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Design of a nitrogen-implanted titanium-based superelastic alloy with optimized properties for biomedical applications

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Abstract

In this study, a superelastic Ti-based biomedical alloy (Ti-25Ta-25Nb) was treated in surface by implantation of nitrogen ions for the first time. The N-implanted surface was characterized by X-ray diffraction, X-ray photoelectron spectroscopy, secondary ion mass spectroscopy, and the superficial mechanical properties were evaluated by nano-indentation and by ball-on-disk tribological tests. To investigate the biocompatibility, the corrosion resistance of the N-implanted Ti-25Ta-25Nb alloy was evaluated in Ringer solution and *in-vitro* cytocompatibility tests on human fetal osteoblasts were carried out. After implantation, surface analysis methods revealed the formation of titanium nitride on the substrate surface. Consequently, an increase in superficial hardness and a significant reduction of friction coefficient were observed compared to the non-implanted sample. Also, a better corrosion resistance and a significant decrease in ion release rates were observed. *In vitro* tests performed on human fetal osteoblasts indicated that the cytocompatibility of the N-implanted Ti-25Ta-25Nb alloy was superior to that of the corresponding non treated sample. Thus, this new functional N-implanted titanium-based superelastic alloy presents the optimized properties that are required for medical applications: high superficial mechanical properties, high corrosion resistance and excellent cytocompatibility.

Keywords: titanium alloy, ion implantation, surface characterization, biocompatibility

1. Introduction

During the last years, intense research was performed in order to develop highly biocompatible titanium-based alloys for medical applications [1-8]. Metastable beta titanium alloys were shown to be excellent candidates, due to the possibility to elaborate them only with fully biocompatible elements such as Ta, Nb, Mo, Zr [1, 9]. A metastable beta titanium alloy can be obtained if a sufficient amount of β -stabilizing elements is added to retain at room temperature the β microstructure (body centered cubic) after quenching from the beta domain. It is shown in literature that the microstructure and the mechanical properties of metastable β titanium alloys are very sensitive to composition changes and thermal treatments [2, 10]. In a given composition domain, the metastable β microstructure may be mechanically unstable. Then, the quenched β microstructure is decomposed into a stress-induced α'' martensite (C-centered orthorhombic) when an external stress is applied. This phenomenon is generally highlighted by a characteristic tensile stress-strain curve showing a nonlinear elastic domain and a stress plateau associated with a double yielding behavior. As the stress-induced martensitic transformation is reversible, a high recoverable strain can be obtained, leading to a superelastic effect [1, 2]. Thus, these non-allergen Ni-free alloys can be envisaged to replace the superelastic NiTi-based materials currently used for the fabrication of functional biomedical devices such as arches, wires and other springs used in orthodontics, cardiovascular or pulmonary stents, wires and catheters for the cardiovascular surgery, staples and prostheses of articulations for the orthopedic surgery [1, 7]... On the other hand, the superelastic effect is accompanied by a very significant reduction of the elastic modulus, which is very beneficial for the use of these alloys like prostheses or implants in osseous sites (articulations prostheses, dental implants...). Indeed, a low elastic modulus reduces the stress-shielding effect that is caused by a difference in the elastic modulus of the bone substitute and the natural human bone that affects long-term performances in service [11, 12].

Nevertheless, the main weaknesses of titanium alloys are their poor friction and wear resistance. Specifically, in the case of head prostheses, their surface has to be treated or reinforced in order to better resist friction and to reduce wear particles release [13-17]. On the other hand, metallic materials implanted in human body, must have a high corrosion resistance to avoid the release of ions and corrosion products in the surrounding tissues that can ultimately affect their biocompatibility [18-23]. A well-known and reliable surface

treatment process that can enhance both surface mechanical properties and corrosion resistance of titanium alloys can be ion implantation [24-26]. As concerns nitrogen implantation, besides the increased surface hardness and wear resistance, this surface modification procedure was associated with improved corrosion resistance [26]. In addition, several *in vitro* and *in vivo* studies demonstrated the superior biological performance of nitrogen-implanted titanium based materials [27-29]. It is worth noting that most of the studies found in literature dealing with the ion implantation of biomedical titanium alloys concern the Ti-6Al-4V and CP-Ti alloys commonly used in medicine and, to our knowledge, no study mentions ion implantation on metastable β Ti-based superelastic alloys.

The aim of the study is then to investigate the superficial characteristic changes of a superelastic Ti-based alloy after implantation of nitrogen ions in surface. The Ti-25Ta-25Nb alloy was chosen because a complete investigation of its microstructure and its superelastic behaviour was carried out in previous studies [30, 31]. In this work, the microstructural changes and the mechanical property modifications in surface were investigated by Grazing Incidence X-ray Diffraction (GIXRD), X-ray Photoelectron Spectroscopy (XPS), Secondary Ion Mass Spectroscopy (SIMS), nano-indentation and ball-on-disk tribological tests after nitrogen implantation. As the N-implanted Ti-25Ta-25Nb superelastic alloy is on great interest for biomedical applications, its corrosion behaviour in simulated body fluid and its biocompatibility using *in vitro* tests performed on human fetal osteoblasts were also evaluated in the present study.

2. Materials and methods

2.1. Alloy preparation and implantation method

The Ti-25Ta-25Nb alloy composition (25 mass. % of Ta, 25 mass. % of Nb and the balance mass.% of Ti) was synthesized by cold crucible levitation melting technique (CCLM) under high vacuum. For the synthesis, pure raw metals were used. Ingots obtained from melting were next homogenized at 1223K for 72ks under high vacuum, followed by water quenching. Ingots were then cold rolled at room temperature until a 90% of reduction in thickness was achieved. From this cold-rolled state, samples were cut in square pieces (20mm x 20mm), then solution-treated in the β -phase domain at 1123K for 1.8ks under vacuum in order to restore a fully recrystallized microstructure and finally water quenched to retain the

metastable β microstructure at room temperature. As previously mentioned in the introduction, an investigation of the microstructure and the tensile properties of the metastable beta Ti-25Ta-25Nb alloy synthesised by this process was carried out in a previous work [31]. In Fig. 1, it is reported the typical tensile curve exhibiting a double yielding behavior characteristic of the superelastic property that possesses such alloy. From this curve a low incipient Young's modulus measured at 55 GPa was found, which is very low compared to that obtained from classical titanium alloys currently used in medicine (110 GPa) and closer to the one that corresponds to the cortical bone (20-30 GPa).

