet stabilization by surfactants added further downstream can
 253 be obtained (Fig. 4). In this way, the per-channel rate of microgel precursor droplets could be
 254 increased roughly 8-fold, while preserving operation in the dripping regime (Seiffert et al.,
 255 2015). The concept has also been applicable to altered viscosity ratios of dispersed and
 256 continuous phases (Friess et al., 2016), suggesting a potential broad applicability to various
 257 types of materials. Via use of a shielding fluid, the concept can also be realized in focused flow
 258 glass capillary setups with two inner capillaries (Josephides and Sajjadi, 2015).

259



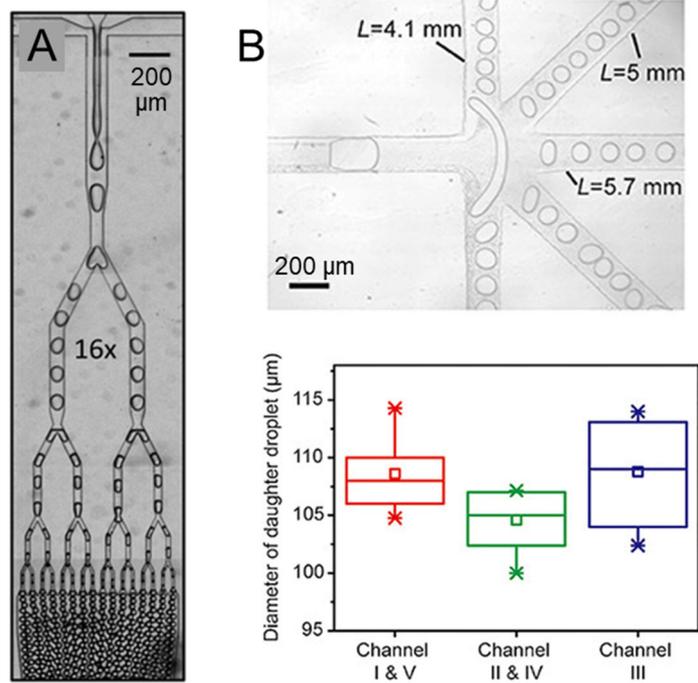
260

261 **Fig. 4:** Delayed surfactant addition to enhance per-channel productivity of a water-in-oil (w/o) emulsion as
 262 applied to prepare precursor particles subsequently gelled to poly(*N*-isopropylacrylamide) microgels.
 263 Left: Scheme of conventional production with surfactant added directly at the DFU. Right: Scheme of
 264 delayed surfactant addition, where surfactant containing continuous phase is added further downstream,
 265 thereby spatially separating droplet formation at enhanced rates and droplet stabilization. Reprinted from
 266 (Seiffert et al., 2015) with permission from Elsevier.

267

268 ***Droplet multiplication by droplet splitting***

269 In a typical microfluidic setup, devices with larger cross-sectional areas of the emulsification
270 nozzle allow for operation at relatively higher flow rates while remaining in the dripping regime.
271 Under such conditions, the volume of the dispersed phase processed into particles is larger in
272 a given time frame compared to small channels, but at the same time also the formed particles
273 are larger. However, as mentioned above for V-shaped devices, preformed mother drops can
274 be split into two or more daughter droplets, thus allowing for size reduction. Importantly, high
275 precision of the splitting channel dimensions (cross-sectional area, length, angle) is required
276 in order to provide a homogeneous splitting into daughter droplets of identical volumes. A
277 number of split arrays purely based on passive splitting without additional energy sources have
278 been evaluated, including series of Y or T-shaped geometries. The concept has been applied
279 for single and double emulsions in planar arrays, where three or four subsequent binary
280 splitting junctions resulted in the target droplet diameter of, e.g., 35 μm (Fig. 5A). Compared
281 to production of the same particle size in a conventional channel, productivity by droplet
282 splitting was 5 to 10 times higher, while acceptable CV in the range of 5-6% were obtained
283 (Abate and Weitz, 2011).



284

285 **Fig. 5:** Droplet splitting to increase numbers of produced particles. (A) Multiple binary splitting technique with
286 each mother droplet split into $2^4 = 16$ daughter droplets. Republished with permission of The Royal
287 Society of Chemistry from (Abate and Weitz, 2011); permission conveyed through Copyright Clearance
288 Center, Inc. (B) Simultaneous mother droplet splitting in 5 daughter droplets, where hydraulic resistance
289 of the respective outlet channels is balanced by outlet channel length to obtain similar daughter droplet
290 sizes. Reprinted from (Li et al., 2019b), with the permission of AIP Publishing.

291

292 Single-step splitting not necessarily has to result in two daughter droplets only. In a
293 multifurcating device with five outlet channels arranged in a planar semi-circle around the inlet
294 channel, five similarly sized daughter droplets could be prepared from each mother droplet in
295 a single step (Fig. 5B). Relatively large mother droplets of up to $350\ \mu\text{m}$ in diameter had to be
296 deformed extensively and temporarily plug all outlet channels to allow for a symmetrical break-
297 up (Li et al., 2019b).

298 Flow obstacles, such as sharp tips or edges, provide further opportunities to divide droplets.
299 For instance, a feeding glass capillary can be coaxially connected to a second downstream
300 capillary containing, e.g., three star-shaped axisymmetric inner walls acting as knives. Here,
301 relatively large single or double emulsion drops could each be split into three daughter droplets,
302 with the three fractions having a narrow size distribution and almost identical mean diameters
303 (Chen et al., 2016). In all those cases of droplet splitting, perfectly symmetrical arrangements
304 and well controlled hydraulic resistance of the (several) outflow paths are an essential
305 precondition to obtain comparable daughter particles. This may be some of the reasons why
306 also active splitting principles based, e.g., on pneumatic valves, non-uniform temperature fields
307 or acoustic waves are explored to tune daughter droplet sizes.

308

309 ***Controlled jetting and tip-streaming operation***

310 Increasing the flow rates of the continuous and dispersed phase in a given channel geometry
311 typically results in the formation of a thread of dispersed phase that expands into the collection
312 channel. In this so-called jetting regime, the liquid jet experiences axisymmetric surface
313 perturbations eventually leading to droplet break-up. The wavelength of these oscillations

314 correlates with the flow rate of the dispersed phase and is a parameter to predict droplet sizes
315 formed in the jetting regime (Cordero et al., 2011) (Zhu and Wang, 2017).

316 In narrowing jets, where Ca of the continuous phase is large, viscous drag results in stretching
317 of the liquid thread of the dispersed phase. In this operation mode, smaller particles can be
318 obtained compared to the dripping regime. The concept is applicable also to double emulsions,
319 where the transition from dripping to narrowing jets has been numerically described based on
320 the Ca of the different phases and the device geometry (Nabavi et al., 2017a). Attention may
321 need to be paid to small satellite droplets occasionally formed in narrowing jets in between
322 the main drops.

323 In tip-streaming, the dispersed phase liquid is forming a sharp cone rather than a long thread.
324 Out of the apex of this cone, a very fine jet expands and eventually breaks up into very small
325 droplets. In order to achieve this mode, a compression of the liquid thread from all sites was
326 found to be necessary, as realized in glass capillary microfluidics, specifically designed
327 polymer-based channels, or hybrid devices (Xu et al., 2017b). By this method, hydrogel
328 particles as small as ~180 nanometer (CV 2.8%) were obtained from very diluted solutions of
329 poly(ethylene glycol) diacrylate (PEG-DA), when the pumping pressure ratio of dispersed and
330 continuous phase was systematically decreased and devices with a low hydraulic resistance
331 of the outlet channel were used (Jeong et al., 2012). However, narrowing jets and tip streaming
332 typically operate stable only within a relatively narrow process window, and may be
333 occasionally interrupted resulting in contaminations with large particles. An optical imaging of
334 the cone interface in tip streaming in combination with an automated adjustment of dispersed
335 phase pumping has been explored to tackle these issues (Moyle et al., 2013).

336 Both, narrowing jets and tip streaming mode allow increased frequencies of droplet production
337 and higher throughput of the dispersed phase, while also enabling to prepare smaller sized
338 particles that are not routinely assessable in the dripping regime. Broader size distributions
339 than in the dripping regime may occur, e.g., CVs of 10-25% were reported for PEG-DA
340 templated by narrowing jets and tip streaming (Wu et al., 2018). However, others reported

341 excellent polydispersity indices (PDI) < 0.04 for 200 nm droplets of a perfluorocarbon liquid
342 prepared as contrast agent via tip streaming (Xu et al., 2017b). Thus, in optimized conditions
343 and with ideally suited fluid properties, a particle uniformity may be reached by narrowing jets
344 or tip streaming that is well acceptable particularly when compared to batch emulsification
345 processes in pharmaceutical production.

346

347 **3.2. Numbering-up – parallel operation based on complex device designs**

348 The typical industrial scale-up concepts such as increasing vessel sizes cannot be applied in
349 microfluidics, as channel dimensions sensitively affect droplet dimensions. Instead, scale-up
350 can be achieved by numbering-up, i.e., a parallel operation of several individual emulsification
351 junctions.

352 Since the supporting infrastructure such as pumps, pressure controller etc. are the most
353 spacious and costly parts of the necessary equipment, one would prefer a parallel operation
354 of several channels in a single device fed by one set of pumps over the use of multiple totally
355 independent set-ups. In flow rate controlled dispersion, this is, however, associated with the
356 challenge to ensure an identical hydraulic resistance at each dispersing site of the device in
357 order to allow for identical droplet sizes. Therefore, a complex design and a precise
358 construction of devices are needed.

359 One should be aware that an accidental partial or full blockage of single channels in a set of
360 parallel channels connected by, e.g., a tree-like feeding system (Li et al., 2008b) can
361 dramatically change droplet sizes obtained in the (partially) blocked and all other channels due
362 to a shift of local flow rates, thus broadening the size distribution. This may lead to out-of-
363 specification results for the entire production run. Additionally, it should be noted that pressure
364 oscillations associated with droplet formations e.g., at T-junctions, if not perfectly synchronized
365 at the different DFU, can effect particle size homogeneity in some parallelized devices (Barbier
366 et al., 2006). An efficient filtering system for the provided fluids and a continuous monitoring of

367 obtained droplet sizes is mandatory to avoid or, at least, rapidly detect particle size
368 inhomogeneities.

369

370 ***Parallelized glass capillary devices***

371 A practical, lab-scale approach for parallelized devices, which is also suitable for operation
372 with polymers dissolved in organic solvents, may be hand-built from glass capillaries arranged
373 together on a microscope slide and all connected to the same inlet and outlet ports (Kim et al.,
374 2013b). However, slight differences in orifices sizes, distances between the orifices in double
375 emulsion set-ups, and unequal flow rates depending on the respective channel position relative
376 to the device inlet are almost unavoidable for such systems and may contribute to a broadened
377 size distribution of the formed droplets.

