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# Characterization and modeling of the influence of artificial ageing on the microstructural evolution of age-hardenable AlSi<sub>10</sub>Mg(Cu) aluminum alloys

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## 1 Abstract

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A comprehensive analysis of the effect of the artificial ageing stage of the precipitation heat treatment T<sub>6</sub>-T<sub>7</sub> process on the microstructure of the age-hardenable AlSi<sub>10</sub>Mg(Cu) aluminum alloy is presented that considers the influence of temperature and time of the ageing conditions on the formation and evolution of the Mg<sub>2</sub>Si strengthening precipitate distributions. A complete quantitative characterization of the strengthening distributions covering a broad range of ageing conditions was obtained using the small angle neutron scattering (SANS) technique, complemented with transmission electron microscopy (TEM) and high-resolution transmission microscopy (HTEM). This information was successfully used to fit Robson's precipitation model for the coupled nucleation, growth, and coarsening of Mg<sub>2</sub>Si strengthening precipitates. A unique set of modeling parameters was obtained for the whole precipitation process. Robson's model is shown to be a powerful tool for predicting the evolution of these nanometer-scale particles in industrial and complex ageing processes, which are critical for designing new components based on the material requirements.

**Keywords:** Aluminum alloy, Precipitation strengthening, Ageing, Small Angle Neutron Scattering, Modeling

## 2 Introduction

The literature has established that the macromechanical performance of age-hardenable alloys is defined by the size distribution of their strengthening precipitates [1–4]. Al–Si–Mg alloys are typical examples of this type of material. These alloys are commonly used in the automobile sector to produce different components [5]. The investigated alloy in this report is the AlSi<sub>10</sub>Mg(Cu). The composition of the alloy is given in Table 1. In this case, the material strengthening occurs because of the formation of Mg<sub>2</sub>Si precipitate distributions during the T<sub>6</sub> heat treatment process [6–9].

Si	Mg	Sr	Ti	Fe	Cu	Zn	Mn
10.0–11.0	0.35–0.50	max. 0.05	0.05–0.15	max. 0.55	max. 0.3	max. 0.3	0.2–0.5

Table 1: The chemical composition of the alloy VDS 233 AlSi<sub>10</sub>Mg(Cu).

A T<sub>6</sub> heat treatment is a precipitation process in which a fine distribution of a hardening phase is created under specific and controlled temperature and time conditions to interact with dislocations to produce the desired material strengthening [10, 11]. In brief, this type of heat treatment starts with a solution treatment at a high temperature to create a supersaturated solid solution state (SSSS) of the alloy. After water quenching at room temperature, the SSSS decomposes, creating a homogeneous size distribution of strengthening precipitates during the artificial ageing process. Mg<sub>2</sub>Si precipitates nucleate first in the form of metastable and coherent β'' phases [12]. This initial precipitate distribution does not strengthen significantly the material [11]. However, if the duration of ageing is increased, the evolution of this distribution to the metastable and semi-coherent β' phase produces the maximum strengthening of the material [12]. For a sufficient ageing time, a total incoherent distribution of the stable β phase is finally obtained [12], which lowers the strength of the material as the duration of the artificial ageing increases [13](Fig. 1). Increasing the ageing temperature accelerates the precipitation process and produces a coarser distribution [10, 11].

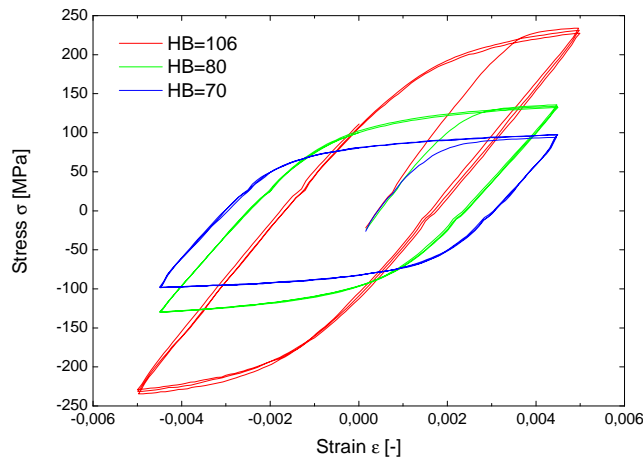


Figure 1: Stress–strain curves from the three first cycles of low cycle fatigue (LCF) tests performed at 250°C for three different hardness states, i.e., ageing conditions. HB denotes the Brinell hardness. Higher hardness states correspond to shorter ageing durations.

Although the size, shape, and approximate composition of the  $\text{Mg}_2\text{Si}$  strengthening precipitates of this alloy [14–17], as well as the precipitation sequence [12] and the strengthening mechanisms [13, 18] have been well-established in the literature, the influence of the ageing conditions of the  $\text{T}_6$  precipitation hardening process on the formation of the  $\text{Mg}_2\text{Si}$  strengthening distributions is not fully established [1, 2, 9]. Additionally, the size distribution of the  $\text{Mg}_2\text{Si}$  strengthening precipitates created in the material has not been quantified. Small angle neutron scattering (SANS) investigations, complemented with transmission electron microscopy (TEM) and high-resolution transmission microscopy (HTEM), is an ideal option for deriving valuable and statistically meaningful experimental information about the strengthening distributions obtained according to different temperature and duration ageing conditions. The investigations in this work are focused exclusively on the artificial ageing process and the equilibrium  $\beta$ - $\text{Mg}_2\text{Si}$  precipitates. Thus, the heat treatment should be denoted as a  $\text{T}_7$  process because the so-called over-ageing stage is of great interest.

Despite successful quantitative characterization of the size distributions of the  $\text{Mg}_2\text{Si}$  strengthening precipitates at various temperatures and durations of the ageing process, the effect of the variation of the ageing conditions on the size distributions cannot be quantitatively predicted. Moreover, the experimental characterization of multi-stage or complex ageing processes are also costly and endless. For these reasons, data from the SANS and TEM investigations are used to fit Robson’s precipitation model for the  $\text{Mg}_2\text{Si}$  strengthening precipitate distributions [3, 19–22]. This model predicts the size distributions of the  $\text{Mg}_2\text{Si}$  strengthening precipitates in both one- and multi-stage artificial ageing processes over a large range of ageing temperatures as well as the whole precipitation sequence from the early nucleation to the final coarsening processes for a unique set of model parameters [22]. Adapting Robson’s model will allow it to be applied for designing and optimizing the artificial ageing processes in an industrial setting to achieve the required mechanical properties for the  $\text{AlSi}_{10}\text{Mg}(\text{Cu})$  alloy.

### 3 Microstructure Analysis

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The information obtained about the dendrite arm spacing (DAS) of the aluminium matrix dendrites; length, orientation, and form of the  $\text{Al}_5\text{FeSi}$  needles; and size and circularity of the eutectic silicon particles are provided in the following sections. Some comments about other less common intermetallic precipitates present in the microstructure configuration of Al–Si–Mg alloy systems are also included.

#### 3.1 Sample preparation

The secondary alloy for this study was melted in an electric furnace at  $730^\circ\text{C}$  and was then poured into metallic molds pre-heated at  $200^\circ\text{C}$ . Once the gate system was removed, bars of 270 mm in length and 23 mm in diameter were obtained. Prior to heat treatment, disc-shaped samples of 10 mm in thickness were prepared from the same bar. All samples were solution treated at  $535^\circ\text{C}$  for 6 hours followed by quenching in water at room temperature.



The samples were then artificially aged at two different temperatures. A temperature of 210°C was used to create samples aged until they reached the early stages of nucleation and were close to the peak hardening, while samples aged at 280°C provided over-aged states. The time between the quenching stage and the beginning of the artificial ageing is critical because this alloy can also precipitate at room temperature, forming small clusters of silicon and magnesium [12]. This time gap is limited to a maximum of 10 minutes. Table 2 lists the different ageing temperature and duration conditions for the seven experimental states and their corresponding hardness measurements.

The following discussion is based on results obtained for states II and VII, corresponding to the peak-aged and the long over-aged states, respectively.

State	Hardness	Ageing Conditions	
		$T_{\text{ageing}}$	$t_{\text{ageing}}$
	HBW 2.5/62.5/20	°C	h
I	119.4	210	0.2
II	124.8	210	0.5
III	108.2	210	2.0
IV	84.0	280	0.5
V	63.7	280	19.0
VI	56.1	280	69.0
VII	50.1	280	211.0

Table 2: The microstructure and collecting hardness values as well as the temperature and length of the artificial ageing process for selected ageing states are listed. The designation “HBW 2.5/62.5/20” for hardness specifies the use of a tungsten carbide ball indenter for the Brinell test with a diameter of 2.5 mm applying a 62.5 kgf force maintained for 20 s.

### 3.2 Dendrite Arm Spacing (DAS)

All aluminum alloy castings exhibit a characteristic dendritic solidification microstructure [23]. The effect of dendrite arm spacing on the mechanical properties of aluminum alloy industrial components has been extensively studied [24].

The results for the peak-aged and long over-aged samples are shown in Figs. 2(a) and 2(b). Here, the variation in the DAS scale in the whole ageing stage is not relevant.