Before ion implantation, square samples were first mechanically polished on silicon carbide abrasive papers (up to 4000 grit) followed by a final polishing step with a colloidal silica suspension (particles size: 50 nm) and then ultrasonically cleaned in acetone, thoroughly washed with ethanol and dried in air. In this work, an original ion implantation technique developed and patented by Quertech Ingénierie, France was used. This process (trademark Hardion+) is based on the use of an Electron Cyclotron Resonance (ECR) ion source to produce a multienergetic ion beam from multicharged ions (N^+ , N^{2+} , N^{3+} ...). The N^{i+} ions have energy and a depth of implantation i time higher than the N^+ ion. Consequently, these multicharged ions are implanted simultaneously at various depths giving a profile of concentration relatively flat and deeper by comparison with the conventional monoenergetic ion implantation technique [32]. The great advantage of this technology is that the ion implantation was carried out in a micro-implanter (30 cm for 10 kg) using permanent magnets mainly composed of the ECR ion source coupled to a vacuum chamber. Currently, the Hardion+ technology can treat only flat surfaces but the use of such micro-implanter technology must make it possible to overcome the geometry limitation imposed by traditional linear ion implantation.

Using this technology, nitrogen ions (N^+ to N^{5+}) were implanted with the same dose (5.10^{17} ions.cm⁻²) in all samples with an average energy of 100 keV (ion beam intensity of 4 mA). During the implantation, the titanium alloy sample is placed on a water cooled holder which moves under the ion beam. Taking into account the speed used (40 mm.s⁻¹), the surface temperature was always measured to be lower than 50°C.

2.2 Surface characterization methods

Phase identification of the N-implanted surface was carried by Grazing Incidence X-ray Diffraction technique (GIXRD) using a Bruker D8 Discover diffractometer working at 40 kV with the Cu K α radiation in a glancing geometry (incidence angle fixed at 1°). On the other hand, an analytical study using X-ray Induced Photoelectron Spectroscopy (XPS) has been carried out on a Kratos Axis Ultra DLD working with a 15 kV X-ray gun and monochromatic AlK α radiation. The analysed area size was 700 x 300 microns and the pass energy was set to 20 eV during region scans (energy resolution of 0.1 eV). Additionally, argon ions at 4 keV have been used to etch for 1800 s (2 mm²) in order to obtain a depth profile. By deconvolution of the binding energies on Ti 2p, N 1s, Nb 3d and Ta 4d core levels, the present chemical bonds have been studied in detail.

Secondary ion mass spectroscopy (SIMS) experiments were carried out on a Hiden analytical system including the IG20 ion gun using Ar⁺ ions at 5 keV and a MAXIM-Quadrupole mass detector. The focused beam has a diameter of 100 μ m at about 500 nA beam current. To avoid edge effects at the crater the beam was rasterised across the sample (0.65 x 0.45 mm). Signals were just measured in the inner gate of approximately 0.15 x 0.1 mm.

Superficial mechanical properties were studied by nano-indentation with a CSM NHT instrument equipped with a diamond Berkovich-type indenter. For the measurements, the continuous stiffness mode was used. Thus, indentations were made perpendicularly to the sample surface using the oscillating mode with 2 mN increasing step that allowed hardness-depth profile plot. 10 measurements were made to plot the average hardness profile, before and after ion implantation.

Ball-on-disk tribological tests were performed with a CSM tribometer to evaluate friction coefficient and wear resistance of treated surfaces. Friction coefficient was monitored during sample surface sliding against an alumina ball (6 mm diameter) in ambient atmosphere. The applied load and sliding velocity were 0.25 N and 0.01 m.s⁻¹, respectively.

2.3. Corrosion behaviour testing

The surfaces of non-treated and N-implanted Ti-25Ta-25Nb specimens with an area of 1 cm² were degreased in acetone, washed with bi-distilled water and dried in air. Their electrochemical behaviour in Ringer solutions of different pH values (2.33; 6.98; 9.1) and Ringer-Brown solution (pH = 7.1) was studied. Solutions were selected to simulate the severe functional conditions of an implant (different pH values) [33-35]:

- Ringer solutions of pH = 2.33 (obtained by HCl addition), pH = 6.98 (normal), pH = 9.1 (obtained by KOH addition) had the composition (g/L): NaCl – 6.8; KCl – 0.4; CaCl₂ – 0.2; MgSO₄·7H₂O – 0.2048; NaH₂PO₄·H₂O – 0.1438; NaHCO₃ – 1.1; glucose – 1;
- Ringer-Brown solution of pH = 7.1 had the composition (g/L): NaCl – 6.0; KCl – 0.4; CaCl₂ – 0.2; sodium lactate – 3.05.

The corrosion behaviour of the N-implanted and non-treated Ti-25Ta-25Nb alloy was studied from the cyclic polarization curves, linear polarization (Tafel representation) and monitoring of the open circuit potentials and corresponding open circuit potential gradients. The cyclic potentiodynamic polarization was applied starting from -0.3 V to +0.6 V (vs. SCE) with a scan rate of 1 mV/sec. Voltalab 80 equipment with its VoltaMaster 4 program were used. From voltammograms, the main electrochemical parameters were determined: E_{corr} - corrosion potential, like zero current potential, E_p – passivation potential at which the current density is constant; $|E_{\text{corr}} - E_p|$ difference represents the tendency to passivation (low values characterize a good, easy passivation); ΔE_p – passive potential range of the constant current; i_p – passive current density.

The linear polarization method was used to obtain the polarization resistance (R_p), the corrosion current densities (i_{corr}), corrosion (V_{corr}) and the ion release rates. The linear polarization was performed for ± 150 mV around the open circuit potential at a scan rate of 1 mV/s using the same Voltalab 80 equipment. VoltaMaster 4 program directly supplies the values of R_p , i_{corr} and V_{corr} . Ion release rates were calculated.

Open circuit potentials E_{oc} were monitored (for 2000 immersion hours in physiological solutions) using Hewlett-Packard multi-meter. The corresponding open circuit potential gradients due to the pH non-uniformity $\Delta E_{\text{oc}}(\text{pH})$ and composition $\Delta E_{\text{oc}}(\text{c})$ of physiological solutions were calculated as:

$$\Delta E_{\text{oc}1}(\text{pH}) = E_{\text{oc}}^{\text{pH}=2.33} - E_{\text{oc}}^{\text{pH}=6.98} \quad (1)$$

$$\Delta E_{\text{oc}2}(\text{pH}) = E_{\text{oc}}^{\text{pH}=2.33} - E_{\text{oc}}^{\text{pH}=9.1} \quad (2)$$

$$\Delta E_{\text{oc}3}(\text{pH}) = E_{\text{oc}}^{\text{pH}=6.98} - E_{\text{oc}}^{\text{pH}=9.1} \quad (3)$$

$$\Delta E_{\text{oc}4}(\text{c}) = E_{\text{oc}}^{\text{Ringer pH}=6.98} - E_{\text{oc}}^{\text{Ringer-Brown pH}=7.1} \quad (4)$$

2.4. In vitro assessment of osteoblast behaviour

2.4.1. Cell culture

The cell culture study was conducted with the American Type Culture Collection CRL-11372™ (hFOB 1.19) human-foetal-osteoblasts, a human pre-osteoblastic cell line. This cell line is transfected with a gene that codes for the temperature sensitive mutant (tsA58) of the SV 40 T-antigen, together with a gene coding for gentamycin resistance [36]. For performing any kind of biological experiments, three samples of each type (N-implanted and non-treated Ti-25Ta-25Nb) were sterilized at 180°C for 30 min and placed individually into the wells of the 6-well (for test samples of 25x25x1 mm in the MTT and LDH assays) and 12-well (for test samples of 10x10x1 mm in the microscopy study) Multiwell culture plates. Surfaces were seeded with cells at an initial density of 1×10^4 cells/cm² and incubated at a temperature of 34°C in a humidified incubator containing 5% CO₂. The hFOB 1.19 cell lines were maintained in Dulbecco's Modified Eagle's Medium/ Ham's Nutrient Mixture F-12 [DMEM/F12-1:1 (w/w) mix] containing 10% (v/v) fetal bovine serum (FBS) and 300 mg/ml neomycin (G418) for specific times. The medium was changed every third day up to 7 days excepting the cytotoxicity assay when the culture medium was replaced in the second and fourth day of culture.