378

379 ***Spatial arrangement of feeding channels and DFU in flow rate dependent particle*** 380 ***templating***

381 Using soft lithography, many identical DFU can be realized in a single device typically
382 replicated in PDMS. High precision device fabrication is needed as alterations of the nominal
383 channel width/depth in the range of $\pm 2\%$ can very strongly alter the flow rates, as shown by
384 simulation for tree-like feeding channels. This phenomenon was much less pronounced for
385 comb-like feeding channels, which connect with the DFU and the subsequent collection
386 channels to a ladder-like network (Tetradis-Meris et al., 2009). However, in a ladder-like
387 network, different channel lengths will always be present from the device inlet/outlet to each
388 individual DFU. Accordingly, based on increasing hydraulic resistance with increasing distance
389 from the device inlet, variation of local flow rates will increase with the number of parallel DFU
390 in ladder-like networks. Some design strategies to tackle this issue are:

391 (1) Ladder-like channel networks employing large feeding/collection channels with low
392 hydraulic resistance R_c per segment compared to the much higher hydraulic resistance R_u of
393 the DFU. Accordingly, mainly the DFU and not the support channels account for local hydraulic
394 resistance at different emulsification junctions. A ratio $R_c/R_u < 0.01 / 2 \cdot N$ was suggested for

395 preparing single emulsions in a DFU with a given R_u , in this case with fluid supply from a
396 second layer of distribution channels (Fig 6A). This means that R_c has to be decreased
397 proportionally (by larger width/height) when increasing the number N of parallel operated DFU
398 to realize a final difference of $R < 1\%$ between any DFU (Romanowsky et al., 2012). The
399 feeding channels can also be placed in the same plane as the DFU, a design that allows
400 coupling several units as desired (Femmer et al., 2015).

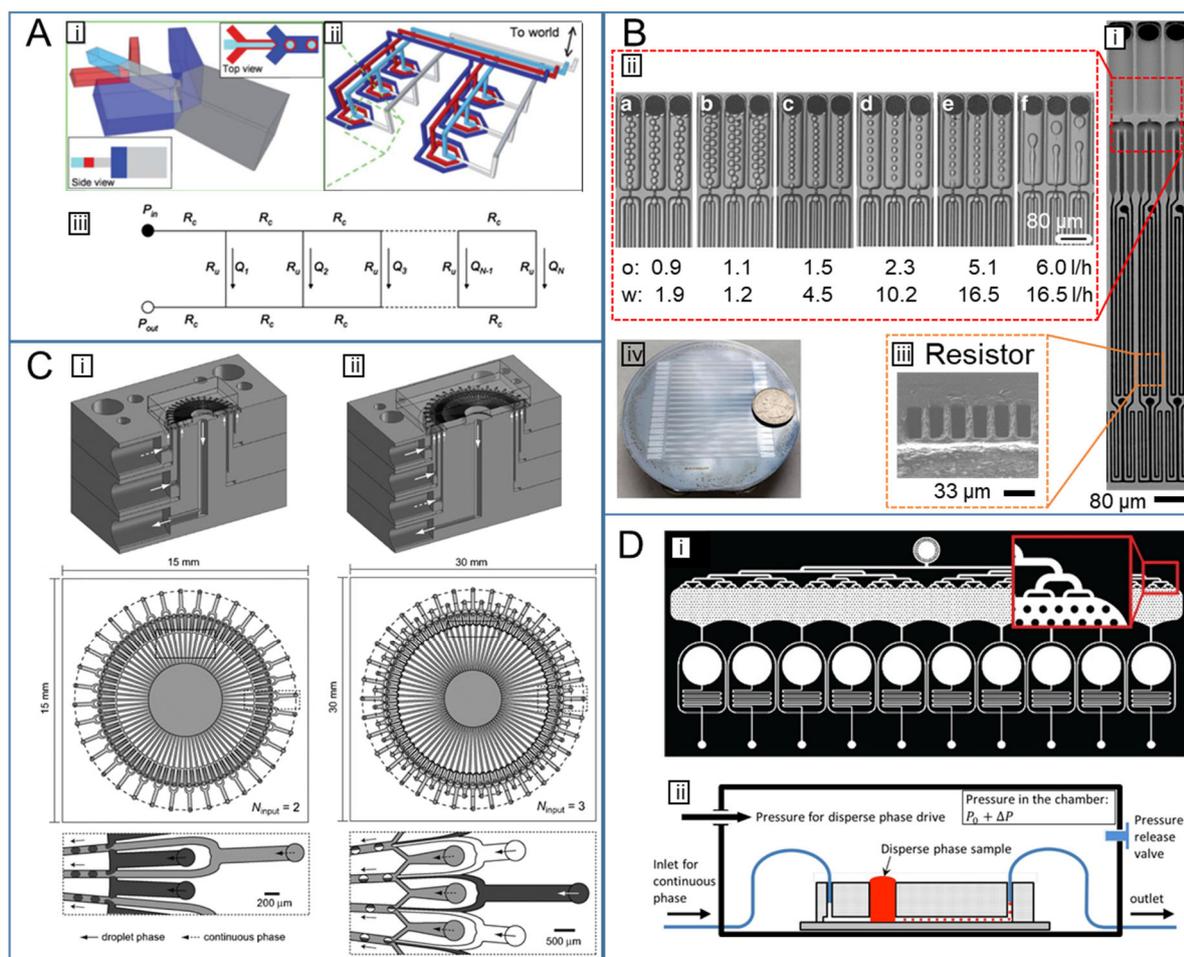
401 Ladder-like support structures have further been used to supply thousands of DFU created by
402 deep reactive-ion etching (DRIE) at a very high DFU packing on silicon wafers (20,160 DFU
403 on $6.3 \times 5 \text{ cm}^2$) (Fig 6B). Here, again, a two layer design has been adapted, in which the DFU
404 are connected to the supply channel with small connections (vias). The channels contain flow
405 resistors with a rectangular cross-section and small width (e.g. $8 \mu\text{m}$) upstream of the DFU. As
406 hydraulic resistance in those rectangular channels primarily scales with channel width and
407 throughput with channel height, both parameters can be (partially) decoupled. These devices
408 allowed to prepare $\sim 20 \mu\text{m}$ oil droplets ($\text{CV} < 5\%$; dispersed phase throughput 1-5 l/h) in the
409 dripping regime under pressure (Fig 6B), and were applicable to produce 5-9 μm poly(ϵ -
410 caprolactone) (PCL) microparticles thanks to device stability in organic solvents (Yadavali et
411 al., 2018) (Yadavali et al., 2019).

412 (2) Circular arrangement of the DFU on a microfluidic chip to have identical lengths of on-chip
413 feeding channels and a common, large outlet port (Fig. 6C). Such devices are supported by a
414 chip interface with concentric annular supply channels of minimized length and, again, show a
415 sufficiently low R_c to allow local differences in $R < 1\%$ for the respective N of DFU (Nisisako et
416 al., 2012). It should be noted that for the depicted single emulsion devices (Fig. 6C), the
417 emulsification principle corresponds to that of a T-junction, but involves feeding of the disperse
418 phase from two sides into the continuous phase operated in cross flow. The chips and holders
419 can be made from glass and steel, respectively, which also supports the handling of various
420 solvents not compatible with, e.g., PDMS based microfluidic chips. With such circularly
421 arranged devices containing, e.g., 144 parallel channels, the production of $\sim 90 \mu\text{m}$ acrylate-

422 based particles with a CV of 2.2% was demonstrated at a throughput of 180 ml/h (Nisisako et
423 al., 2012) (Nisisako and Torii, 2008).

424 (3) Shared on-chip feeding reservoirs with integrated pillars and funnel-like ports towards the
425 individual DFU (Fig 6D). Here, the pillars should realize a homogeneous flow distribution
426 upstream of all channels. The pillars may also allow for filtering and removal of particles that
427 may occasionally be present in the different phases as impurities, thus enhancing the
428 robustness of the production process. In principle, two of such feeding structures could be
429 arranged in a single plane, allowing for preparation of single emulsions. In a specific example,
430 a chip with one feeding structure for the continuous phase was combined with a pressure
431 chamber, in which the dispersed phase was delivered through drill-holes contacting the
432 channel plane. This set-up with 10 parallel channels allowed for the production of $\sim 50 \mu\text{m}$
433 droplets with a CV of 1.5% at a throughput of dispersed phase of 26 ml/h (Lim et al., 2015).
434 With this technology, on-demand drop formation with either identical or different dispersed
435 phases (e.g. for drug combinations encapsulated separately) at the different ports of the
436 parallel device might be achieved.

437



438

439 **Fig. 6:** Feeding channels and spatial arrangements for optimized parallel operation. (A) PDMS-device
 440 containing (i) double-emulsion DFU with (ii) ladder-like microchannel network for parallel operation. (iii)
 441 Model of hydraulic resistance R_c of supply/collection channel segments and R_u of the DFU; flow rate Q ,
 442 N number of channels; $P_{in/out}$ are feeding and collection ports of the PDMS device. Republished with
 443 permission of The Royal Society of Chemistry from (Romanowsky et al., 2012) ; permission conveyed
 444 through Copyright Clearance Center, Inc. (B) Silicon-based rectangular array with 36x560 individual DFU
 445 supported by a ladder-like support structure. (i) Images of 3 DFU, each being $80\ \mu\text{m} \times 1.6\ \text{mm}$. (ii)
 446 Magnification of nozzles when operated for hexadecane (o-phase) emulsification in aqueous Tween 80
 447 (w-phase) at the stated flow rates. (iii) Magnification of flow resistor channels upstream of each nozzle.
 448 (iv) Photograph of silicon chip. Adapted from (Yadavali et al., 2019) under a Creative Commons
 449 Attribution 4.0 International License (<https://creativecommons.org/licenses/by/4.0/>). (C) Glass/steel
 450 device with circular arrangement of DFU for parallel operation. (i) Single emulsion device. (ii) Device for
 451 3 phases, here with microchip to prepare Janus particles. Republished with permission of The Royal
 452 Society of Chemistry from (Nisisako et al., 2012); permission conveyed through Copyright Clearance
 453 Center, Inc. (D) Device with (i) modified tree-like feeding structure with shared buffer reservoir containing
 454 pillars for fluid distribution and (ii) pneumatic feeding principle for the dispersed phase. Reprinted from
 455 (Lim et al., 2015), with the permission of AIP Publishing.