Any change in the DAS is expected to occur during the solution heat treatment or quenching stages [6]. The spacing between adjacent arms in the dendrites is mainly determined by the solidification and cooling stages in the casting process, but is also influenced by the solute content [23]. The DAS decreases with decreased solidification time [23].

The results of the dendrite arm spacing (DAS) measurements are presented in Table 3. The spacing between dendrite arms influences the mechanical properties of the AlSi<sub>10</sub>Mg(Cu) alloy. However, it is not responsible

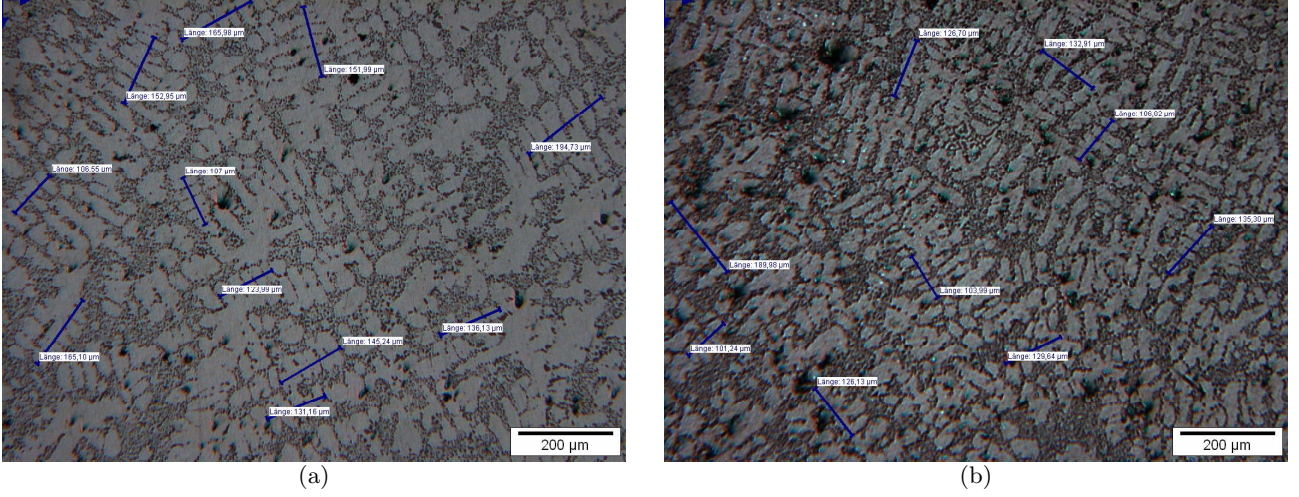


Figure 2: Micrographs showing the DAS measurements: (a) for the peak-aged state; (b) for the long over-aged state. The DAS is not altered during the artificial ageing process.

for the observed variation in the mechanical performance of the material because it does not change during the ageing process.

Sample	DAS $\mu\text{m}$	Si eutectic $\mu\text{m}$	$\text{Al}_5\text{FeSi}$ $\mu\text{m}$
I	22.90	1.63	24.34
II	22.53	1.64	12.00
III	19.81	1.70	6.13
IV	21.02	1.53	18.16
V	22.36	1.55	8.48
VI	22.76	1.79	9.81
VII	22.67	1.57	16.22

Table 3: Results of the DAS measurements, the average diameter of the silicon eutectic particles, and the length of the  $\text{Al}_5\text{FeSi}$  needles for the seven ageing states are presented. The roundness ratio of the silicon eutectic is approximately 80%.

### 3.3 Eutectic silicon particles

Alloys containing an excess of silicon above the stoichiometrically required amount to form  $\text{Mg}_2\text{Si}$  strengthening precipitates exhibit a modified composition and density of the particles [13] and preferentially form additional phases during the latter stages of ageing. However, this excess of silicon does not contribute significantly to the material strengthening [25].

The required characteristics to describe a silicon particle distribution are particle size, shape, and spacing. Figs. 3(a) and 3(b) show that the size and roundness of the eutectic silicon particles do not change during the artificial ageing process. The silicon eutectic sizes are shown in Table 3. The roundness ratio  $R$  is a parameter defined inside the image analysis software used, which is linked to the area  $A$  of the analyzed particle and the

longest distance between two points in the perimeter of the particle  $L$  and is calculated as shown in Eq. (1). A roundness value close to 100% is characteristic of spherically-shaped particles. In this study, the roundness of the silicon particles is approximately 80%.

$$R = \frac{4A}{\pi L} \cdot 100. \quad (1)$$

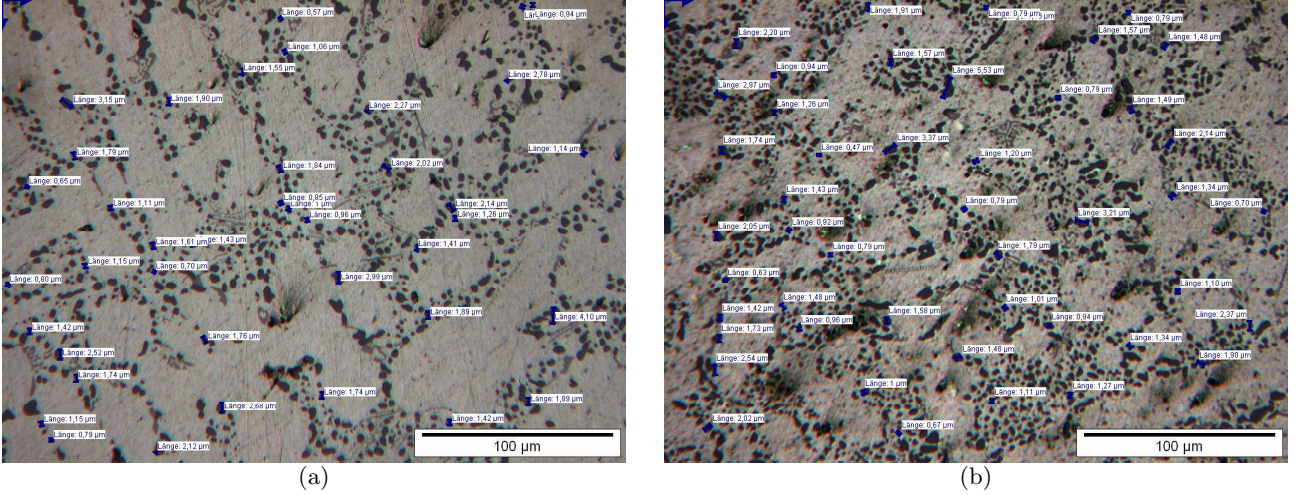


Figure 3: Micrographs showing Si particles: (a) for the peak-aged state; (b) for the long over-aged state. The size of the eutectic particles remains unchanged.

In fact, the silicon particles are altered during solution heat treatment at 535°C prior to the ageing process. The fibrous morphology of eutectic silicon upon solidification is transformed into spherical particles due to the addition of a strontium chemical modifier to the chemical composition of the alloy. During the solution heat treatment stage, the well-known spheroidization process of the silicon particles also occurs. Eutectic silicon particles coarsen during the 6 hours solution heat treatment phase due to the Ostwald ripening phenomenon, thereby reducing the total interfacial energy between the aluminum matrix and the silicon particles [6]. Fig. 4 shows two micrographs that depict the remarkable effect of the solution treatment on the morphology of the silicon particles.

The eutectic silicon particles contribute to the global strengthening of the material and must be considered [25]. However, because there is no change in the size, spacing, or roundness of these precipitates during the ageing process (Table 3), just as observed for the dendrite arm spacing, the eutectic silicon particles are eliminated as a cause for the change in the mechanical properties of the alloy [25].

### 3.4 Al<sub>5</sub>FeSi needles

Adding iron to the AlSi<sub>10</sub>Mg(Cu) alloy for chill-casted components is not preferable because it causes the material to become more brittle [23, 26]. The iron content comes from the secondary nature of the alloy. This element content forms different Fe intermetallic precipitates, such as the Chinese script Al<sub>15</sub>(MnFe)<sub>3</sub>Si<sub>2</sub> and Al<sub>8</sub>Mg<sub>3</sub>FeSi<sub>6</sub> phases, but most Fe is present in the form of Al<sub>5</sub>FeSi needles (shown in Fig. 5).