2.4.2. Cytotoxicity assay

The effect of non-treated and N-implanted Ti-25Ta-25Nb samples on cell viability was assessed using lactate dehydrogenase (LDH) release assay. This study was performed by using a cytotoxicity detection kit (TOX-7, Sigma-Aldrich Co.) according to the manufacturer's protocol. High OD_{490nm} values are indicative of a reduction in the cell viability

2.4.3. Fluorescence staining of the actin and fibronectin

Cell attachment, spreading and morphology were assessed by fluorescent staining of actin cytoskeleton at 1 h, 4 h and 24 h post seeding. Briefly, at the end of the incubation period, cells were washed with phosphate buffer saline (PBS) three times and fixed with 4% paraformaldehyde for 20 min. Fixed cells were rinsed with PBS and permeabilized for 1 h with 0.1% Triton X-100/2% BSA followed by three 10-min PBS washes. Afterwards, the permeabilized cells were incubated with phalloidin-TRITC (tetramethylrhodamine B isothiocyanate) (20 µg/mL; Sigma-Aldrich Co.) for 1 h at 37°C to label actin filaments. The nuclei were stained for 15 min with DAPI (4',6-diamidino-2-phenylindole) at a concentration of 2 µg/mL in PBS. For immunolocalization of the extracellular matrix (ECM) protein, fibronectin (FN), fixed and permeabilized samples were processed as previously shown [37].

In both studies, images from representative fields of each sample were viewed with an inverted fluorescent microscope Olympus IX71 and captured by means of Cell F image acquiring system.

2.4.4. Cell proliferation assay

To evaluate cell proliferation potential, the cells on days 2, 4 and 7, after seeding on non-treated and N-implanted Ti-25Ta-25Nb surfaces were incubated with 0.5 mg/mL MTT [3-(4, 5-dimethyl thiazol-2-yl) 2, 5-diphenyltetrazolium bromide) solution according to tetrazolium-based colorimetric assay described by Mosman [38]. The mitochondrial enzyme succinate-dehydrogenase within viable cells is able to reduce the yellow dye MTT to a purple colored product (formazan) which is soluble in dimethyl sulfoxide (DMSO). The amount of formazan produced, whose absorbance was recorded at 550 nm with the Thermo Scientific Appliskan microplate reader, was proportional to the number of viable, metabolically active cells present.

2.4.5. Statistical analysis

For the statistical analysis of the LDH and MTT assays, one-way ANOVA with Bonferroni's multiple comparison tests was performed. All values are expressed as mean value \pm standard deviation of three independent experiments and differences at $p\leq 0.05$ were considered statistically significant.

3. Results and discussion

3.1. Surface characterization

GIXRD patterns from the Ti-25Ta-25Nb alloy are presented in Fig. 2. For the non-treated surface, the diffracted peaks are related to the β phase (body centered cubic) which constitutes the microstructure of the bulk alloy. After nitrogen implantation, additional peaks are observed and can be clearly attributed to the face centred cubic nitride phase TiN formation.

To determine the chemical states, high resolved XPS spectra have been measured at the N 1s, Ti 2p, Nb 3d and Ta 4d core levels as shown in Fig. 3. The binding energy of the N 1s state was measured to be between 397.0 eV and 397.3 eV according to the convolution of

three signals from TiN (literature 397.2 eV [39]), TaN (397.0 eV [40]) and NbN (397.3 eV [41]). All nitrogen atoms are in chemical bond to titanium, to tantalum and niobium, excepting the surface contaminations with hydrocarbon. The Ti 2p_{3/2} signal at 455.2 eV is contributed to TiN [39] (metallic bulk core level 454.1 [42, 43]). Ti-O bonds around 459 eV [44] are predominantly very close to the surface due to the contamination effects. NbN signals occurred at 203.6 (literature 203.8 [45]) whereas the metallic peak was measured at 202.2 eV (literature 202.4 eV [45]). A weak oxide peak at 207.2 eV (207.4 eV [45]) on the very surface occurred as well (artificial argon signals around 242 eV are related to the etching). In relation to Niob the Tantalum pentaoxide peak is much stronger at 229.9 eV (230.1 eV [46]). In depth the nitride formation looks very similar to Nb and Ti. The peaks occurred at 228.0 eV for the nitride and 226.3 eV (226.4 eV [47]) for the metallic state. Close to the surface contamination effects become visible for all studied excitation states. Beside classical hydrocarbon the affinity of oxide formation has been shown for all alloying elements. All peaks have been deconvoluted by means of Gauss-Lorentz GL (30) peak functions and Shirley backgrounds with respect to the orbit split of the 2p_{1/2} and 2p_{3/2} state having an area ratio of 1:2, respectively the 3d/4d_{3/2} and 3d/4d_{5/2} orbital split with an area ratio 2:3. Thus, Ti 2p_{1/2}, Nb 3d_{3/2} and Ta 4d_{3/2} have not been discussed separately.

SiMS depth profiles were shown in Fig. 4. They verify the observed elemental distribution by XPS. Especially, the TaN⁻, TiN⁻ and NbN⁻ ion distributions are very similar. The varying count rates are related to different sputter efficiencies. This behaviour can also be observed on the Nb⁻ and Ta⁻ profile. The shape of the depth profiles describes and follows the multi ion implantation process and at least in the arising sputter efficiency.

3.2 Nano-indentation and friction measurements

Mean hardness vs. tip penetration depth profiles of N-implanted and non-implanted samples are shown in Fig. 5. From the N-implanted sample, the nano-indentation profile begins at a high hardness value and then decreases until reaching the non affected zone in the core of the alloy. One can notice that the non-implanted sample profile also presents that trend to a lesser extent. This can be attributed to work hardening due to mechanical polishing. In Fig. 5, a hardness increasing in the first 600 nm due to nitrogen implantation is clearly evidenced. A maximum hardness value at about 8.5 GPa is then obtained after N-implantation, which corresponds to an increasing of almost twice the hardness value of the bulk alloy.