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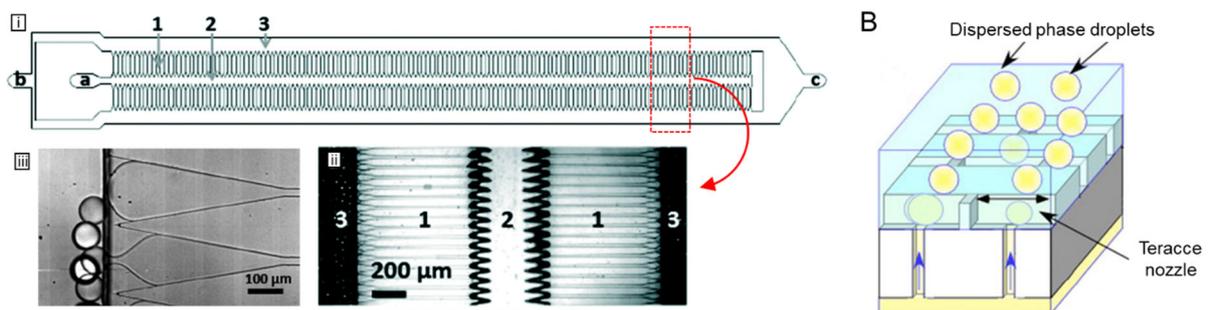
458

459 ***Parallelization of step emulsification***

460 In contrast to flow focusing geometries, where the balance of flow rates and other parameters
461 affect the droplet size in a sensitive manner, step emulsification is an alternative principle with
462 a number of benefits. No shear forces of the continuous phase are needed, thus reducing the
463 number of critical experimental parameters. However, the dispersed phase flow rate at a single
464 DFU needs to be slow in step emulsification in order to allow for faster nozzle clearance than
465 refilling. Still, minor flow rate alterations of the disperse phase have no strong impact on particle
466 size at least when remaining in the dripping regime at low Ca (Kobayashi et al., 2012) (Amstad
467 et al., 2016). Droplet diameters are dominated by device dimensions and, for a given fluid,
468 were discussed to directly scale with channel height h either by the factor $h^{1/3}$ (terrace nozzle)
469 or h^2 (triangle nozzle; see Fig. 1) due to the involved planes of drop confinement at the different
470 nozzle geometries (Nakajima, 2017) (Amstad and Weitz, 2017). Based on the relative
471 independence of droplet size and dispersed phase flow rate, undesired coupling of droplet
472 pinch-off between different DFU can be avoided. Via surface modification of the device
473 nozzles, the droplet sizes can be tuned since increasing contact angles between dispersed
474 phase and substrate (i.e. nozzle dewetting by dispersed phase in favor of wetting by
475 continuous phase) allows to produce smaller particles (Eggersdorfer et al., 2018). Combining
476 triangular nozzles with a gradual step (larger height of nozzle opening) resulted in larger
477 droplets at reduced frequencies (Eberhardt et al., 2019).

478 Considering those aspects, the power of step-emulsification is not necessarily found in its
479 throughput per DFU but the scalability by parallelization of DFU without major issues of
480 hydraulic resistance homogeneities for dispersed phase feeding. Accordingly, various types of
481 parallel arrangements of nozzles have been reported. For instance, a PDMS device with a total
482 of 550 triangular nozzles placed in two rows along a dead-end central feeding channel
483 (“millipede device”) produced water-in-oil droplets of the desired diameter in the range of 20-
484 150 μm , e.g., 150 μm droplets of PEG solution with a CV of 3% at a throughput of 150 ml/h.
485 The authors speculate that, based on the dense packing of the DFU, further upscaling of this
486 arrangement could lead to 600 liter per square meter and production hour ($\text{l/m}^2\text{h}$), while placing

487 the DFU like a membrane in between a feeding and a collection layer could lead to 11,000
 488 l/m²·h (Amstad et al., 2016) (Fig. 7A). Such a “membrane-like” vertical straight-through
 489 arrangement was realized with 24,772 terrace style nozzles in a 4x4 cm² silicon array (nozzle
 490 area 4 times 11x11 mm²) integrated in a steel housing, which produced 1.4 l/h of a higher
 491 viscosity oily phase into oil-in-water emulsions (87 μm droplets, CV 2%), equaling 2,800 l/m²·h
 492 (Kobayashi et al., 2012). The droplet diameter can be reduced to, e.g., 4-7 μm with finer
 493 nozzles, which needed to be numbered up to 176,176 DFU on a 11x11 mm² area to allow for
 494 an oil phase throughput of 1.5 ml/h in laboratory scale (Fig 7B) (Khalid et al., 2017). Terrace
 495 nozzles are also used in devices with dead-end supply channels and shared terraces for
 496 several DFU, demonstrating suitability for continuous preparation of ~5 μm particles from PCL
 497 and poly(*D,L*-lactide) (PLA) with stable sizes (CV ~2%) as investigated over several hours
 498 (Vladislavjevic et al., 2018).



499
 500 **Fig. 7:** Parallelization in step flow emulsification. (A) “Millipede” device (i) with two rows of DFU (ii) and triangular
 501 nozzles (iii). In-figure labels: a/b/c = inlet dispersed phase/inlet continuous phase/outlet; 1/2/3 = individual
 502 DFU/distribution channel for dispersed phase/collection channel with continuous phase. Republished
 503 with permission of The Royal Society of Chemistry from (Amstad et al., 2016) ; permission conveyed
 504 through Copyright Clearance Center, Inc. (B) Device with terrace shaped nozzles. Adapted with
 505 permission from (Khalid et al., 2017) © 2017 WILEY - VCH Verlag GmbH & Co. KGaA, Weinheim.

506
 507 A number of practical issues may need to be considered in parallelized step emulsification. As
 508 no flow of the continuous phase is needed to shear-off nascent droplets, high volume fraction
 509 emulsions can be achieved. Still, droplet transportation away from the DFU is required. This
 510 droplet clearance has been realized by a pressure-expandable vessel for droplet collection
 511 resulting in a net slope at the DFU (Dangla et al., 2013). More commonly, a flow of medium

512 through the collection area is applied. In fact, if the formed droplets are not removed from the
513 collection area at sufficient rates, they may approximate in the channel. This sets high
514 demands for efficient droplet stabilization such as by surfactants, may disturb the formation of
515 subsequent droplets by collision events (Mittal et al., 2014), or could even clog the channel in
516 case of rapidly solidifying formulations. In some studies, the flow rate of the continuous phase
517 was found to be optimal at 1.5x to 10x of the dispersed phase (Eberhardt et al., 2019). The
518 aforementioned vertical straight-through operation (Kobayashi et al., 2012), particularly for
519 dispersed phases with significant buoyancy, can help to prevent such effects as also shown in
520 a recent version of the millipede device (Stolovicki et al., 2018).

521 Device stability to solvents is another relevant point for the fabrication of particles from
522 hydrophobic polymers, as frequently used in controlled drug delivery. High solvent resistance,
523 at least for common fluids in pharmaceutical drug particle preparation, can be found in silicon
524 devices (Kobayashi et al., 2012), which may also be sterilized based on their heat stability.
525 Also, metal-based terrace devices made from stainless steel (Li et al., 2019a) or, probably with
526 lower stability, from aluminum (Zhang et al., 2019) are of potential relevance in this respect.
527 The millipede device, if fabricated from glass, can also be operated with organic solutions as
528 shown by their use for PCL microparticle preparation (Ofner et al., 2017).

529 The viscosity of the dispersed phases, which may be high in particle production from
530 concentrated polymer solutions, can heavily influence the possible flow rates and thus the
531 throughput of step emulsification devices in the dripping regime. For instance, in a direct
532 comparison of dichloromethane (0.56 mPa/s), *n*-decane (0.92 mPa/s) and paraffin oil (~50
533 mPa/s) in DFU with triangular nozzles, increasing flow rates in the 1-12.5 μ l/min per DFU
534 resulted in identically increasing droplet sizes for the different solvents at a given condition,
535 with the exception that paraffin oil could only be processed in a controlled manner up to
536 2 μ l/min (Eberhardt et al., 2019). When comparing *n*-tetradecane (2.7 mPa/s), medium chain
537 triglyceride (MCT, 20 mPa/s) and soybean oil (50 mPa/s) in terrace nozzles, the possible flow
538 rates for particle preparation dropped to 33% (MCT) and 6% (soybean oil) relative to *n*-
539 tetradecane, which was in accordance with predictions by computational fluid dynamics

540 (Vladisavljevic et al., 2011). At the same time, when the geometries and flow resistances of
541 parallelized step emulsification devices were optimized for oils, not all channels may be
542 productive during operation with low viscosity liquids when applying a slow flow rate regime
543 (Vladisavljevic et al., 2011). This indicates boundaries of the versatile use of a given step-
544 emulsification device.

545 A number of geometrical alterations of triangular nozzle devices were suggested in order to
546 produce droplets from high viscosity liquids. Considering the relevance of good wetting of the
547 nozzle by the continuous phase, concepts were derived in which the continuous phase is
548 added earlier in the step-emulsification channel to surround the dispersed phase liquid at the
549 step (Li et al., 2015) or in which bypass channels are used to feed the continuous phase into
550 the nozzle opening for better wetting of nozzle walls and centering of nascent droplets (Dutka
551 et al., 2016). Such backflow channels have also been successful in parallelized triangular
552 nozzles to process high viscosity liquids (155 mPa/s) (Hati et al., 2018).

553 In order to allow for more complex particle structures being prepared or to enable the
554 entrapment, e.g., of hydrophilic components in hydrophobic matrix materials, double emulsions
555 are a suitable technique. Given the need of preferential wetting of the nozzle by the respective
556 continuous phase at each emulsification step (w/o; [w/o]/w), integrating a cascade of oppositely
557 wettable micro-sized nozzles in highly parallelized one-piece step-emulsification devices is
558 challenging, which promotes the use of modular systems. Accordingly, direct coupling of two
559 step-emulsifiers either based on PDMS or on glass have been used to prepare w/o/w and
560 o/w/o emulsions, respectively, in some cases tailored to contain either one or multiple inner
561 phase droplets (Eggersdorfer et al., 2017) (Ofner et al., 2019).