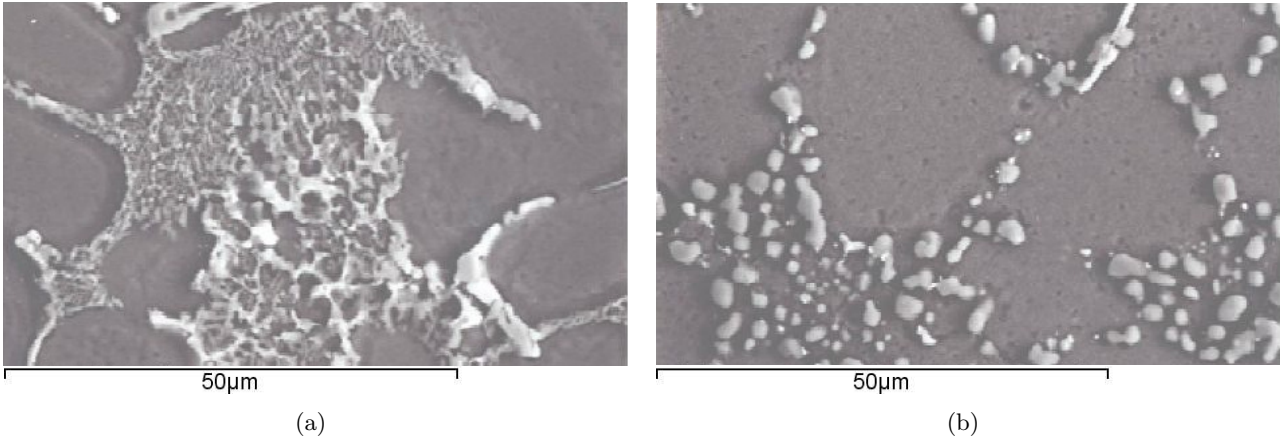


Figure 4: Micrographs showing eutectic silicon particles in the interdendritic area: (a) at the as-cast state prior to heat treatment (fibrous morphology); (b) after solution heat treatment and prior to the ageing stage (spheroidal morphology).

The length of these  $\text{Al}_5\text{FeSi}$  needles is mostly determined during the solidification and cooling stage of the casting process. The slower the cooling rate after melting, the longer the needles become. Compared to the analyzed intermetallic phases, the length of the  $\text{Al}_5\text{FeSi}$  needles is remarkably scattered, even for measurements performed for the same state in different samples. However, no trend is observed, and the variation in length is completely random, suggesting an inhomogeneous distribution of iron in the casted material because, to some extent, these  $\text{Al}_5\text{FeSi}$  needles have not homogeneously precipitated in the interdendritic zone.

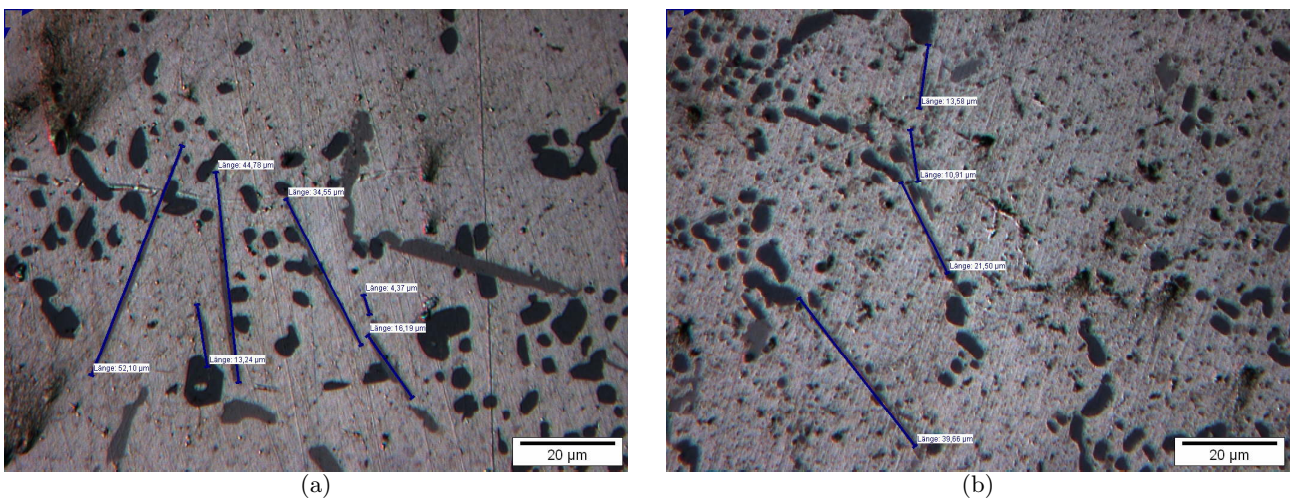


Figure 5: Micrographs showing  $\text{Al}_5\text{FeSi}$  needles with scattering lengths: (a) for the peak-aged state; (b) for the long over-aged state.

Additionally, at the solution heat treatment stage, the  $\text{Al}_8\text{Mg}_3\text{FeSi}_6$  phase decomposes slowly to release magnesium into the solution, which causes more  $\beta\text{-Mg}_2\text{Si}$  strengthening precipitates to nucleate and coarsen [13]. This decomposition produces very fine iron-rich precipitates, probably of the  $\text{Al}_5\text{FeSi}$  phase [6], which explains why the typical Chinese script form  $\text{Al}_8\text{Mg}_3\text{FeSi}_6$  precipitates in the as-cast state have almost disappeared at any ageing state.

A partial dissolution of the original  $\text{Al}_5\text{FeSi}$  phase formed in the casting process during solution treatment

may be expected [6]. The dissolution of this phase occurs by decomposition of the  $\text{Al}_5\text{FeSi}$  into  $\text{Al}_6\text{Fe}$  and Si particles. Excess silicon then contributes to the  $\text{Mg}_2\text{Si}$  strengthening precipitates and may also explain the scattered length measurements of the  $\text{Al}_5\text{FeSi}$  needles.

The intermetallic  $\text{Al}_5\text{FeSi}$  phase in which the majority of the iron content is present is not considered as a key factor in explaining the change in the alloy's mechanical properties because there is no systematic trend in the change of the length of the  $\text{Al}_5\text{FeSi}$  needles, as shown in Table 3. Regardless, its presence in the material causes an undesirable embrittlement of the material.

## 4 Characterization of the $\text{Mg}_2\text{Si}$ strengthening precipitates

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This section presents the results obtained for the characterization of the  $\text{Mg}_2\text{Si}$  strengthening precipitates using SANS and TEM/HTEM. The SANS measurements were performed with the instrument SANS-2 at the German Engineering Materials Science Center (GEMS) in Geesthacht (Germany) in 2010. These tests were validated with TEM/HTEM analyses at the Helmholtz-Zentrum Geesthacht Center for Materials and Coastal Research. Nanometer-scale  $\text{Mg}_2\text{Si}$  particles are the main strengthening precipitates that strengthen the general Al-Si-Mg alloy systems [6–9]. While the precipitation sequence has been extensively reported in the literature, emphasizing the first stages of nucleation and the formation of clusters from the decomposition of the supersaturated solid solution state of the alloy [12, 27, 28], the distributions of these  $\text{Mg}_2\text{Si}$  strengthening precipitates are still unknown.

The presence of copper in the alloy chemical composition due to its secondary nature leads to the formation of  $\text{Al}_2\text{Cu}$  strengthening precipitates. However, this element is present at a maximum of 0.3% in the alloy (Table 1) and is not considered to be responsible for the strengthening during the ageing process.

### 4.1 Small angle neutron scattering measurements

TEM is currently employed to study the evolution of the precipitation kinetics in this kind of alloy [12]. However, this procedure is not ideal due to the reduced material volume considered in the study, which is reduced to very thin foils. For cases where the objective is to analyze a relatively large volume leading to statistically meaningful results, scattering techniques that use synchrotron or neutron sources are ideal alternatives [29].

Despite the benefits of the small angle X-ray scattering procedure, such as its high resolution and short measurement time [30, 31], this technique has problems characterizing these nanometer-scale particles because both solutes (magnesium and silicon) involved in the nucleation, growth, and coarsening of these precipitates have a similar atomic number as the aluminum matrix, resulting in a negligible contrast between the matrix and the precipitates [29, 32]. SANS does not have this restriction and has thus been used to characterize the size distribution of the  $\text{Mg}_2\text{Si}$  strengthening precipitates present in this Al-Si-Mg alloy system. SANS principles can be easily found elsewhere, e.g., [33, 34].

### 4.1.1 Sample preparation and experiment

Disc-shaped samples (diameter 23 mm, thickness 3.5 mm) were prepared from the same original bar (bar length, 270 mm) casted as described (Sect. 3.1). These samples were heat-treated prior to performing the SANS measurements. Solution treatment at 535°C for 6 hours and quenching in water at room temperature were conducted for all of the samples before ageing them at 180°C, 210°C, 240°C, and 300°C for different times. A neutron wavelength of 5.8 Å (wavelength spread of 10%) and different detector distances (mostly 1 m, 4.5 m, and 14.0 m) were used with appropriate collimators to cover the scattering vectors from 0.025 nm<sup>-1</sup> to 3 nm<sup>-1</sup> using unpolarized neutrons. The measured intensities were corrected for collecting transmission, background, and detector efficiency effects. Absolute cross-sections were obtained by calibrating against a Vanadium sample.

The macroscopic differential cross-section is calculated from the intensity measured by the detector  $I$  as [34]:

$$\frac{d\Sigma}{d\Omega} = \frac{I}{I_0 \cdot D \cdot \Delta\Omega \cdot T \cdot \eta} . \quad (2)$$

In Eq. (2),  $I$  is the measured intensity flux at the detector,  $I_0$  is the primary beam intensity flux before being scattered,  $D$  is the sample thickness,  $\Delta\Omega$  is the solid angle covered,  $T$  is the sample transmission, and  $\eta$  is the detector efficiency.