Fig. 6 presents typical evolutions of friction coefficients as function of the number of cycles for N-implanted and non-implanted samples during ball-on-disk tribological tests. It is clearly showed that the N-implanted sample possesses a reduced friction coefficient by comparison with the non-treated sample. After an accommodation period, the friction coefficient of the N-implanted sample is stabilized at about 0.3, which corresponds to a significant reduction of around a factor 1.7 by comparison with the non-implanted sample. In Fig. 6, it is also presented optical micrographs showing the wear tracks obtained after 200 cycles of friction. It can be clearly observed that the N-implanted sample has a narrow wear track compared to the non-treated one. The measurements indicate a wear track thickness reduction of about 85% under the experimental conditions used: 200 cycles under 0.25N load.

These increased hardness and wear resistance are clearly due to the nitride formation on surface, which was detected by GIXRD, XPS and SIMS. In addition, hardness profile measurements are in good agreement with the XPS profile. Indeed, both methods show that the affected depth is about 600 nm after nitrogen implantation.

3.3. Corrosion behaviour

From cyclic potentiodynamic curves (Fig. 7) it can be observed that both non-treated and N-implanted Ti-25Ta-25Nb alloys exhibited passive behaviour in all studied biofluids. The N-implanted alloy had a nobler electrochemical behaviour than the non-treated sample, proved by the more favourable electrochemical parameters (Table 1): more electropositive corrosion (E_{corr}) and passivation (E_p) potentials, lower tendencies to passivation $|E_{\text{corr}} - E_p|$, i.e. an easier passivation, and lower passive current densities i_p , showing a more resistant passive layer. These facts can be explained by the presence of the nitrated layer that acts as a barrier layer against the metallic ion transport through it, into aggressive solutions. The most favourable electrochemical parameters for N-implanted alloy were registered in neutral Ringer and Ringer-Brown solutions, the normal functional conditions of an implant; these parameters revealed a very good passive behaviour. In severe functional conditions, simulated by the acid (pH = 2.33) and alkaline (pH = 9.1) Ringer solutions, the main electrochemical parameters for N-implanted Ti-25Ta-25Nb presented better values than the non-treated alloy, indicating a good passive behaviour. Passive current densities i_p for N-implanted alloy represented about a half from the passive current densities of the non-treated alloy as result of the better protection, conferred by the deposited layer.

Corrosion current densities i_{corr} and corresponding corrosion V_{corr} and ion release rates (obtained from linear polarization measurements) of the N-implanted alloy (Table 2) are lower than those of the non-treated one and are placed in the „Perfect Stable” resistance class [30-33], proving the protective properties of the nitrated layer. The polarisation resistances R_p are higher for the N-implanted alloy than for non-treated sample, demonstrating a more protective, more resistant layer that reduced the ion release rate and corresponding alloy toxicity. The lowest corrosion and ion release rates were obtained in neutral Ringer and Ringer-Brown solutions, condition in which an implant “works” the most time from its “service life”. Taking into account that the N-implanted alloy presented better anticorrosive resistance, it results that the layer is very effective both in normal and severe functional conditions.

Monitoring of the open circuit potentials for 2000 immersion hours in the studied simulated biofluids (Fig. 8) showed that these potentials for the N-implanted alloy are more electropositive than those of non-treated alloy, attesting a more resistant layer; the open circuit potentials tended to nobler values in time and reached a constant level at about 1000 immersion hours, indicating the improvement of the protective state [48, 49]. In acid and alkaline Ringer solutions, the slight more electronegative open circuit potentials than in neutral Ringer solution were obtained due to the higher aggressivity of these solutions; nevertheless, their values are placed in the noble potential range (about -100 mV), revealing a passive resistant state.

Open circuit potential gradients (Table 3) had low values that cannot generate galvanic cells, because only differences of 0.6 – 0.7 V (in absolute value) [50, 51] can initiate and maintain this type of corrosion. It can be concluded that the galvanic corrosion cannot develop on the implant surface even in the severe case of a very large pH difference between acid (2.33) and alkaline (9.1) pH values.

3.4. *In vitro* biological response

Cell culture models are routinely used to gain insight into the bone-cell response to candidate orthopedic biomaterials *in vitro*. In order to evaluate the effect of nitrogen implantation by Hardion+ micro-implanter technology on the biocompatibility of the recently developed beta-type titanium alloy, Ti-25Ta-25Nb [52], we first studied the LDH release in the culture medium at different periods of time. This procedure gave us information about the cytotoxicity of the analyzed materials as represents an indicator of cell membrane integrity.

As shown in Fig. 9, hFOB 1.19 cells grown on non treated and N-implanted Ti-25Ta-25Nb samples proved to display low LDH release in culture medium nearly comparable to that of tissue culture plates (TCPS) (data not shown). These results indicate that nitrogen implantation has no direct effect on the release of LDH and does not exert cytotoxic effects. The lack of cytotoxicity of these materials is also due to their noble potential and passive behaviour in biofluids that we have proved in the corrosion experiments.

This finding is also supported by the morphological features of the osteoblasts noticed by fluorescence microscopy (Fig. 10) of the cells labeled with phalloidin-TRITC at 1 h, 4 h and 48 h after cell plating. It can be seen that, at 4 h post-seeding, on both surfaces the initial round cells rapidly assumed the elongated, spindle-shaped appearance typical of osteoblast-like cells plated onto TCPS. Cortical actin bundles predominantly appeared at the peripheries of cells, consistent with initiation, expansion, and completion of cell adhesion to the ECM proteins adsorbed on the material surfaces. The cells attained a well spread morphology characterized by significantly higher cell area, more evident on N-implanted surface, at 48 h after culture initiation. They displayed prominent actin stress fibers that appeared to be arranged parallel to the long axis of the cell. Taking into account that there is a close relationship between stress fiber formation and cell adhesion [53], this finding suggests that nitrided Ti-25Ta-25Nb surfaces favour bone cell adhesion. Furthermore, as compared to non-treated samples, a higher number of attached cells on N-implanted Ti-25Ta-25Nb were observed.

This last observation is in agreement with the MTT reduction assay which is very important since it provides information about the cell growth and the metabolic activity of the cells. It might be noticed (Fig. 11) that the levels of MTT conversion increase with time, showing an increasing number of viable cells as a result of the active cell proliferation over the 7 day incubation period. Further, the optical density values were higher in the case of N-implanted alloy with a more significant statistical difference at 4 and 7 days after cell plating. This finding indicates a better osteoblast response to N-implanted Ti-25Ta-25Nb.

The above results are consistent with studies performed by Chien et al. [54] on cell compatibility of TiN and TiAlN films prepared by a reactive radio frequency (RF) magnetron sputter deposition system on nickel-based alloys showing that cell viability and proliferation increased in comparison with the corresponding uncoated surfaces. Since surface topography was almost identical (data not shown) for both surfaces, the differences observed in the cell morphology and proliferation must be a consequence of the chemical composition or wettability or a synergistic effect of both. Thus, we noticed that besides changing surface

composition, Hardion+ implantation technique slightly increased the surface wettability of Ti-25Ta-25Nb. The water contact angles of non-treated and N-implanted alloy, measured with a CAM 100 equipment, ranged from $73.514^0 \pm 2.653^0$ to $66.270^0 \pm 1.442^0$. Consequently, N-implantation method makes the surfaces slightly more hydrophilic. The most general rule connecting material properties with cell-substrate compatibility is that anchorage-dependent mammalian cells prefer surfaces with moderate hydrophilicity (water contact angle around 60 degrees) [55]. This finding could explain the potential of N-implanted Ti-25Ta-25Nb to increase osteoblast proliferation in comparison with the non-treated sample.