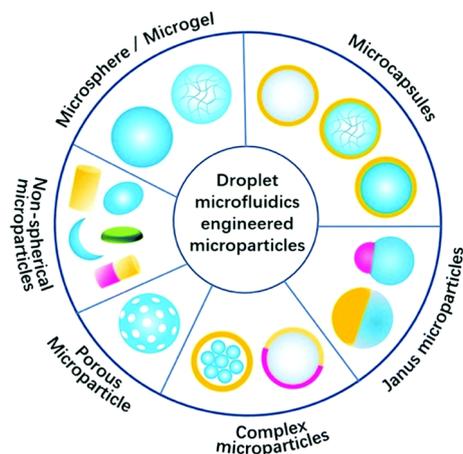
562

563 **4. Application of droplet-based microfluidics for preparing functional polymer carriers**

564 Microfluidics are extensively explored in life sciences for various applications, including the
565 microfluidic encapsulation of active compounds into microparticles, the embedding of living
566 cells in hydrogel carriers (Hasturk and Kaplan, 2019) (Alkayyali et al., 2019), or the microfluidic

567 preparation of nanoparticulate drug carriers through nanoprecipitation by non-solvents or
568 colloidal assembly (Liu et al., 2018) (Li and Jiang, 2018). When using gas as continuous phase,
569 amorphous drug nanoparticles can be obtained by microfluidic spray drying for otherwise
570 crystalline compounds (Amstad et al., 2015) (Steinacher et al., 2019), which is of interest to
571 accelerate the dissolution of hydrophobic drugs. Furthermore, when a jet of fluid is impacted
572 by droplets ejected from a laterally placed vibrating nozzle, core-shell or multicompartment
573 particles could be prepared in a gas-phase without the need of microchannels or immiscible
574 continuous phase solvents (Visser et al., 2018) (Kamperman et al., 2018). Given the
575 substantially growing number of microfluidic concepts to prepare functional polymer-based
576 carrier particles, here, a focus will be placed on selected examples of droplet-based particle
577 preparation approaches that could contribute to either improve existing or to realize new
578 features of drug carrier systems. For cell encapsulation, nanoprecipitation, and in-air
579 microfluidics, readers may be referred to the literature cited above.

580 When polymer solutions are processed in droplet-based microfluidics, the submicron size-
581 range can be reached by using highly diluted polymer solutions as dispersed phases. This
582 principle is based on massive droplet shrinkage upon solvent extraction, which is quite
583 inefficient, e.g., in terms of mass throughput. Therefore, the focus of the following sections will
584 be on microparticles, where the ability of microfluidics of precisely constructing single or
585 multiple phase droplets can be fully utilized. Accordingly, a broad design space of microparticle
586 shapes and ultrastructures may become accessible (Fig. 8) (Li et al., 2018), which should
587 allow tailoring physicochemical properties of such carriers and the drug release function.



588
 589 **Fig. 8:** Scheme of exemplary microparticle shapes and ultrastructures accessible by droplet-based
 590 microfluidics. Republished with permission of The Royal Society of Chemistry from (Li et al., 2018) ;
 591 permission conveyed through Copyright Clearance Center, Inc.
 592

593 From an industrial perspective, the beauty of microfluidic fabrication is not sufficient to justify
 594 the use in pharmaceutical production due to required investments and risks associated with
 595 any new technology. Instead, applications requiring distinct particle properties that could not
 596 be realized with other more established technologies are the most likely paths for market
 597 entrance of droplet-based microfluidics in pharmaceutical carrier production (Holtze, 2013).

598 Based on the technological principles of droplet based-microfluidics, beneficial features for
 599 preparing drug release systems might be:

600 *Exclusion of burst release through efficient drug embedment.* Burst release due to drug
 601 molecules located at or near the particle surface is a known characteristic of several approved
 602 sustained release formulations (Lundström et al., 2009) (Tunn and Wiedey, 2009), but only
 603 acceptable if the drug has a broad therapeutic window. Overcoming undesired burst release
 604 may allow extending the dosing intervals with a given amount of encapsulated drug and
 605 broaden the spectrum of drugs applicable in microparticulate depot formulations. The burst
 606 release from PLGA particles was studied with bupivacaine as a model drug used in its free
 607 base form, a compound with good water solubility at neutral pH as well as in acidic
 608 microenvironments (Shah and Maniar, 1993) as present in degrading PLGA particles (Ding
 609 and Schwendeman, 2008). Using flow focusing microfluidics for particle preparation, a massive

610 reduction in burst release was observed compared to the batch emulsification technique due
611 to better distribution in 11 μm particles with a drug loading as high as 20 wt.% (Xu et al., 2009).

612 *Co-delivery of compounds.* The encapsulation of compounds into different compartments of
613 the same particle may be preferred over mixtures of particles loaded with either of the
614 compounds when two or more substances should be co-delivered. Examples include cases
615 where the substances are chemically incompatible, have different solubility/stability profiles or
616 time points of anticipated release, or should produce active substances or imaging probes in
617 situ, e.g., by cleavage reactions of prodrugs (Liang et al., 2015). For instance, in vaccine
618 delivery by particulate carriers, a co-delivery of adjuvant and antigen to the same cell is needed
619 (Huang et al., 2019). Furthermore, shielding encapsulated peptide from intestinal enzymes
620 may be mediated by enzyme inhibitors entrapped in an outer matrix material (Araujo et al.,
621 2015). Janus particles and other substructured particles can host substances in several
622 individual compartments as templated by microfluidic techniques (Khan et al., 2015b), which
623 can be supportive to such co-delivery concepts.

624 *Multi-step local release.* In pharmaceutical product development, simple therapeutic
625 approaches such as multiple injections are typically preferred over complex delivery systems
626 like microparticles. However, for certain sites of the body that cannot be easily reached or only
627 through interventions at risk of tissue damage, multistep release formulations may be of
628 interest, e.g., for intraoperative placement. A conventional approach would be mixtures of
629 particles with different release pattern, e.g., due to different particle sizes (Berkland et al.,
630 2001). In this regard, microfluidics could provide technical solutions by parallel channels of
631 different geometries simultaneously producing differently sized particles into one batch, thus
632 avoiding later mixing issues expected especially for small particles in dry state. Furthermore,
633 material chemistry offers a broad toolbox of stimuli-responsive materials proposed for on-
634 demand local release (Fu et al., 2018) (Bruneau et al., 2019) (Kauscher et al., 2019). While
635 many of these materials are primarily of academic interest with many questions still to be
636 addressed (safety including toxicity and long term in vivo responses, physiological relevance
637 of applied stimuli, polymer degradation/elimination, production schemes, etc.), which affects

638 the realistic chances to be translated into pharmaceuticals in the next 10 to 20 years, it is still
639 relevant to evaluate the underlying principles and test their transferability to established
640 pharmaceutical additives.

641 *Sensitive compounds.* Turbulent mixing and exposure to hydrophobic interfaces can be
642 detrimental at least to some biopharmaceuticals. These issues motivate the use of low-shear
643 techniques and specific environments for proteins such as present in hybrid particles from
644 hydrogels with hydrophobic membrane layers (Kong et al., 2013). Thin intermediate hydrophilic
645 layers in between a miscible dispersed phase and the continuous hydrophobic phase, as can
646 also be used to prevent premature solidification of dispersed phases by gelling agents added
647 in the continuous phase, could serve as a protective layer during microfluidic templating for
648 sensitive molecules in the dispersed phase.

649 *Patient-specific formulations for precision medicine.* The concept of dose adaptation to patient
650 needs in precision medicine can only be successful when drug formulations are available in a
651 sufficiently fine grid of doses or such doses can be rapidly produced. While feasible for
652 injections or infusions of dissolved drugs, patient-specific particulate formulations are not yet
653 established in the clinics. Microfluidics, with its given throughput, may allow preparing such
654 formulations, e.g., by microfluidic devices prefilled with a number of drugs to be chosen from
655 as needed. Existing microfluidic technologies (see e.g. Fig. 6D) may be adapted for such
656 purpose. Cancer vaccines incorporating patient materials may be another example for
657 applications with need of individualized small-batch production.

658 *Plant size reduction, workers protection and incubator technologies.* Despite investments are
659 required for setting up and validating microfluidic processes in industrial production, financial
660 aspects may be in favor of microfluidics in selected cases. For instance, when plants for small
661 batch sizes of highly toxic substances use dedicated rooms/equipment and require a relatively
662 high number of personnel for operation and process supervision due to short shifts in all-body
663 protection suits, downscaling of equipment to fit into incubators as possible for microfluidic
664 setups may substantially reduce overall costs.

665

666 **4.1 Preparation of particles from preformed polymers**

667 The preparation of drug loaded particles through emulsion techniques is well established for
668 hydrophobic and hydrophilic bioactive compounds by batch processes (Wischke and
669 Schwendeman, 2008). Most commonly, hydrophobic polymers such as PLGA serve as matrix
670 materials, which are dissolved in an organic solvent (o-phase) like dichloromethane or ethyl
671 acetate having a limited solubility in an aqueous phase (w-phase). Hydrophobic drugs can be
672 co-dissolved or dispersed in the o-phase, which is then emulsified in the w-phase forming an
673 o/w emulsion. The slow extraction of the o-phase solvent into the w-phase, from where it may
674 evaporate, results in shrinkage of nascent particles and eventually their solidification. For a
675 hydrophilic payload, an additional w_1 phase is very commonly used to form $w_1/o/w_2$ double
676 emulsions. The emulsification and solvent extraction/evaporation process has been realized
677 for a number of (model) compounds in microfluidic channels. Based on the increasing
678 availability of ready to use microfluidic equipment packages and available technical application
679 notes, more studies are expected to be published for various drugs encapsulated into
680 preformed polymers by microfluidic techniques in the next couple of years. Computational
681 tools, including fluid dynamics simulation, have also been successful to predict droplet and
682 polymer particle sizes under variation of flow conditions (Vladisavljevic et al., 2014), thus
683 supporting efficient research and development.

684 As some channel materials such as PDMS tend to swell in organic solvents resulting in channel
685 diameter alteration with time, o-phase solvents have been exchanged, e.g., to dimethyl
686 carbonate (Hung et al., 2010). Alternatively, solvent resistant glass, steel, or silicon devices
687 have been used (Herranz-Blanco et al., 2017) (Ofner et al., 2017) (Yadavali et al., 2018). Given
688 the permeability of PDMS to dimethyl carbonate and the high surface to volume ratio of
689 microfluidic channels, on-chip solvent extraction/evaporation has been reported (Hung et al.,
690 2010). For devices with non-permeable channel walls, particles can be collected in an open
691 vessel for solvent evaporation or commercially available membrane modules may be
692 integrated in the devices for solvent exchange in cross-flow.