### 4.1.2 SANS Results

The neutron scattering cross-sections for the four ageing temperatures applied for different ageing times are presented in Fig. 6. These values agree with previous SANS investigations also performed for similar Al-Si-Mg alloy systems [9]. The scattering intensity curves exhibit the typical shoulder contribution at intermediate scattering vectors  $q$  of a unique distribution of strengthening precipitates of nanometer-scale (basically,  $\beta'$ - and  $\beta$ -Mg<sub>2</sub>Si precipitates according to the precipitation kinetics [12, 27, 28]) for every ageing temperature with no cross-over at large scattering vectors  $q$ .

As the ageing time period is extended to smaller scattering vector  $q$  ranges, the shift of the characteristic shoulder confirms that nucleation, growth, and coarsening of the Mg<sub>2</sub>Si strengthening precipitates are taking place during the artificial ageing stage. Therefore, the initial supersaturated solid solution of magnesium and silicon in the aluminum matrix is decomposed at the beginning of the artificial ageing, forming an initial  $\beta$ -Mg<sub>2</sub>Si strengthening precipitate distribution that evolves to incoherent and coarse distributions as the ageing time increases. This evolution of these strengthening precipitates causes the mechanical properties of the alloy to deteriorate.

The obtained scattering curves are high quality. This feature happens because the artificial ageing treatments were selected before performing the measurements, which enables the use of long data acquisition times. If *in-situ* measurements are being taken, short time intervals (under 30 seconds) must be taken, such as when studying the first precipitation stages, because relatively fast changes in the precipitation are expected at the

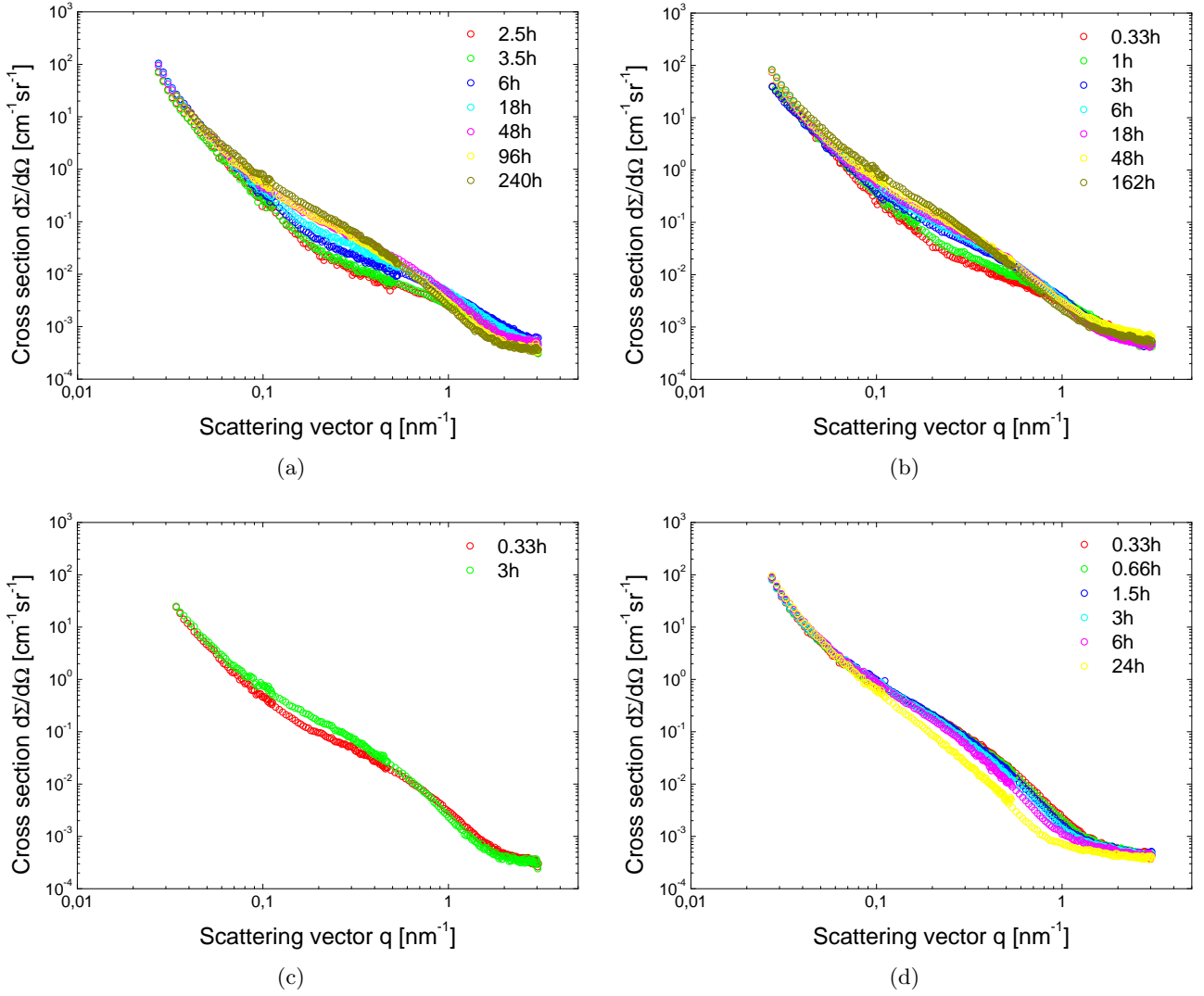


Figure 6: SANS scattering curves obtained from  $\text{AlSi}_{10}\text{Mg}(\text{Cu})$  samples artificial aged at: (a)  $180^\circ\text{C}$  for 2.5 h, 3.5 h, 6 h, 18 h, 48 h, 96 h, and 240 h; (b)  $210^\circ\text{C}$  for 0.33 h, 1 h, 3 h, 6 h, 18 h, 48 h, and 162 h; (c)  $240^\circ\text{C}$  for 0.33 h, and, 3 h; (d)  $300^\circ\text{C}$  for 0.33 h, 0.66 h, 1.5 h, 3 h, 6 h, and 24 h. Note that both axes are plotted on log-scales, and the x-axis is the reciprocal of dimensions in real space.

beginning of the artificial ageing process [29]. Additionally, the sample thickness of measured samples prevents multiple scattering from occurring.

At all of the ageing temperatures in this study, the scattering curves show a constant background using a large scattering vector  $q$  ( $q = 4\pi\sin(\theta)/\lambda$ , where  $2\theta$  is the scattering angle). For small scattering vectors  $q$ , larger structures, such as primary precipitates, inclusions, dispersoids, and grain boundaries, contribute to the effect. Larger structures contribute with a  $q^{-4}$ -tail in the visible  $q$ -range. For example, in the quenched state, where the material is at the supersaturated solid solution state, the scattering curves should be composed of a straight line  $q^{-4}$  contribution and a constant background at large scattering vector  $q$  values.

As the ageing time increases at every ageing temperature, the particle size distribution is expected to dramatically vary from the peak-aged states to the long over-aged ones according to the observed evolution of the scattering cross-sections.

### 4.1.3 Analysis of the size distribution of Mg<sub>2</sub>Si strengthening precipitates

The measured scattering curves should be further analyzed in terms of the size and fraction of precipitates to fit the Robson simulation model for predicting these precipitate distributions, as shown in Sect. 5. The precipitates are assumed to have a rod-like shape, which was confirmed by the TEM images presented in Sect. 4.2. The local monodisperse approximation is adopted for the treatment of interparticle interference because the neutron scattering cross-sections show the characteristic shoulder caused by a high particle number density in the nanometer range [35]. In this approximation, the scattering cross-section of a dispersion of precipitate particles can be described by Eq. (3).

$$\frac{d\Sigma}{d\Omega}(q) = (\Delta\eta)^2 \cdot \int_0^\infty n(R) V(R)^2 F(q, R)^2 S(q, R, f) \cdot dR, \quad (3)$$

where  $d\Sigma/d\Omega$  is the macroscopic differential scattering cross-section,  $\Delta\eta$  is the difference in the scattering length densities between particle and matrix,  $n(R) dR$  is the unknown size distribution of precipitates,  $V(R)$  is the particle volume,  $F(q, R)$  is the particle form factor, and  $S(q, R, f)$  is the structure factor describing the interparticle interference effect [30].

Finally, the particle size distribution  $n(R)$  is approximated by a normal distribution function of the radius:

$$n(R) = \frac{n_0}{\sqrt{2\pi}\sigma} \cdot \exp\left(-\frac{(R - R_0)^2}{2\sigma^2}\right). \quad (4)$$

According to Eq. (4), the maximum and width of the distribution function,  $R_0$  and  $\sigma$ , respectively, and the total density  $n_0$  of the particles contained in the precipitate distributions at the different ageing states are determined by fitting a scattering curve calculated by Eq. (3) to the corresponding measured scattering curves shown in Fig. 6 using the least-squares procedure. This method is automatically computed using software developed by the GEMS Center. The results of this analysis are shown in Fig. 10.