Taking into consideration that cell-material interaction is a complex and dynamic bi-directional process mimicking to some extent the natural interactions of cells with ECM, namely the cells sense and accept information from cues in the ECM and, at their turn, they produce and frequently remodel their microenvironment [56, 57], the next objective of our study has been to evaluate synthesis, secretion and formation of the FN matrix by immunofluorescence. FN is an abundant component of the ECM being adsorbed to this, together with other adhesion molecules, from either serum within cell culture media, or body fluid within the *in vivo* environment. Cells tend to adhere and rearrange adsorbed ECM proteins on the material surface in a fibril-like pattern. Afterwards, the ECM undergoes proteolytic degradation, a mechanism for the removal of the excess ECM, and attached cells begin to synthesize and secrete their own insoluble matrix [58]. As shown in Fig. 12, at 2 days after cell seeding, the fluorescent staining for FN was detected mainly intracellularly and, to a lesser extent, surrounding the cells. At 7 days post-seeding, FN fibrils were present in the subcellular area, immediately surrounding the cell periphery and between the cells. These results demonstrate that hFOB 1.19 cells grown in contact with both non-treated and N-implanted Ti-25Ta-25Nb are able to synthesize, secrete and deposit an abundant FN matrix organized into fibrillar networks.

4. Conclusion

In this study, nitrogen implantation performed on a superelastic Ti-based alloy was carried out for the first time. It was shown that N implantation on the Ti-25Ta-25Nb alloy composition leads to chemical and mechanical modifications in surface.

The formation of titanium nitride was clearly identified by GIXRD, XPS and SiMS analyses on the N-implanted surface.

A hardness increase was detected in the first 600 nm and a maximum hardness value at about 8.5 GPa is then obtained after N-implantation, which corresponds to an increasing of almost twice the hardness value of the bulk alloy. On the other hand, the N-implantation advantageously modified the wear property reducing the friction coefficient by a factor of 2.7 and the wear track by 85% after 200 cycles under a 0.25N load.

Lower corrosion and ion release rate for the N-implanted surface than for the non-implanted one was obtained, showing the beneficial effects of the implantation. Long term monitoring of the open circuit potentials proved a nobler behaviour of N-implanted Ti-25Ta-25Nb alloy due to the barrier properties of the nitride layer.

In vitro tests performed on human fetal osteoblasts indicated that both non-treated and N-implanted Ti-25Ta-25Nb alloy exhibited good level of cytocompatibility. Furthermore, our data show that the tested N-implanted surface can play an active role in tissue biocompatibility resulting in an enhancement of biological properties.

Consequently, this work shows that a N-implanted superelastic Ti-based alloy is very useful for biomedical applications due to the observed better corrosion resistance, excellent biocompatibility and desirable mechanical properties combining the bulk superelastic property and the hard and wear resistant surface. This whole of properties can be particularly appreciated for various medical devices such as cardiovascular stents, dental implants, orthodontic arcs, neurologic clips, orthopaedic prosthesis ...

Acknowledgements

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Figure captions

Fig. 1. Stress-strain tensile curve of the superelastic Ti-25Ta-25Nb alloy (from [31]).

Fig. 2. GIXRD patterns of non-treated and N-implanted Ti-25Ta-25Nb alloy.

Fig. 3. XPS depth profiles of N 1s, Ti 2p, Nb 3p and Ta 4d excitation states at a N-implanted Ti-25Ta-25Nb alloy

Fig. 4. Negative ion depth profiles on a N-implanted Ti-25Ta-25Nb alloy.

Fig. 5. Hardness profiles of non-treated and N-implanted Ti-25Ta-25Nb alloy.

Fig. 6. Friction coefficient of non-treated and N-implanted Ti-25Ta-25Nb alloy, as a function of the number of cycles and optical micrographs showing the wear tracks after 200 cycles under 0.25N.

Fig. 7. Cyclic potentiodynamic curves for Ti-25Ta-25Nb alloy.

Fig. 8. Monitoring of open circuit potentials for Ti-25Ta-25Nb alloy.

Fig. 9. Comparative evaluation of LDH activity expressed in culture media by hFOB 1.19 cells grown on non-treated and N-implanted Ti-25Ta-25Nb.

Fig. 10. Fluorescence micrographs of hFOB 1.19 cells grown on non-treated and N-implanted Ti-25Ta-25Nb surfaces. Staining with phalloidin-TRITC to detect F-actin filaments (red) and DAPI to label the nuclei (blue).

Fig. 11. Representative data of MTT cell proliferation assay. Formazan absorbance (OD 550 nm) was expressed as a measure of cell proliferation at different time points over the 7-day incubation period. Results are presented as means \pm SD (n = 3). *p < 0.01 vs. hFOB 1.19 cells grown on non-treated Ti-25Ta-25Nb; **p < 0.001 vs. hFOB 1.19 cells grown on non-treated Ti-25Ta-25Nb

Fig. 12. Fluorescence micrographs showing the immunoreactivity of fibronectin synthesized, secreted and organized into an extracellular fibrillar network by hFOB 1.19 cells grown on non-treated and N-implanted Ti-25Ta-25Nb.

Table captions

Table 1. Main electrochemical parameters for Ti-25Ta-25Nb alloy in simulated human biofluids

Table 2. Corrosion and ion release rates for Ti-25Ta-25Nb alloy in simulated human biofluids

Table 3. Monitoring of the open circuit potential gradients for Ti-25Ta-25Nb alloy