693 Bulk emulsification techniques for drug encapsulation in hydrophobic polymers are often linked
694 to a number of critical issues, e.g., a potential drug instability at interfaces during shear-based
695 emulsification, drug loss to the continuous water phase, improper encapsulation as
696 characterized by drug molecules being deposited in larger aggregates at or near the particle
697 surface, drug instability during storage, and release pattern with high initial rates (burst) and/or
698 incomplete release. Microfluidic techniques could address at least some of these points so far.
699 As noted above, the massive burst release of bupivacaine from conventionally prepared
700 ~11 μm or ~44 μm PLGA particles could be overcome by microfluidic preparation with a more
701 homogeneous drug entrapment in the particle core. In this context, another relevant feature
702 may be the much more homogeneous particle size (Fig. 9A) (Xu et al., 2009), thus avoiding a
703 fine particle fraction with higher surface to volume ratio contributing to faster release. Similar
704 observations were made for encapsulation of Cyclosporin A in 5-10 μm PLGA particles
705 comparing o/w microfluidic emulsification to spray drying and bulk emulsion/solvent
706 evaporation, where encapsulation efficiencies were much higher and release was more
707 prolonged for particles from microfluidics (Keohane et al., 2014).

708 While water can generally penetrate into compact PLGA particles at a given rate, the exchange
709 of water and drug under release conditions can be accelerated by introducing pores, e.g., by
710 adding salt to the w_1 phase serving as osmotic agent during particle formation (Wischke et al.,
711 2010). Accordingly, ammonium carbonate as porogen allowed for a faster but still controlled
712 release of a dye, coumarin-6, from ~90 μm PLGA particles from microfluidic production
713 (Amoyav and Benny, 2019). Exchanging PLGA with PEG-PLGA resulted in massively
714 accelerated burst-like release of Cyclosporine A due to high microparticle porosity and rapid
715 water uptake under release conditions as promoted by domains from hydrophilic PEG chains
716 (Keohane et al., 2014). Furthermore, (open) pores can be introduced by using w/o/w rather
717 than o/w emulsions, which may have contributed to the observed ~30% burst release of well
718 water-soluble lidocaine hydrochloride from porous PCL particles (Vladisavljevic et al., 2014).

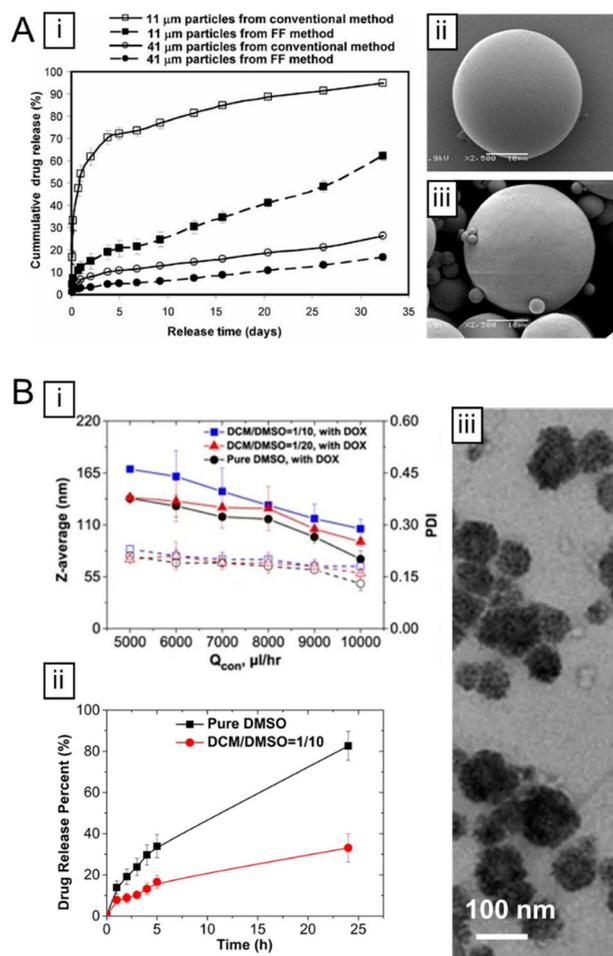
719 Encapsulation efficiency is a very relevant feature in particle production and is linked to both
720 the formulation approach and the characteristics of the employed substances. A number of

721 drugs were investigated for microfluidic encapsulation in polyesters, including, e.g., paclitaxel
722 (He et al., 2011), enoxacin (Kim et al., 2013a), risperidone (Jafarifar et al., 2017), latanoprost
723 and dexamethasone (Leon et al., 2015). Other than in large stirred vessels, only a relatively
724 small volume of continuous phase is in contact with the nascent particles within the
725 confinement of a microfluidic channel. Therefore, only this fluid volume may account for
726 exchange processes with the o-phase, namely solvent extraction for particle hardening and
727 drug loss by diffusion and/or convective forces. For size ratios of droplet-to-channel width of
728 <0.8 , the mass transfer of dimethyl carbonate into water in an o/w two-phase system was
729 dominated by diffusion (Vasiliauskas et al., 2015). Generally, hydrophobic drugs often have
730 high encapsulation efficiencies in bulk o/w emulsions, which should also transfer to
731 microfluidics. As expected, for very poorly water-soluble anticancer and antiinflammatory drugs
732 like sorafenib and celecoxib, a quantitative encapsulation in microparticles from acetalated
733 dextran was observed at least when the polymer was used above a certain threshold
734 concentration. It was concluded that the polymer accumulates at the interface and serves as
735 diffusion barrier during droplet shrinkage and particle solidification (Vasiliauskas et al., 2015).
736 While more studies on the balance of solidification speed and drug leakage in microfluidic
737 channels are still to be conducted, including model studies in zero-flow conditions for
738 determining diffusion coefficients (Kinoshita et al., 2016), it is obvious that the
739 solubility/miscibility of the o-phase solvent in the water phase can mediate drug loss due to
740 instable droplet interfaces and solubility enhancement.

741 Dimethyl sulfoxide/dimethyl carbonate as a mixed o-phase for PLGA were emulsified in the
742 jetting regime into small droplets, from which the solvents were subsequently extracted leading
743 to the formation of numerous nanoparticles through a proposed diffusion-limited cluster
744 aggregation mechanism (Xu et al., 2017a). At achieved drug payloads of >15 wt.%, the
745 encapsulation efficiencies by this technique were up to 88% for hydrophobic tamoxifen and
746 80% for hydrophilic doxorubicin, which were 20% higher than for doxorubicin loaded particles
747 formed by nanoprecipitation. Furthermore, better entrapment of doxorubicin within the polymer

748 matrix material could be concluded for particles from those solvent mixtures based on their
749 substantially slower release (Fig. 9B) (Xu et al., 2017a).

750 The encapsulation of hydrophilic substances such as proteins is typically conducted by w/o/w
751 emulsion, where – by concept – microfluidics provide much higher precision of entrapment
752 than bulk double emulsification. In particular, bulk double emulsions may regularly be
753 associated with fusion of w_1 and w_2 phase droplets during high shear emulsification of the w_1/o
754 primary emulsion in the continuous w_2 phase. By microfluidic double emulsion, successful
755 protein encapsulation has been demonstrated for PCL (Pessi et al., 2014) or PLA particles
756 (Bokharaei et al., 2017) at least with bovine serum albumin as a relatively stable model protein.
757 Encapsulation efficiencies of 84% and a distinct burst release were observed (Pessi et al.,
758 2014). However, there might be a lack of subsequent protein release from polyester matrices
759 as can be expected by restricted protein diffusion in hydrophobic polymers and potentially
760 protein unfolding phenomena. Employing very thin hydrophobic polymer shells as membranes
761 around a liquid core may reduce the diffusion length through the polymer phase and allow
762 controlling the release of hydrophilic molecules. Such thin-shell (70 – 150 nm) semipermeable
763 capsules from high molecular weight PLGA templated by w/o/w emulsions released hydrophilic
764 dyes in vitro and in vivo over several weeks. In addition to dye diffusion and polymer
765 degradation, the mechanical failure of the thin shell under mechanical stress during injection
766 and in the tissue may have had a major contribution to release (Lee et al., 2017).



767

768 **Fig. 9:** Release characteristics of drug loaded PLGA particles. (A) Bupivacaine loaded particles. (i) Release
 769 pattern with reduction of burst release by microfluidic preparation (FF: microfluidic flow focusing). (ii, iii)
 770 SEM images indicating particle surface structure and size distribution (mean ~11 μm) for particles from
 771 microfluidic preparation (ii) compared to conventional bulk emulsification technique (iii). Reproduced with
 772 permission from (Xu et al., 2009) Copyright © 2009 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim.
 773 (B) Doxorubicin loaded nanoparticles prepared by mixed o-phase solvents. (i) particle size distribution
 774 depending on flow rate of continuous water phase (flow rate of dispersed phase 50 μl/min), (ii)
 775 doxorubicin release depending on solvents used for preparation, (iii) TEM image of PLGA particles with
 776 black spots assigned as doxorubicin aggregates. Adapted from (Xu et al., 2017a) under a Creative
 777 Commons Attribution 4.0 International License (<https://creativecommons.org/licenses/by/4.0/>).
 778

779 Beside hydrophobic polymers, also preformed hydrophilic polymers can be dispersed into
 780 droplets by microfluidics. In order to stabilize the dispersed phase droplets, chemical
 781 crosslinking is often used as main stabilization principle, as addressed in section 4.2. However,
 782 also physical interactions such as complexation or temperature-induced gelation are common
 783 principles in such hydrogel particles, also termed as microgels. Furthermore, the clustering of
 784 hydrophobic chain segments of amphiphilic polymers into bridged micelles can be used to

785 build physical netpoints in microgels (Wang et al., 2013). A typical material to prepare
786 microgels via physical interaction is alginate. It gels by complexation of divalent cations like
787 Ca^{2+} , which are added on the chip either directly as a calcium chloride solution during droplet
788 templating or by ligand exchange through an additional aqueous phase (Mazutis et al., 2015)
789 (Hati et al., 2016). Alternatively, calcium ions are often provided to previously templated
790 droplets in the collection bath (Chuah et al., 2009) (Hou et al., 2019). Model proteins or
791 antibodies codissolved in the alginate solution have been encapsulated in the hydrogel phase
792 by this procedure (Hati et al., 2016), or separate protein and alginate phases allowed liquid
793 core enclosure by the alginate hydrogel (Yu et al., 2019). In a typical setting, laborious removal
794 of the continuous oil phase is required prior to further use.

795 Also hybrid particles from preformed polymers have been prepared with hydrophobic cores
796 and hydrophilic shells (Wu et al., 2013) or vice versa (Kong et al., 2013), where the different
797 compartments can act as depots, release rate controlling units, or both.