## 4.2 Transmission electron microscopy

TEM/HTEM was used to determine the shape of precipitates and the preferred nucleation sites of the particles for fitting the precipitation model to predict the Mg<sub>2</sub>Si strengthening precipitate distributions from the calculated cross-sections presented in Fig. 6 and to complement the results of the SANS measurements.

### 4.2.1 Sample preparation

Material from ageing state II (Table 2) was investigated using TEM/HTEM techniques. For these investigations, the material was cut by diamond saw, from which a 2.3 mm disc was drilled out and was subsequently ground to a thickness of approximately 100  $\mu\text{m}$ . This disc was thinned electrochemically using a solution of 33% HNO<sub>3</sub>



with methanol at a temperature of  $-30^{\circ}\text{C}$ . A voltage of 20 V was applied, generating a perforated specimen after approximately 1 minute with an average current of 80 mA during thinning.

#### 4.2.2 TEM/HTEM Results

For TEM investigations, the specimen was tilted slightly to record all micrographs along the  $\langle 001 \rangle$  zone axis of the aluminum matrix, in agreement with previously reported analyses of  $\text{Mg}_2\text{Si}$  precipitates [12]. HTEM micrographs of the aluminum matrix showed the expected atomic arrangement under this zone axis: a cubic lattice of atoms within the fcc aluminum. Fourier transformation of the micrographs gave maxima that reassembled the diffraction pattern of an unordered fcc crystal in the  $\langle 001 \rangle$  orientation.

At a low magnification, elongated structures are visible in diffraction contrast in the specimen (Fig. 7(a)). They show no distinct contrast at atomic resolution but essentially retain the atomic structure of the aluminum matrix. From the high resolution pictures, it can be observed that, while distortion is present in the lattice, it is not caused by either a precipitate or by a coherent precipitate with equal crystallographic structure compared with the matrix.

Many small rod-shaped precipitates with a radius in the 5 to 10 nm range can be found in the peak-aged specimen. Two phase contrast images of  $\text{Mg}_2\text{Si}$  strengthening precipitates are shown in atomic resolution in Fig. 7(b). In this figure, when an objective aperture is applied, a diffraction contrast showing distortion of the aluminum matrix is visible. Nevertheless, this distortion contrast cannot directly be related to the precipitates.

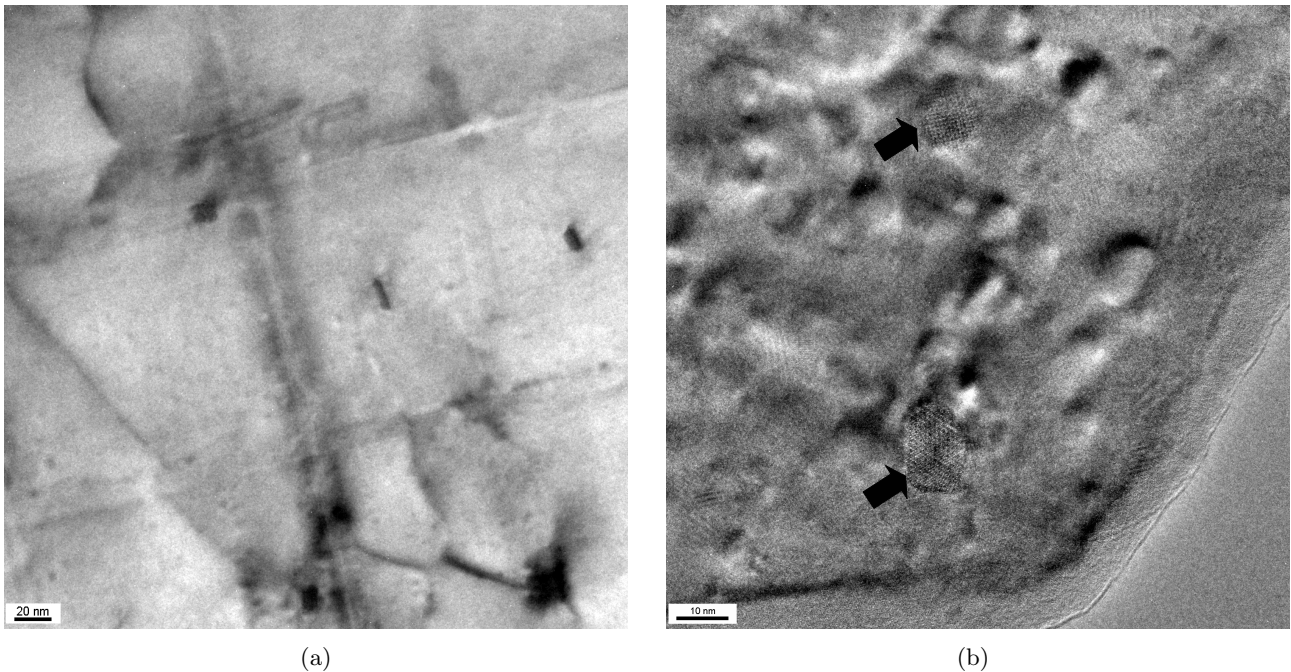


Figure 7: (a) Atomic resolution micrograph of a rod-shaped coherent  $\beta''\text{-Mg}_2\text{Si}$  precipitate in the vicinity of an elongated structure with the same atomic structure as the aluminum matrix; (b) Atomic resolution filtered micrograph of two semi-coherent  $\beta'\text{-Mg}_2\text{Si}$  precipitates.

Higher magnification images of these precipitates reveal an atomic structure similar to those obtained for the

Mg<sub>2</sub>Si phase in previous work [12], which corresponds to the thermodynamic metastable  $\beta'$  precipitate phase. However, the ultimate conclusion could only be drawn from simulations of the HTEM pictures. It is striking that the precipitates were not aligned with the atomic  $\{001\}$  layers of the aluminum matrix; rather, they deviated by approximately  $10^\circ$  (Fig. 8(a)), indicating that these precipitates were not fully coherent with the surrounding matrix, as is expected from metastable  $\beta'$ -Mg<sub>2</sub>Si phases. However, the resolution is not sufficient to determine the continuity of the atomic layers at the interface.

Large precipitates were also present in the specimen (Fig. 8(b)). These large precipitates can be  $\beta'$ - or  $\beta$ -phase. It was not possible to generate pictures with atomic resolution for these structures. However, the crystallography is not cubic based on crystallography results, and implies that these large particles are not  $\beta$ -phase.

Finally, the precipitates and dislocations cannot be clearly correlated, implying that the nucleation of the precipitates takes place homogeneously throughout the aluminum matrix.

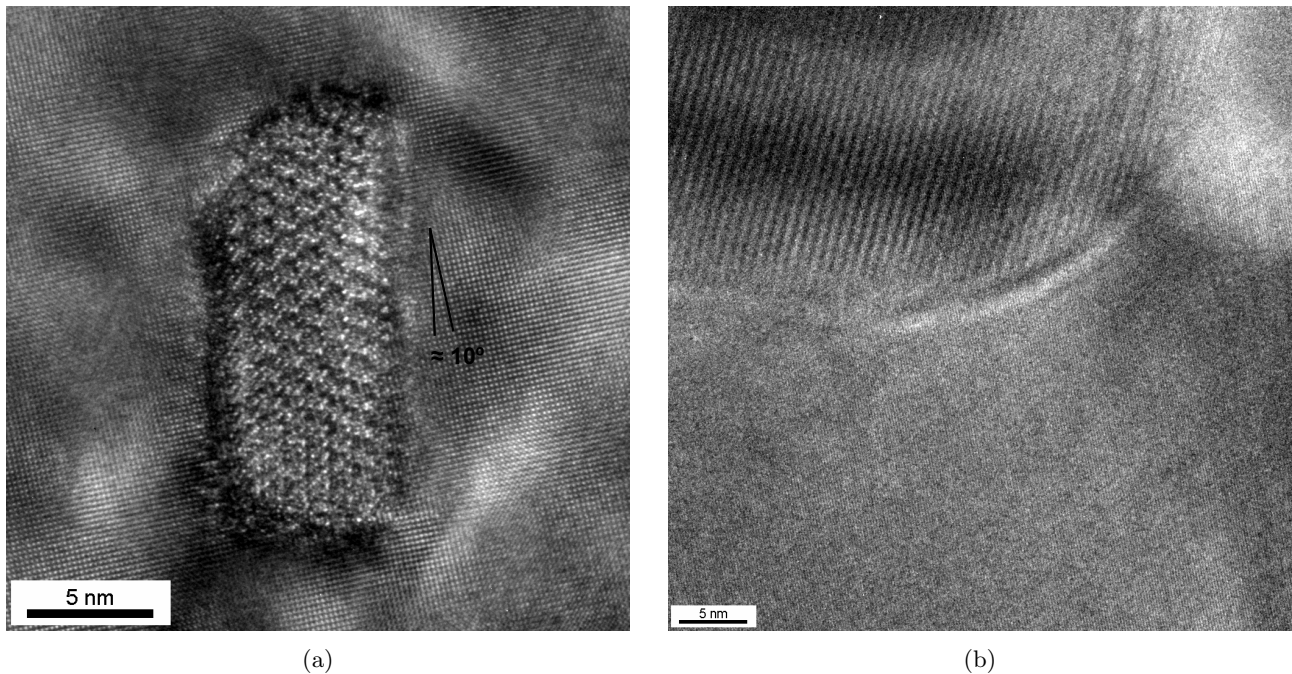


Figure 8: (a) Atomic resolution micrograph of a semi-coherent  $\beta'$ -Mg<sub>2</sub>Si precipitate. A deviation of approximately  $10^\circ$  with respect to the atomic  $\{001\}$  layers of the aluminum matrix is observed; (b) TEM micrograph of a large Mg<sub>2</sub>Si precipitate.