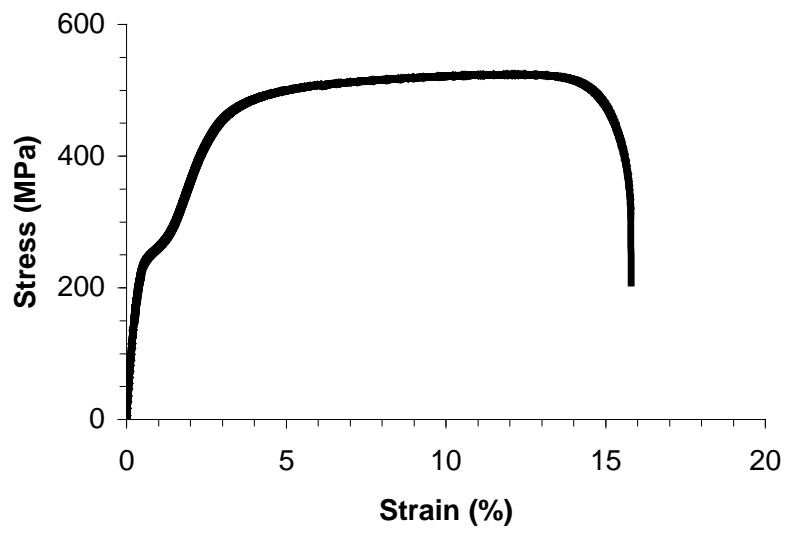


Fig.1

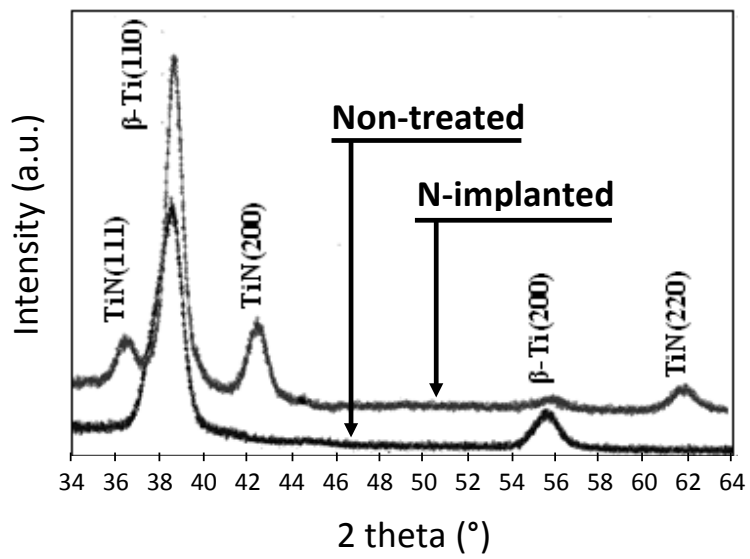


Fig.2

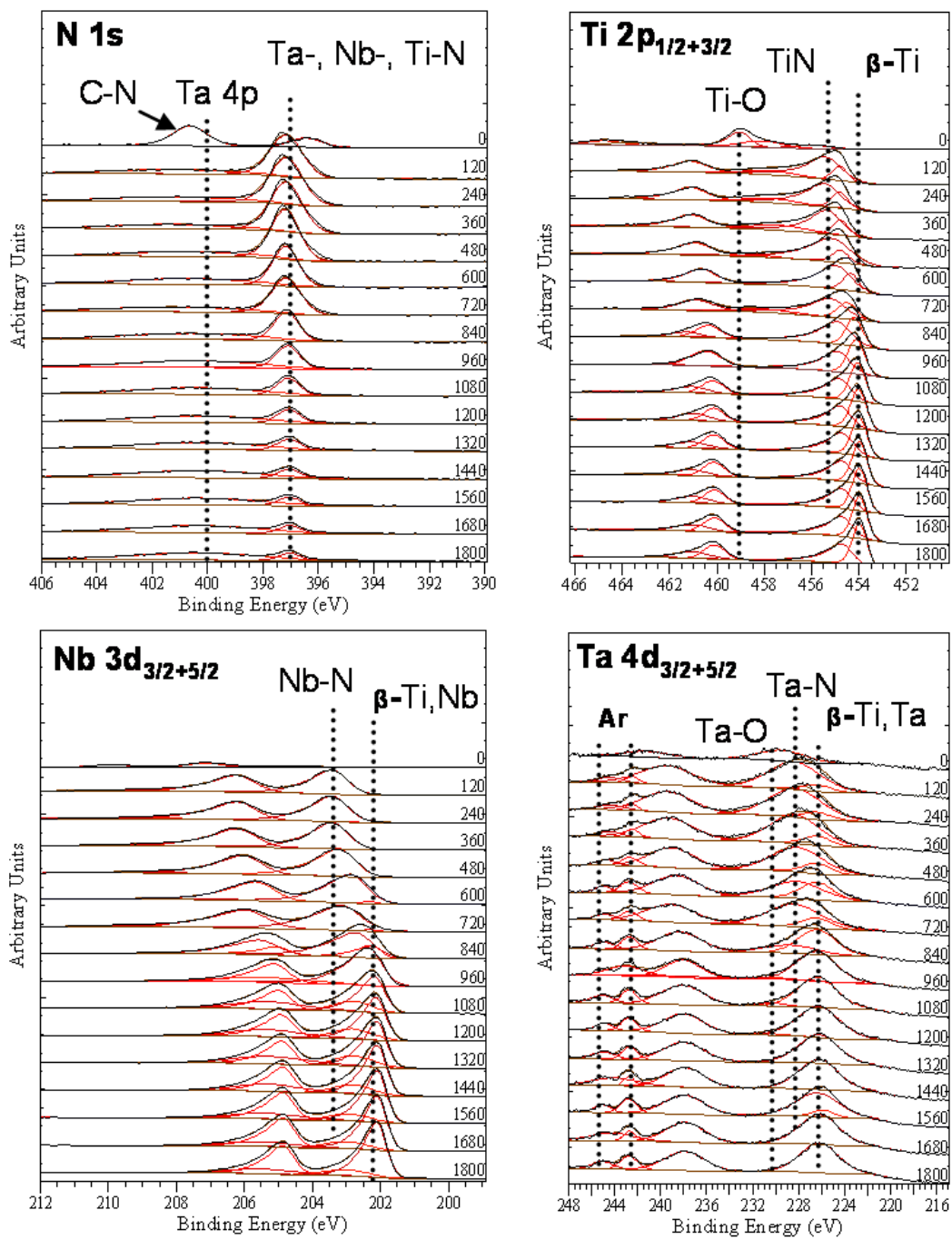


Fig.3

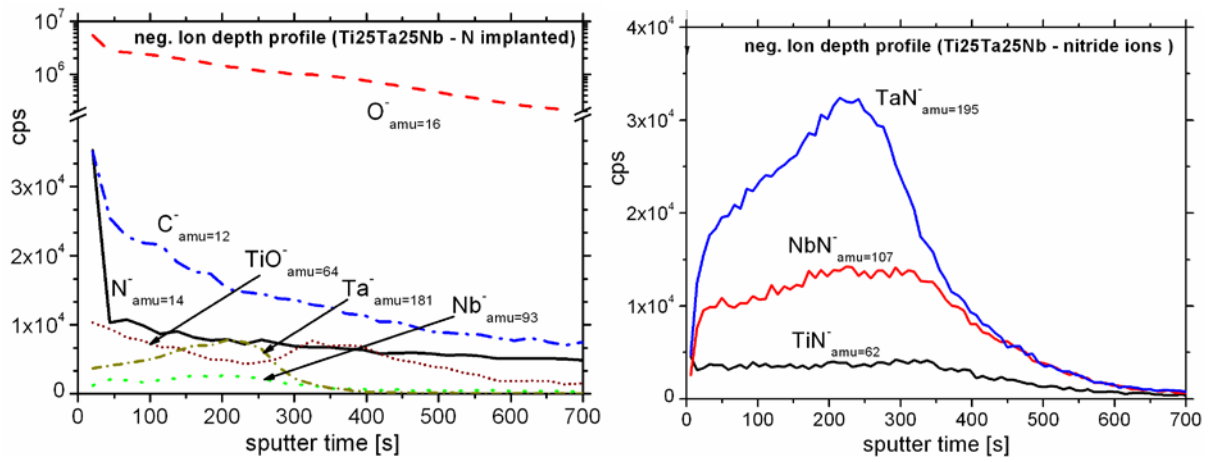