798

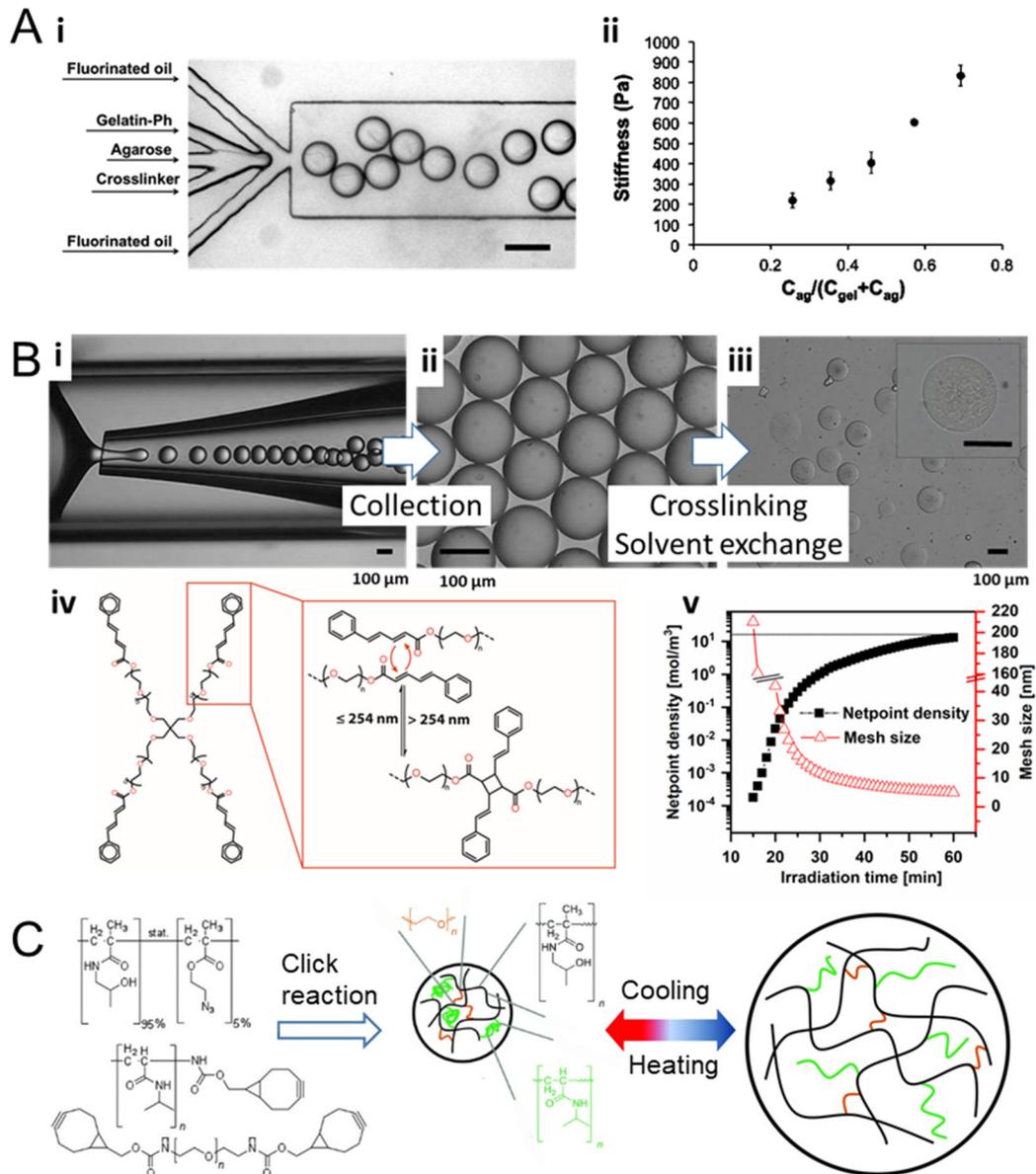
799 **4.2 Drops as microreactors for polymer (network) synthesis**

800 Droplets are transported in microchannels in a synchronized manner in one plane. Based on
801 this feature, they are accessible to various types of subsequent modifications, e.g.,
802 deformation, splitting, fusion, liquid injection, mixing, or illumination. Using the tool box of
803 materials chemistry, droplets can serve as microreactors for transforming
804 monomers/oligomers into polymers, particularly polymer networks, in order to adapt and
805 maintain a desired particulate structure and shape. In this way, particles with various material
806 properties can be made available, e.g., by systematically changing the flow rates of
807 comonomer phases and thus the copolymer composition during screening experiments.
808 Furthermore, as bulk polymer network materials cannot be easily processed into micrometer-
809 sized particles with defined shapes at a relevant throughput, in situ crosslinking of droplets for
810 network formation is the way of choice for such materials. For pharmaceutical applications, the
811 level of (chemical) homogeneity of the material, the structure of netpoints, the degradability of
812 the polymer network, the nature of degradation products, and the potential interference of

813 crosslinking reactions with simultaneously present drug molecules are, however, subjects
814 requiring particular attention when following this path.

815 Microgels, i.e. microparticulate hydrogel networks, are commonly obtained by in situ polymer
816 network synthesis, most often via photoinduced radical polymerization of (meth)acrylate
817 moieties (Plamper and Richtering, 2017). Such concepts can also be transferred to
818 microfluidics. The heat generated during photoinitiated free radical polymerization of
819 tripropylene glycol diacrylate precursors in the presence of a catalyst could also be used to
820 induce a second catalyzed polymerization, eventually resulting in an interpenetrating network
821 of polyacrylate and polyurethane/urea (Li et al., 2008a). Further crosslinking concepts explored
822 for microfluidic microgel fabrication include Schiff bases formed between aldehyde and amine
823 moieties of polysaccharides (Zhang et al., 2011), crosslinking of polyethylenimine with
824 diisocyanates by an addition reaction (Polenz et al., 2015), or horseradish peroxidase
825 catalyzed gelation of tyramine functionalized gelatin used in mixture with agarose as a
826 thermogelling material (Chau et al., 2014). In the latter case, the enzyme induces gelation by
827 conversion of tyramine to phenoxy radicals that are known to couple through radical reactions
828 (Sakai and Nakahata, 2017), eventually resulting in complex structured agarose-gelatin
829 networks with controllable mechanical stiffness (Fig. 10A) (Chau et al., 2014). If radical
830 polymerization should be avoided, photoinduced 2+2 cycloaddition reactions of cinnamylidene
831 acetic acid (CAA) moieties may be employed to create netpoints. In this study, an o/w emulsion
832 of PEG-CAA dissolved in CHCl_3 was exposed to UV irradiation at 308 nm, followed by solvent
833 exchange with water to obtain stable microgels (Fig. 10B). By irradiation conducted for
834 increasing time periods without the presence of photoinitiators, the network density and thus
835 the mesh size could be controlled (Tuncaboylu et al., 2017), which might allow controlling drug
836 release rates. When chain segments, which are subject to a reversible hydration at a lower
837 critical solution temperature like poly(*N*-isopropylacrylamide), are incorporated in microgels
838 constructed e.g. by azide-alkyne click reaction, size expandable particles can be obtained
839 (Fig 10C) (Hackelbusch et al., 2015).

840



841

842 **Fig.10:** Microgel synthesis by polymerization/crosslinking reactions in template droplets. (A) Enzymatic
 843 crosslinking by horseradish peroxidase. (i) Templating w/o droplets from agarose, tyramine
 844 functionalized gelatin, and HRP plus hydrogel peroxide. (ii) Effect of agarose to gelatin ratio on microgel
 845 stiffness. Adapted with permission from (Chau et al., 2014). Copyright (2014) American Chemical
 846 Society. (B) Photoinduced 2+2 cycloaddition reaction. (i) o/w droplet templating; star shaped PEG
 847 bearing CAA moieties dissolved in CHCl_3 serves as dispersed phase, (ii) particle collection in aqueous
 848 poly(vinyl alcohol) in a sealed chamber, (iii) microgel particles after crosslinking and evaporation of CHCl_3
 849 resulting in solvent exchange with water, (iv) principle of 2+2 cycloaddition reaction, (v) Effect of
 850 irradiation time at 308 nm on polymer network properties. Reprinted from (Tuncaboylu et al., 2017) with
 851 permission from Elsevier. (C) Preparation of thermosensitive microgels containing temperature-
 852 responsive poly(*N*-isopropylacrylamide) segments by strain promoted azide-alkyne cycloaddition
 853 reaction, an example of click chemistry. Adapted with permission from (Hackelbusch et al., 2015); ©
 854 2015 WILEY - VCH Verlag GmbH & Co. KGaA, Weinheim.

855

856 The release of substances from microgels by diffusion may be rate controlled by the polymer
857 network density (Klinger and Landfester, 2012). For instance, bevacizumab, a therapeutic
858 antibody for cancer therapy, could be released from PEG-based microgels prepared by thiol-
859 en click chemistry with an apparent correlation of release rate and microgel mesh sizes
860 (Gregoritz et al., 2018). Moreover, it may also be possible to slowly release substances with
861 hydrodynamic diameters larger than the nominal average mesh size of the network considering
862 the high mobility of hydrated polymer chain segments and common network inhomogeneity.

863 In hybrid microgels, features contributed by their components can be combined. Liposomes
864 added to the microgel-forming dispersed phase prior to crosslinking offer opportunities to buffer
865 rapid release of hydrophilic drugs by incorporation in the liposome core. Furthermore,
866 liposomes allowed the loading of microgels with hydrophobic compounds by solubilization in
867 the double layer of liposomes (Jeong et al., 2014).

868 In addition to the use of drug-compatible crosslinking strategies, the payload may be
869 incorporated after network formation by diffusion, e.g., by soaking microgels in drug solution.
870 Apparently, when drug easily diffuses into the microgel during loading, the gel presents a
871 limited diffusion barrier only and drug release will be relatively fast if no binding motives exist
872 inside the gel. Therefore, coating strategies for hydrogels, as also applicable on the
873 macroscale (Wischke et al., 2013), might allow to hinder the immediate release of hydrophilic
874 compounds from microgels. For instance, this concept was explored by covering anionic
875 hyaluronic acid-based microgels with cationic poly(L-lysine) after protein loading (Labie et al.,
876 2019).

877 Switchable swelling is another main principle to rate drug release from microgels. Through,
878 e.g., temperature- or pH-induced increase of microgel hydration, diffusivity and thus release of
879 incorporated substances may be triggered at the time of interest (Klinger and Landfester, 2012)
880 (Agrawal and Agrawal, 2018). However, when inducing microgel shrinkage to switch-off the
881 release, an outwards directed fluid flow might temporarily boost release if no additional
882 principles of drug retention such as ionic interactions with the gel matrix or covalent drug
883 binding are implemented.