## 5 Robson Model

Most of the simulation models used for predicting the precipitation process, such as the Johnson–Mehl–Avrami model [36] or the cluster dynamic model [37], are focused on the first stages of precipitation. Such models are unable to consider both the competitive growth and the coarsening processes of nucleated precipitates. Furthermore, the applicability and usefulness of these methods has been reported exclusively for isothermal ageing processes [38].

Conversely, there are models that take into account not only the nucleation and growth processes, but also the final coarsening stage when the matrix is depleted of solutes that contribute to strengthening precipitates, such as the original Langer and Schwartz model [39] and the Kampmann–Wagner numerical model [40]. Furthermore, these models are accurate enough to consider non–isothermal and complex artificial ageing treatments. The Kampmann–Wagner simulation model has been used to model the evolution of mechanical properties of Al–Si–Mg alloy systems during multi–step ageing [41]. However these simulation models for the evolution of the strengthening precipitates are fundamentally complicated and require a large amount of input data as well as a precise characterization and analysis of the precipitates. Consequently, these models are reduced to very specific alloy systems and precipitation processes.

## 5.1 Problem description and modeling objectives

Robson’s numerical model consists of a nucleation model for the computation of the number of new precipitate nuclei formed at every step, a growth/dissolution rate for the previously nucleated particles, and a mass continuity equation of the solutes involved in the precipitation kinetics. This simulation model is suitable for dilute alloy systems [20].

This numerical precipitation model, which was further developed from the model proposed by Kampmann and Wagner [40], is used to predict the evolution of the equilibrium  $\text{Mg}_2\text{Si}$  strengthening precipitate distributions based on the temperature and ageing time conditions of the artificial ageing process. The model requires some simplifying assumptions, which are summarized as follows:

- Thermodynamically stable  $\text{Mg}_2\text{Si}$  strengthening precipitates present the stoichiometric composition. These are spherical–shaped precipitates with uniform thermodynamic properties.
- $\text{Mg}_2\text{Si}$  strengthening particles are the only precipitate in which magnesium is significantly soluble.
- Nucleation of  $\text{Mg}_2\text{Si}$  precipitates is homogeneous within the aluminum matrix, as found in TEM investigations performed in Sect. 4.2 and also reported [25].
- The growth and coarsening of  $\text{Mg}_2\text{Si}$  strengthening particles are diffusion–controlled processes in which magnesium diffuses to the interface between the precipitates and the matrix.
- According to alloy composition, all of the magnesium is depleted into solution during the solution heat treatment. Magnesium is the limiting reagent, and silicon is present in excess.
- Because the diffusivity of magnesium and silicon solutes is similar at high temperatures [42], the solid solubility limits are obtained directly from a simple quasi–binary Al– $\text{Mg}_2\text{Si}$  alloy system [43].

According to the formulated assumptions, it is possible to predict the growth and coarsening processes of the  $\text{Mg}_2\text{Si}$  precipitates without differentiating the characteristics of the different strengthening phases obtained during the precipitation sequence rather than needing to consider separate treatments for the  $\beta''$ –,  $\beta'$ –, and

the  $\beta$ -Mg<sub>2</sub>Si phases because the precipitate curvature in the local composition of the interface between particle and matrix is taken into account. Therefore, the model allows the precipitation process to be simulated even for complex artificial ageing treatments by considering a dilute quasi-binary Al-Mg<sub>2</sub>Si alloy and a simplified unique precipitate size distribution in which all of the new Mg<sub>2</sub>Si precipitates nucleated are incorporated.

## 5.2 Mathematical modeling

The critical radius  $r^*$  of the particles nucleated in every new step is calculated using the Gibbs–Thomson equation in the singular case in which the concentration gradient at the interface between matrix and precipitates is exactly zero ( $r = r^*$ ) [40], according to:

$$r^* = \frac{2\sigma V_a}{kT \ln \frac{c}{c_\infty^\alpha}} . \quad (5)$$

In Eq. (5),  $\sigma$  is the interfacial energy between the aluminum matrix and the Mg<sub>2</sub>Si particles,  $V_a$  is the atomic volume of magnesium,  $k$  is the Boltzmann constant,  $T$  is the ageing temperature,  $c$  is the instantaneous concentration of magnesium in the matrix, and  $c_\infty^\alpha$  is the concentration of magnesium in the matrix in equilibrium with precipitates Mg<sub>2</sub>Si assuming a planar interface.

From the calculated value of the critical radius  $r^*$ , the activation energy  $G^*$  required for the nucleation of new particles of the calculated size is computed according to [20]:

$$G^* = \frac{4}{3}\pi r^{*2} \cdot \sigma . \quad (6)$$

The activation energy  $G^*$  is added to the activation energy for the diffusion of magnesium in aluminum  $Q$ . Neglecting the incubation period, the steady-state nucleation rate  $J$  during artificial ageing is calculated by [44]:

$$J = N_0 \cdot \frac{kT}{h} \cdot \exp\left(-\frac{G^* + Q}{kT}\right) , \quad (7)$$

where  $N_0$  is the number density of nucleation sites, and  $h$  is the Planck constant. In the case of homogeneous nucleation, the parameter  $N_0$  is computed as the number of magnesium atoms per unit volume.

It can be inferred that the driving force for the nucleation process is the instantaneous concentration of magnesium  $c$  in the matrix. This stage of the precipitation process is consequently exhausted when the concentration of magnesium  $c$  is close to the equilibrium concentration  $c_\infty^\alpha$  at the corresponding temperature.

Similarly, for the nucleation phenomenon, the growth rate of the previously created Mg<sub>2</sub>Si strengthening precipitates is calculated by assuming a diffusion controlled process in the case of spherical precipitates according to Eq. (8) [45]. This expression can be positive or negative, which requires the consideration of both growth and dissolution rates. At each calculation step, the precipitates that are larger than the calculated critical radius  $r^*$



will grow and the smaller particles will shrink, corresponding to [39, 40]:

$$\frac{dr}{dt} = \frac{D}{r} \cdot \frac{c - c_r^\alpha}{c^{\alpha'} - c_r^\alpha}, \quad (8)$$

where  $D$  is the diffusion coefficient of magnesium in aluminum at a defined temperature,  $r$  is the discrete precipitate radius,  $c_r^\alpha$  is the concentration of magnesium in the matrix at the interface and  $c^{\alpha'}$  is the concentration of magnesium in the  $\text{Mg}_2\text{Si}$  precipitate. The parameter  $c_r^\alpha$  is calculated for every particle radius from the generalized Gibbs–Thompson equation:

$$c_r^\alpha = c_\infty^\alpha \cdot \exp\left(\frac{2\sigma V_a}{kT} \cdot \frac{1}{r}\right). \quad (9)$$

Finally, the dependence of the diffusion parameter  $D$  on the ageing temperature  $T$  is expressed according to the following Arrhenius–type equation:

$$D = D_0 \cdot \exp\left(-\frac{Q}{RT}\right). \quad (10)$$

According to Eq. (5), the critical size calculated at every step  $r^*$  depends on the equilibrium magnesium content at the interface between the precipitate and the matrix, given by the phase diagram [46]. It is clear that the thermal stability of the precipitates depends on the changes in temperature during the ageing process. Thus, in the case of a non–isothermal ageing process, an increase in temperature leads to the dissolution of existing precipitates. Additionally, a decrease in the ageing temperature induces successive nucleation, growth, and coarsening of the precipitates. However, this phenomenon does not occur in these conditions because a single step ageing process at constant temperature was used.

Strengthening  $\text{Mg}_2\text{Si}$  nanometer–scale particles are supposed to cover the radius size range between 1 Å at nucleation and 1 μm for the long over–aged precipitates at long ageing times [1]. This radius size range has been divided into three hundred different classes using a logarithmic scale, thereby achieving a highly accurate tracking of the small precipitates, similar to previous implementations of Robson’s model [3, 19]. At every time step, the nucleated precipitates calculated using Eq. (7) are placed in the corresponding size class according to the present critical radius  $r^*$  calculated from Eq. (5). Additionally, based on the growth/coarsening and their opposite processes for the previously nucleated particles, the increase/decrease of the size limits for all of the classes is calculated using Eq. (8). The initial distribution at the beginning of the increment is then reorganized. Once the new nucleated particles are placed and the existing ones have been redistributed, the remaining concentration of magnesium in the matrix  $c$  is computed. A schematic view of the implemented procedure for Robson’s model is given in Fig. 9.