Fig.4

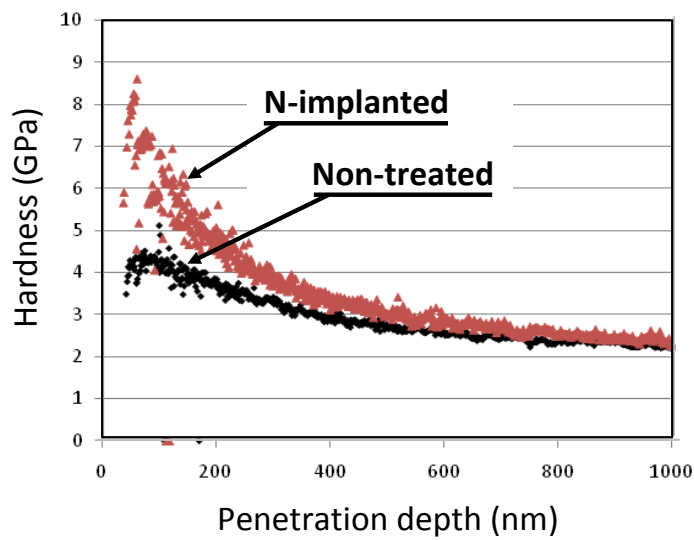


Fig.5

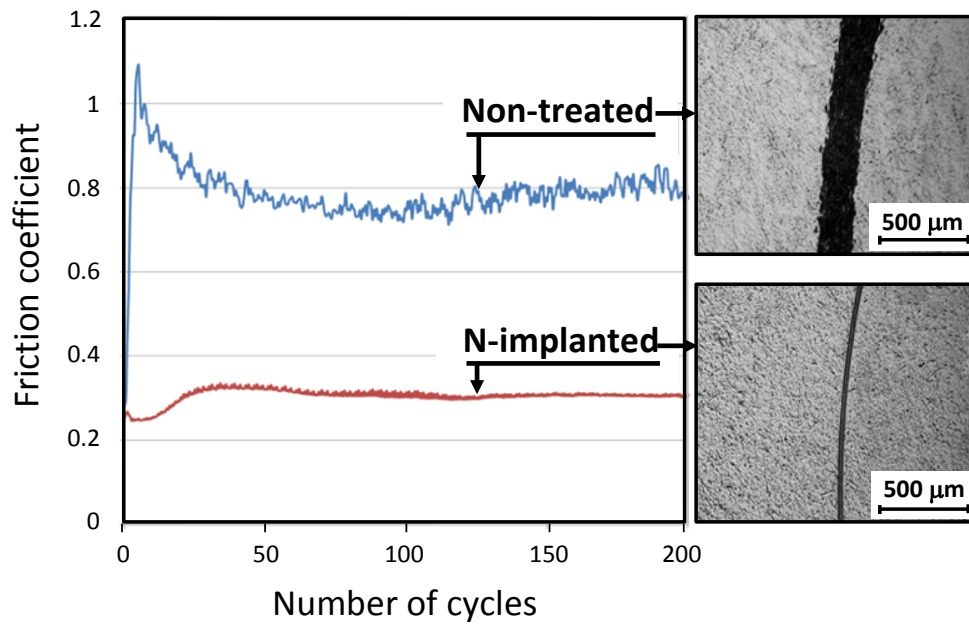


Fig.6

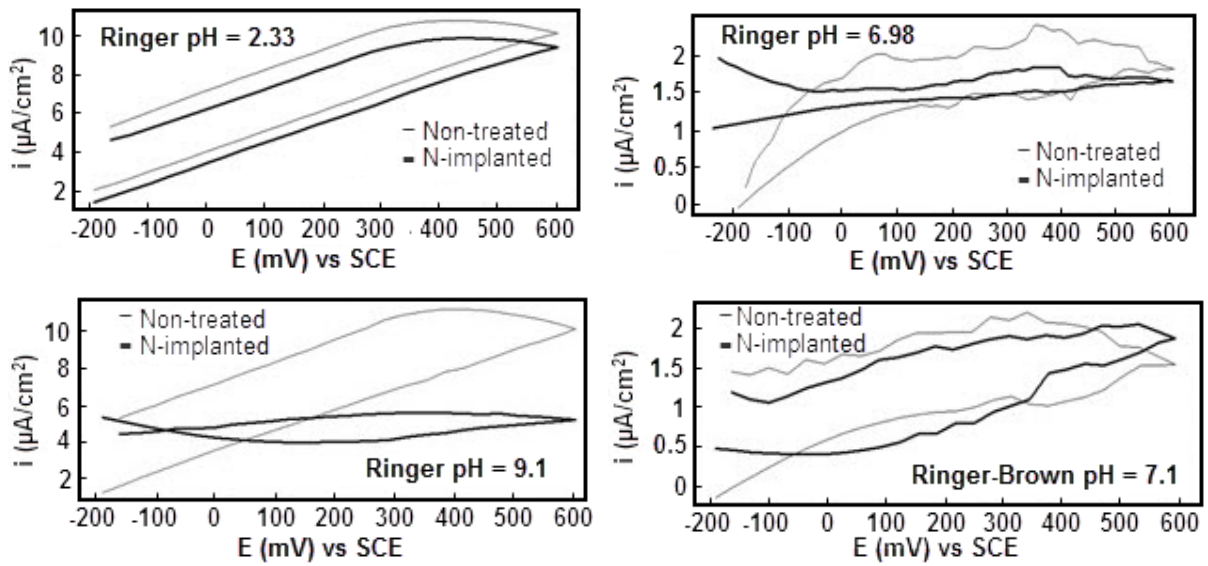