884 Beside hydrophilic polymers, also hydrophobic materials can be of interest to synthesize
885 network materials in microfluidic droplets. Such micronetwork particles, possessing a thermal
886 transition (e.g. a melting transition) above body temperature, may be deformed above this
887 temperature and fixed in the shape of interest through the shape-memory effect (Friess et al.,
888 2014). By microfluidic templating and subsequent crosslinking of PCL-based micronetworks,
889 spherical monodisperse particles were produced that could be loaded by swelling with
890 hydrophobic model compounds. The monodisperse particles could subsequently be
891 transformed and fixed as prolate ellipsoids of various aspect ratios, all originating from the
892 same mother particles for best comparability. Those micronetworks were used to explore
893 shape, size, and time dependency of phagocytic uptake, which is of interest for intracellular
894 delivery (Friess et al., 2019).

895

896 **4.2 Particle designs with complex structures**

897 The ability of microfluidic devices to dose defined quantities of liquids into or aside of each
898 other and subsequently apply manipulation including externally triggered crosslinking opens a
899 wide field of possible particle (sub)structures, e.g., different shapes, compartments, local
900 physicochemical properties of particles as well as payloads. Most prominent are Janus
901 particles, which contain at least two hemispheres with distinctly different properties such as
902 different matrix materials or different payloads. Such particles can be prepared by coaxial flow
903 of two inner phases injected side-by-side into one outer phase, where they are immediately
904 stabilized, e.g., by crosslinking. The volume ratio of the dispersed phases, the time until
905 crosslinking, and the presence of surfactants can be critical when building Janus particles from
906 two distinctly different immiscible materials, as they may undergo a transition towards a
907 capsule-like shape driven by reduction of interfacial energies (Khan et al., 2014). When the
908 two hemispheres should be composed of distinctly different materials, sufficient adhesion
909 between the half-spheres can also become a critical issue.

910 As an alternative to side-by-side injection of the dispersed phases, Janus particles can be
911 obtained through manipulation and/or phase separation of single or double emulsions, i.e.,

912 starting from an initially concentric arrangement. In this case, geometric assistance can be
913 provided by collection channels with a smaller height than the diameter of double emulsion
914 droplets resulting in droplet squeezing and stretching. The resulting viscous friction induced a
915 dislocation of the inner droplet of the double emulsion to the end of the plug of the middle
916 phase, in which the particle shape could be fixed by online UV irradiation associated with matrix
917 material polymerization (Chen et al., 2009).

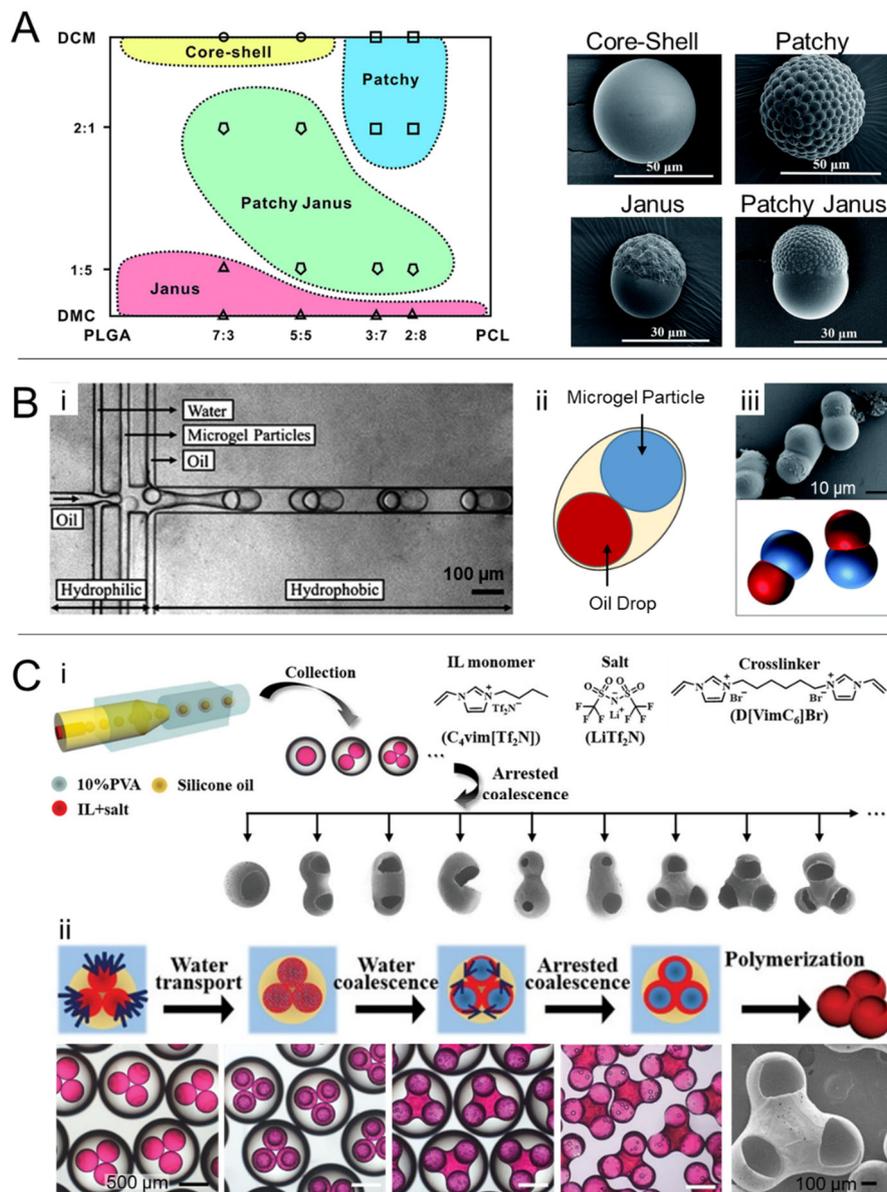
918 Other concepts rely on phase separation, e.g., starting from co-dissolved polymers in single
919 emulsion droplets, which continuously phase separate when the solvent is extracted. In such
920 processes, the nature and concentration of polymer, solvent, and surfactants are of high
921 relevance, as they define the interfacial tensions between the two different polymer-rich
922 phases and the continuous phase. In case of resulting negative spreading parameters, both
923 dispersed polymer-rich phases can undergo dewetting and minimize their surface-to-volume
924 ratio by forming half spheres. Depending on the precipitation kinetics of the two materials, a
925 full phase separation of the two initially co-dissolved materials to Janus particles has been
926 observed, e.g., for PLA-polystyrene model systems (Min et al., 2014). For phase separation of
927 PLGA/PCL mixtures from o/w emulsions, the o-phase solvent was demonstrated to massively
928 affect spreading parameters and thus particle characteristics. While dimethyl carbonate
929 resulted in Janus particles at all PLGA/PCL ratios due to always negative spreading
930 parameters of the polymer phases, dichloromethane produced PCL-PLGA core-shell or golf-
931 ball like particles due to preferential wetting of the PLGA phase (Fig. 11A) (Cao et al., 2015).

932 Particles with characteristic anisotropic shapes can also be obtained from double emulsions
933 with several inner phase droplets. These inner phase droplets can serve as a mask over which
934 a thin polymer layer is stretched after crosslinking of low-concentration solutions of polymer
935 precursors in the middle phase and subsequent solvent extraction. When one microgel (m)
936 particle and one oil droplet were co-encapsulated in $(m+o_1)/w/o_2$ double emulsions with a
937 relatively low volume of the w-phase, the immiscible inner phase objects (m, o_1) were located
938 next to each other and could slightly deform the w-phase droplet (Fig. 11B). The w and the o_2

939 phases can then be simultaneously solidified by acrylate chemistry, leading to marked
940 anisotropic particle shapes after drying (Thiele and Seiffert, 2011).

941 A partial coalescence of several inner phase droplets from identical materials in
942 double/multiple phase emulsions may occur depending on the quality of their interface
943 stabilization. This coalescence may be time-dependent statistic events, thus requiring
944 immediate fixation of a desired intermediate shape by a fast middle phase polymerization.
945 Alternatively, coalescence may be controllable with relatively stable intermediate states , which
946 supports to obtain well-defined geometries in a reproducible manner. Such states were
947 realized for Pickering emulsions due to jamming of nanoparticulate stabilizers at the interfaces
948 of coalescing inner phase droplets (Studart et al., 2009). An arrested coalescence was also
949 reported for $o_1/o_2/w$ double emulsions containing an o_1 -phase from ionic liquids and salt as
950 osmotically active agent. The diffusion of water into the o_1 phase eventually led to a $w/o_1/o_2/w$
951 system. An intermediate state with a stability over several days is adapted after partial
952 coalescence of the o_1 phases within each o_2 droplet, allowing crosslinking of the o_1 phase by
953 acrylate chemistry (Fig. 11C) (Feng et al., 2019).

954 Stop-flow microfluidics is an alternative technique to produce various particle shapes. It differs
955 from droplet-based microfluidics by (i) operating commonly with the flow of one single phase
956 fluid that (ii) is repetitively stopped for channel illumination through a mask defining the volume
957 accessible for polymerization (Bong et al., 2014). Pulsed-laser illumination can also be applied
958 in droplet-based microfluidics at the collection channel (Krüger et al., 2019), thus defining the
959 portion of droplets/plugs that will be solidified. This technique, however, demands a perfect
960 synchronization of drop formation/passage rate and laser pulses in the illumination area and
961 an extremely fast polymerization for reproducible particle shapes.