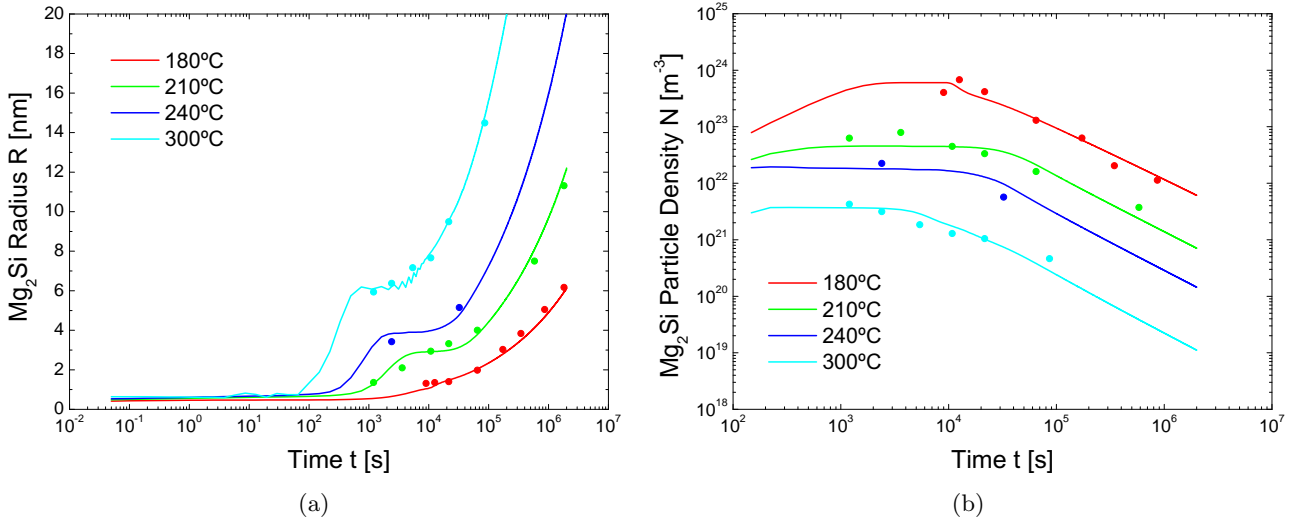


Figure 10: Results of SANS measurements for the evolution of  $\text{Mg}_2\text{Si}$  strengthening particles and predictions using Robson’s model for four different ageing temperatures 180°C, 210°C, 240°C, and 300°C referring to: (a) precipitate average size (note that the x-axis is plotted on a log-scale); (b) particle density (note that both axes are plotted on a log-scale). Note that straight lines correspond to predictions, while symbols represent experimental data.

at higher temperatures. This singular feature of the curve at 180°C is the consequence of a significant overlap between nucleation and coarsening, while for the other temperatures, the transition from nucleation to coarsening is more distinct. This phenomenon results from a low interfacial energy  $\sigma$  parameter that dominates in the class range of the majority of the strengthening precipitate distributions.

The double log-scale graph in Fig. 11 shows that the slope of all of the curves for the long ageing times (where coarsening of the strengthening distribution is taking place) is approximately 1/3, indicating that particle coarsening by Ostwald ripening dominates [47], confirming that large precipitates coarsen in the overageing process at the expense of the small ones when the matrix is depleted of the limiting solute. Thus, the overall energy in the material is reduced by a diffusion-driven process. This observation agrees well with the Lifshitz–Slyozov–Wagner (LSW) theory, which uses a kinetic equation appropriate for an infinitely dilute array of spherical particles in a stress-free matrix that predicts that the average particle radius will increase as a power function of the ageing time raised to 1/3 [48].

The governing driving force for coarsening is the difference between the matrix concentration in the vicinity of small and large particles, which leads to diffusion flux of the solute atoms from smaller particles to larger ones [49]. This phenomenon is caused by the Gibbs–Thomson effect included in the precipitation model, which changes the phase equilibrium between nanometer-scale precipitates and the matrix and also increases the solute concentration around small particles due to the interfacial curvature.

The input data used for Robson’s model are listed in Table 4 with assumptions to simplify the problem. The Al–Si–Mg alloy system has been extensively examined, and the literature provides numerous examples from which the required parameters for the model can be extracted. It should be noted that the interfacial energy  $\sigma$  between the aluminum matrix and the  $\text{Mg}_2\text{Si}$  strengthening precipitates has been approximated by a sigmoidal

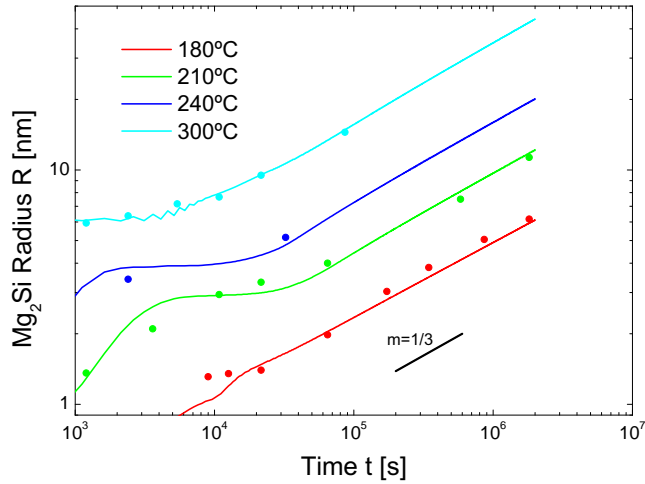


Figure 11: Evolution of the  $\text{Mg}_2\text{Si}$  strengthening precipitate size for four different ageing temperatures for long ageing times during which the Ostwald ripening coarsening process happens. Straight lines correspond to predictions, while symbols correspond to experimental data. Note that both axes are plotted on a log-scale.

function of the precipitate size as described in Sect. 5.3.1, while for previous reports it has been assumed to be constant [21]. The interfacial energy  $\sigma$  (that is, the sigmoidal function implemented for this parameter) and the diffusivity constant  $D_0$  are the only two parameters fitted to SANS measurement data.

Parameter	Value	Units
$N_0$	$2.0239 \cdot 10^{24}$	part/m <sup>3</sup>
$Q$	133000	J/mol
$V_a$	$3.95 \cdot 10^{-5}$	m <sup>3</sup> /mol
$\sigma$	Sigmoidal function; Eq. (11)	J/mol
$c_{ini}$	0.41	wt %
$c^{\alpha'}$	63.4	wt %
$D_0$	$9.81 \cdot 10^{-5}$	m <sup>2</sup> /s
$\rho_{\text{AlSi}_{10}\text{Mg}(\text{Cu})}$	2753	kg/m <sup>3</sup>
$\rho_{\text{Mg}_2\text{Si}}$	1900	kg/m <sup>3</sup>

Table 4: Input data for fitting the Robson precipitation model for the  $\text{AlSi}_{10}\text{Mg}(\text{Cu})$  alloy. The sigmoidal function parameter and the diffusivity constant  $D_0$  have been fitted to the SANS measurement data.

In Table 4, the parameter  $c_{ini}$  is the concentration of magnesium in weight percentage present in the aluminum matrix at the supersaturated solid solution condition at the beginning of the precipitation process. This parameter coincides with the total concentration of magnesium in the alloy based on the simplifications made for modeling the process. The total concentration of magnesium in the material and the density of the  $\text{AlSi}_{10}\text{Mg}(\text{Cu})$  alloy  $\rho_{\text{AlSi}_{10}\text{Mg}(\text{Cu})}$  have been measured at the laboratories of the Volkswagen AG. Note that the measured magnesium concentration falls within the established limits for the alloy composition given in Table 1.



### 5.3.1 Interfacial energy $\sigma$

The interfacial energy  $\sigma$  between formed precipitates and matrix for other aluminum alloys is not a constant, and some approximations have been successfully implemented in which this model parameter is directly dependent on the considered precipitates' size [22]. However, the applications of these approximations are reduced to very narrow and specific ranges of precipitate size and do not extend to the whole precipitation strengthening treatment. Additionally, the coarsening of the strengthening precipitates is usually accompanied by a loss of coherency in the matrix, so that an increase in the interfacial energy is also expected [50]. Extending this effect to the whole precipitation process implies that the transition from the very fine and coherent  $\beta''$ -Mg<sub>2</sub>Si precipitates to the coarse and incoherent  $\beta$ -Mg<sub>2</sub>Si strengthening particles should be accompanied by an increase in the interfacial energy  $\sigma$ . In fact, the assumption that the interfacial energy  $\sigma$  during the nucleation and the coarsening processes stays constant has been noted as a weakness in classical nucleation theory [22]. In this work, the interfacial energy  $\sigma$  needs to be a function of the precipitate size.