962

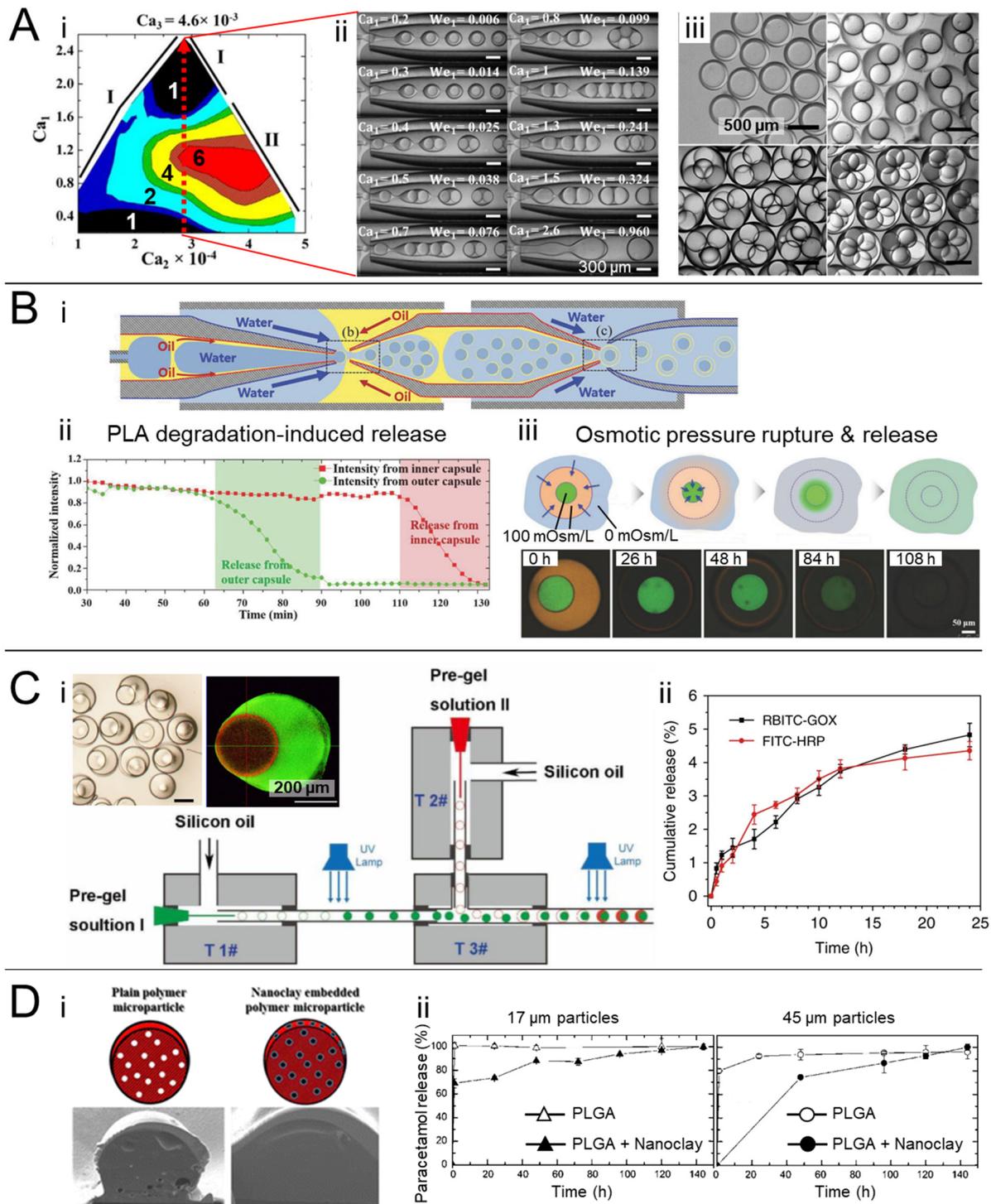
963 **Fig.11:** Designing complex shape particles from droplets template by microfluidics. (A) Phase separation of o-
 964 phases during solvent extraction from o/w single emulsion droplets containing co-dissolved PCL and
 965 PLGA at different ratios, resulting in different shapes including Janus particles depending on the
 966 employed solvent and PLGA/PCL ratio. Republished with permission of The Royal Society of Chemistry
 967 from (Cao et al., 2015); permission conveyed through Copyright Clearance Center, Inc. (B) Janus particles
 968 formation by co-injection of oil phase and microgel (m) particles in (m+o₁)/w/o₂ double emulsions. (i)
 969 Experimental setup. (ii) Scheme of double emulsion droplet containing one microgel particle and one oil
 970 droplet. (iii) Images and cartoon of dried Janus particles after o₁ and w-phase polymerization.
 971 Republished with permission of The Royal Society of Chemistry from (Thiele and Seiffert, 2011);
 972 permission conveyed through Copyright Clearance Center, Inc. (C) Multi-unit particles formed by
 973 arrested coagulation and crosslinking of ionic liquid phase (o₁) after salt-induced water penetration into
 974 o₁. (i) Scheme of emulsification and reagents employed in o₁ phase. (ii) Range of possible particle
 975 shapes. (iii) Exemplary procedure to produce triple-core particles. Adapted with permission from (Feng
 976 et al., 2019) © 2019 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim.

977

978 The preparation of double emulsions by microfluidics offers the opportunity to incorporate
979 several inner-phase droplets per particle (compare Fig. 11). For incorporation of a
980 pharmaceutical compound, this may be advantageous as unintentional damage of one inner-
981 phase compartment during particle handling or application would not necessarily result in
982 leakage of the entire payload of the particle. At the same time, the production of multi-core
983 double emulsions is linked to specific demands regarding flow conditions and device
984 geometries. Two-step devices, where the orifices for the first and second emulsification ($w_1/o/2$)
985 were spatially separated and followed one after the other, had the advantage of only minor
986 impact of w_1 flow on the sizing of o-phase droplets (Shao et al., 2013). By adjusting flow rates
987 and thus controlling the capillary numbers for producing $w_1/o/w_2$ emulsions, a transition in
988 encapsulation from one to six to one droplet(s) could be observed when increasing the w_1 flow
989 rate, while transition regions produced fluctuating numbers of w_1 droplets (Fig 12 A) (Nabavi
990 et al., 2017b).

991 Multiple inner phase compartments can also be produced inside each other leading to higher
992 order core shell structures. When thin shell capsule-in-capsule PLA particles were formed from
993 $w_1/o_1/w_2/o_2/w_3$ quadruple emulsions, the payload of the w_1 and w_2 phases could be
994 consecutively released in accelerated PLA degradation experiments or through osmotic
995 pressure induced capsule wall rupture (Fig. 12B) (Lee et al., 2018). Another concept presented
996 two identical hydrophilic phases each containing a different protein payload. To obtain such
997 particles without mixing of the two dispersed phases, one protein loaded PEG-DA droplet was
998 first prepared and crosslinked followed by injection and fusion with a second protein loaded
999 droplet, which again was crosslinked by UV irradiation (Fig. 12C). Despite intended by the
1000 authors for a use as synthetic cells comprising two enzymes in different hydrogel
1001 compartments, the concept may be applicable also to pharmaceuticals as a slow protein release
1002 from the gel has been observed (Tan et al., 2017). Alternatively, such constructs might, in the
1003 future, be considered for in situ on-demand synthesis of bioactive molecules in the body by an
1004 enzymatic cascade.

1005 By loading microparticles with colloidal particles, either as filler component or with incorporated
1006 drug, composite particles can be obtained. PLGA particles (45 μm) containing dispersed
1007 nanoclay showed a slower paracetamol release compared to particles without filler as
1008 assigned to enhanced particle density and drug diffusion path length, while this effect was less
1009 prominent for 17 μm PLGA particles (Fig. 12D) (Ekanem et al., 2015). Microgel particles
1010 incorporating ketoprofen loaded polyacrylate nanoparticles have been prepared by coupling a
1011 micromixer for nanoemulsion preparation with a coaxial microfluidic device for microgel
1012 templating followed by simultaneous UV crosslinking. In this case, the drug loaded
1013 nanoparticles could liberate from the covalently crosslinked microgel with time (Khan et al.,
1014 2015). A release of nanoparticles from Janus microgel particles has also been reported by
1015 disintegration of physical crosslinks in response to temperature and/or pH changes (Chen et
1016 al., 2013). Overall, by complex particle designs, a number of release principles may become
1017 accessible that add to the most common diffusion-based release of single compounds from
1018 particulate carriers.



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Fig.12: Concepts for drug incorporation and release in multi-compartment particles. (A) Tuning numbers of encapsulated droplets in $w_1/o/w_2$ emulsions. (i) Color map correlating Ca_1 (w_1 phase) and Ca_2 (o-phase) at constant Ca_3 (w_2 phase) with the number of formed inner phase droplets. Non-marked regions are transition regions where fluctuating numbers of droplets have been encapsulated. (ii, iii) Images of droplet pinch-off and final emulsion. Adapted from (Nabavi et al., 2017b) under a Creative Commons Attribution 4.0 International License (<https://creativecommons.org/licenses/by/4.0/>). (B) Capsule-in-capsule PLA particles. (i) Formation of $w/o/w/o/w$ quadruple template emulsion. (ii) Consecutive release of two dyes from the outer and inner capsule in accelerated degradation condition (basic methanol solution). (iii) Loading both capsule cores with salt (100 mOsm/l) and placing them in distilled water caused water influx

1029 (blue arrows) and cracking of capsule walls facilitating dye release by osmotic triggers. Adapted with
1030 permission from (Lee et al., 2018) © 2018 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim. (C) Double
1031 compartment PEG-based hydrogel particles with protein payload. (i) Optical and fluorescence
1032 microscopic images. (ii) Preparation scheme with droplet injection and crosslinking. The second inner
1033 phase droplet fuses with the presolidified first inner phase droplet. (iii) Release of dye-labeled glucose
1034 oxidase (GOX) and horseradish peroxidase (HRP). Reproduced from (Tan et al., 2017) under a Creative
1035 Commons Attribution 4.0 International License (<https://creativecommons.org/licenses/by/4.0/>). (D)
1036 Composite PLGA particles with 2 wt.% incorporated nanoclay (plate-like montmorillonite). (i) Scheme
1037 and SEM images of particles with reduced porosity. (ii) Paracetamol release for 17 µm and 45 µm
1038 particles with or without nanoclay filler. Adapted with permission from (Ekanem et al., 2015), Copyright
1039 (2015) American Chemical Society.

1040

1041 **Conclusions**

1042 Droplet-based microfluidic techniques represent a continuously growing field, where many
1043 fundamental principles have been established that now can facilitate the efficient preparation
1044 of functional polymer particles. As instrumentation for microfluidics and technical services in
1045 chip manufacturing became increasingly available, it can be expected that droplet-based
1046 microfluidics will continue to attract more scientists that aim at encapsulating drug molecules
1047 into tailor-made polymeric carriers.

1048 For some of the potential beneficial features (compare section 4) that microfluidic
1049 manufacturing may have for the preparation of drug carriers, some pieces of literature already
1050 indicate that this technology may meet the described expectations. For some of the other
1051 proposed points, including shear sensitive compound encapsulation and multi-step local
1052 release, experimental proof demonstrating advantages of microfluidic processing is still to
1053 come.

1054 The use of droplet-based microfluidics in pharmaceuticals will, despite the availability of far
1055 progressed methodologies to increase throughput, likely not be in mass products with tons of
1056 annual sales. Instead, high value products, presumably with small batch sizes might be a more
1057 likely scenario. Given the possibility for aseptic manufacturing and the beneficial feature of
1058 microfluidics to allow handling of small volumes, possible applications in pharmaceuticals may
1059 be found in precision medicine. Examples may be particles incorporating patient material, such
1060 as for cancer vaccination, or the use of prefilled microfluidic devices by which the quantity of

1061 encapsulated doses for slow-release formulations or on-demand release systems may be
1062 individually defined based on patient-specific needs.

1063

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1068

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