A transition from the lower value of  $\sigma$ , corresponding to the small nuclei, to the higher value for the coarse precipitates is included and agrees with previous investigations [22]. In this way, nucleation and coarsening can be correctly predicted. The quantitative value of this parameter assumes a characteristic rod shape for these particles and the interface created with the matrix. Under these assumptions, the selected function used to describe the dependence of the interfacial energy on the precipitate size  $\sigma(r)$  is a generalized logistic function for modeling the growth of a magnitude [51]. The resulting interfacial energy function  $\sigma(r)$  is shown in Fig. 12(a) and is given by the expression:

$$\sigma(r) = A + \frac{K - A}{(1 + Q \cdot \exp(-B \cdot (r - M)))^{1/\nu}}, \quad (11)$$

where  $A = 0.020$  is the lower asymptote,  $K = 0.033$  is the upper asymptote,  $Q = 45.0$  is fitted according to the value of  $Y(0)$ ,  $B = 8.75 \cdot 10^9$  is the growth rate, and  $M = 8.0 \cdot 10^{-11}$  is the radius of maximum growth if  $Q = \nu$ . Parameter  $\nu = 0.59$  denotes the place where maximum growth occurs.

Note that, in the assumptions made for Robson's model, the Mg<sub>2</sub>Si strengthening precipitates are assumed to be spherical, which does not agree with the conclusions drawn from TEM images in Sect. 4.2. In fact, the intensity scattering curves obtained from the SANS measurements were analyzed for cylindrical particles. As a consequence, the size of the particles in Fig. 10 refers to the radius of the rod-shaped precipitates' cross-section.

### 5.3.2 Diffusivity constant $D_0$

The sensitivity of Robson's model to the diffusivity constant  $D_0$  is also remarkable. This diffusion parameter has been identified from the discrete values of the diffusion constant  $D$  obtained from the individual fitting procedures done at every ageing temperature, assuming the activation energy for the diffusion of magnesium in aluminum  $Q$  given in Table 4. The parameter fitting results are shown in Fig. 12(b). The value for the

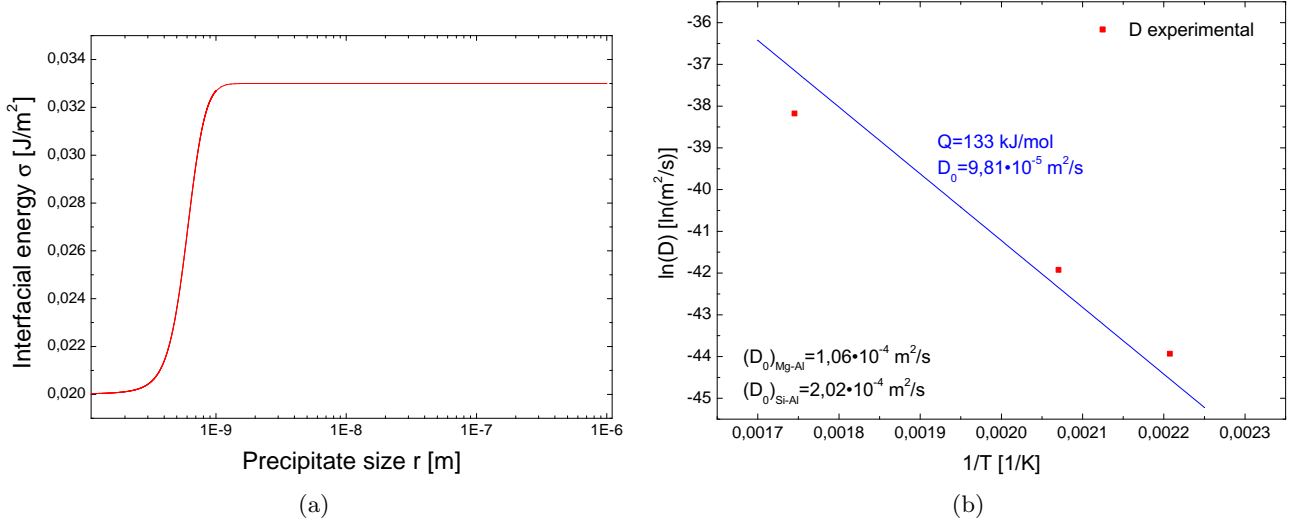


Figure 12: Resulting fit of the material parameters of Robson’s model: (a) interfacial energy as a function of the characteristic size of every precipitate class  $\sigma(r)$  (note that x-axis is plotted on log-scale); (b) diffusivity constant  $D_0$ .

diffusivity parameter  $D_0$  ( $D_0 = 9.81 \cdot 10^{-5} \text{ m}^2/\text{s}$ ) is similar to values reported in the literature from experimental measurements ( $D_0 = 10.6 \cdot 10^{-5} \text{ m}^2/\text{s}$ ) [42].

### 5.3.3 Solubility

The input data for the equilibrium concentration of magnesium  $c_\infty^\alpha$  are calculated using the projection of the magnesium concentration as the standard reference for the solid solubility. Table 5 gives the calculated equilibrium magnesium concentration  $c_\infty^\alpha$  at each temperature from the quasi-binary Al–Mg<sub>2</sub>Si alloy system phase diagram [46].

Temperature	$c_\infty^\alpha$
K	wt %
453	0.1358
473	0.1585
513	0.1885
573	0.2328
673	0.3549

Table 5: The equilibrium concentration of magnesium  $c_\infty^\alpha$  in the matrix in equilibrium with Mg<sub>2</sub>Si precipitates, assuming a planar interface for the temperature range of interest, as calculated from the quasi-binary Al–Mg<sub>2</sub>Si phase diagram [46].

## 5.4 Outlook

The Robson model has been adapted to simulate the precipitation process of the  $\text{Mg}_2\text{Si}$  nanometer-scale strengthening particles of the  $\text{AlSi}_{10}\text{Mg}(\text{Cu})$  alloy system. Due to its physical fundamentals and the unique set of model parameters obtained, the model can be used to design the appropriate artificial ageing heat treatment processes by considering not only different ageing stages but also different heating and cooling rates. Thus, the model can be interpreted as an ideal tool for finding a compromise between the duration and temperature of the ageing process according to the required distributions of  $\text{Mg}_2\text{Si}$  strengthening precipitates to satisfy the mechanical demands of the application.

The adaptation of a constitutive mechanical model of simulation based on metal physics can be coupled with the predictions of the  $\text{Mg}_2\text{Si}$  strengthening precipitate distributions provided by Robson's model. The macroscopic mechanical response of the material to a loading history can then be easily simulated to establish a relationship between the results of the microstructure predictions and the parameters of the mechanical model. Moreover, the *in-situ* evolution of the mechanical properties in long duration, high temperature tests as a consequence of the changing size distribution of precipitates can also be predicted by coupling the models.

## 6 Conclusions

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This work presents a comprehensive analysis of the microstructure of the age-hardenable  $\text{AlSi}_{10}\text{Mg}(\text{Cu})$  alloy and its evolution in the artificial ageing stage of a  $T_6$ - $T_7$  heat treatment process.

The nanometer-scale  $\text{Mg}_2\text{Si}$  strengthening precipitates formed in the artificial ageing state of the  $T_6$  heat treatment process were found to be responsible for the observed changes in the mechanical behavior of the alloy. SANS was successfully used to characterize the precipitate distributions formed, providing good results with meaningful statistics. Four different ageing temperatures and ageing times were considered. Higher ageing temperatures produced coarser size distributions of  $\text{Mg}_2\text{Si}$  strengthening precipitates with a lower density. These investigations focused mainly on the intermediate growth and final coarsening processes.

TEM/HTEM was performed to verify the analysis of the  $\text{Mg}_2\text{Si}$  strengthening precipitates. The metastable  $\beta'$  and the thermodynamic equilibrium  $\beta$  phases of these precipitates had rod-like shapes. The formation of new precipitate nuclei takes place homogeneously in the aluminum matrix.

The Robson precipitation model was successfully fitted using the radii analysis and particle density obtained for the  $\text{Mg}_2\text{Si}$  strengthening precipitate distributions based on the SANS experimental data and the TEM analysis. The model was adapted to the present alloy system by considering only one precipitate size distribution for the different metastable and equilibrium  $\text{Mg}_2\text{Si}$  phases.

A unique set of parameters was used to predict the whole precipitation process and the whole range of temperatures using Robson's model. The interfacial energy parameter  $\sigma$  and the diffusivity constant  $D_0$  were the two parameters fitted in the model. The model showed a high sensitivity to both values. The interfacial energy

parameter  $\sigma$  was explicitly dependent on the precipitate size. A sigmoidal approximation function was used to establish the dependency of the size of the precipitates  $\sigma(r)$ . The obtained value for  $D_0$  was similar to experimental values reported in the literature.

Therefore, the size distributions of  $\text{Mg}_2\text{Si}$  strengthening precipitates were quantitatively characterized experimentally by testing different temperatures and durations of the artificial ageing process. The SANS data and TEM images allowed the successful adaptation of Robson's precipitation model. The adapted model correctly predicted the nucleation, growth, and coarsening stages of the  $\text{Mg}_2\text{Si}$  strengthening precipitate distributions with a simple procedure using a unique set of parameters. Moreover, the model successfully predicted the results for a wide range of ageing temperatures and different precipitation stages. These features make the model optimal for designing and optimizing complex non-isothermal artificial ageing processes for real precipitation heat treatment processes in industrial applications.

## 7 Acknowledgments

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