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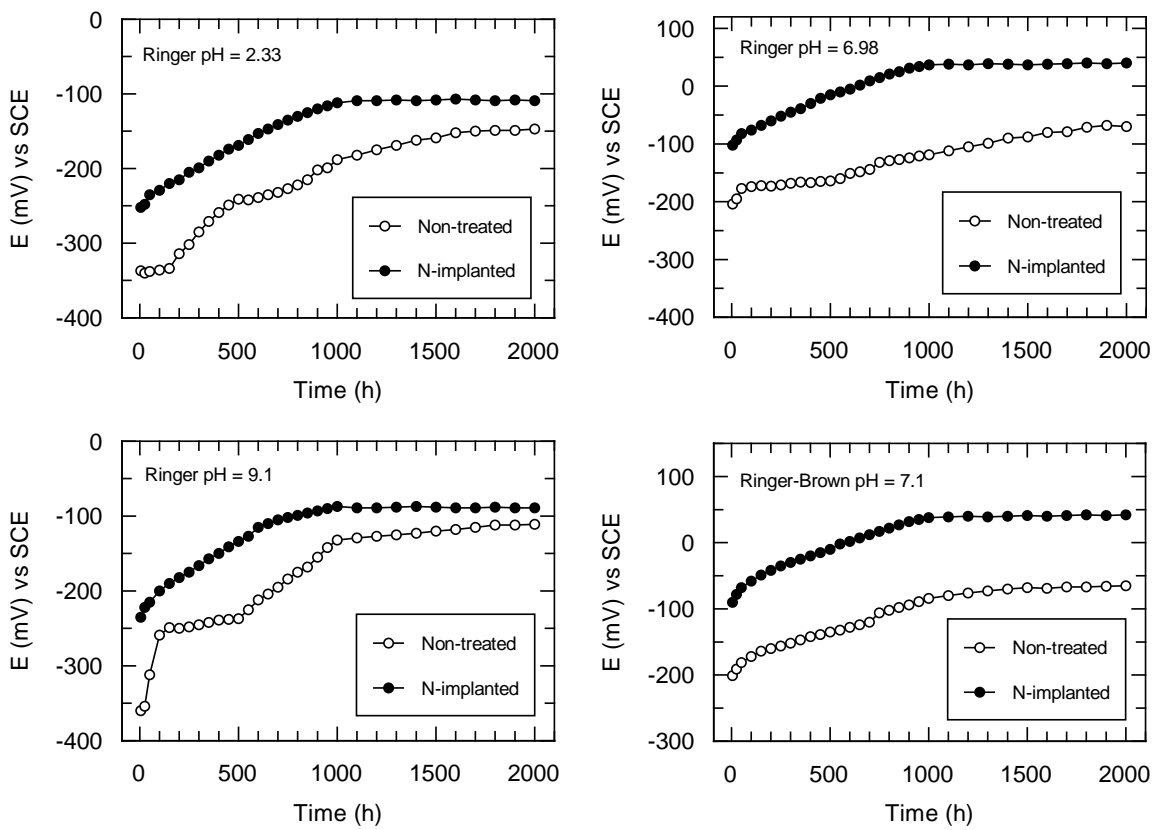


Fig. 8

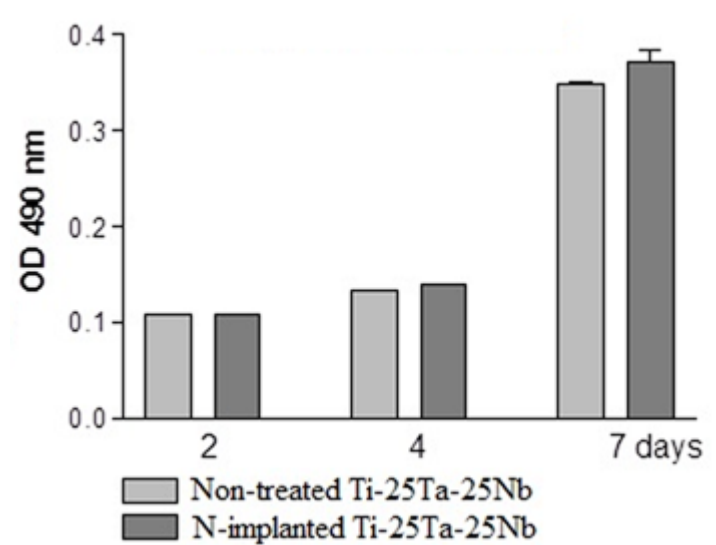


Fig. 9

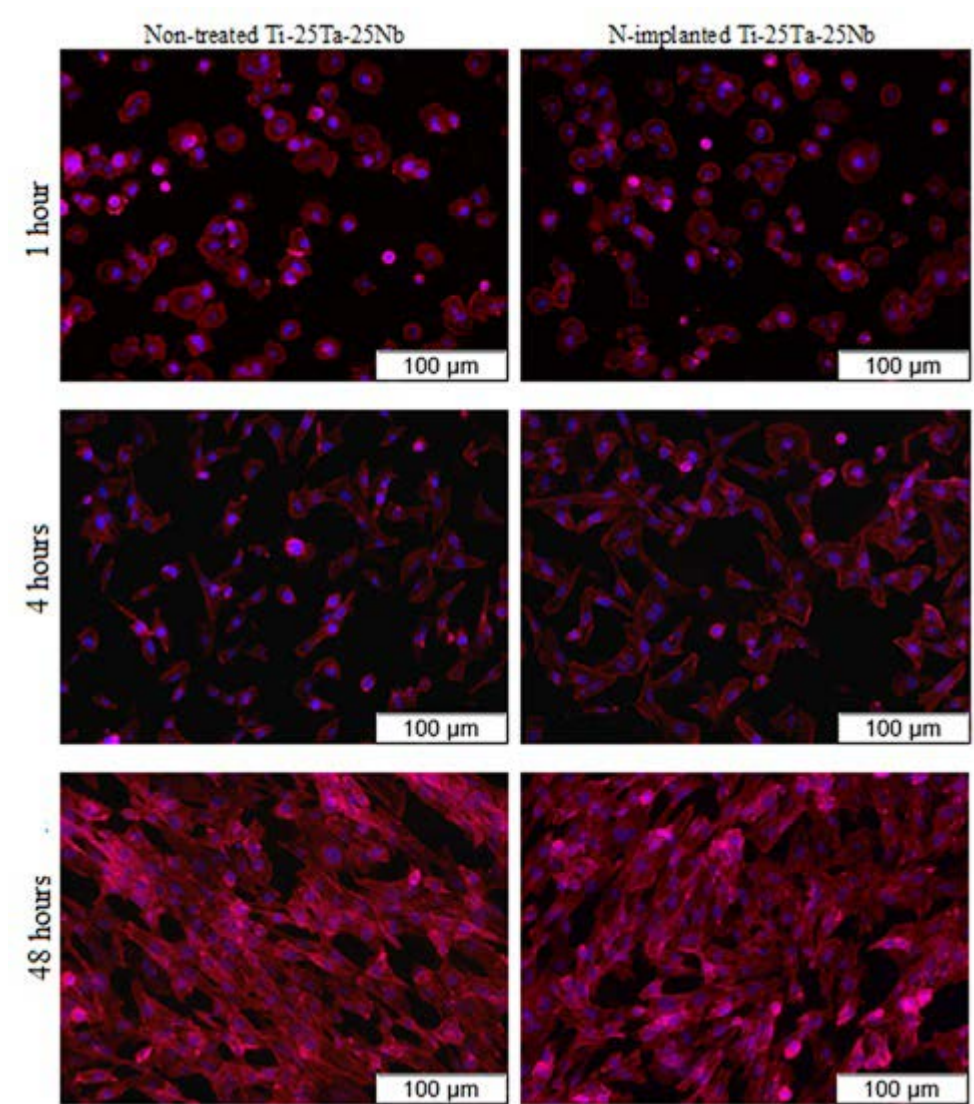


Fig. 10

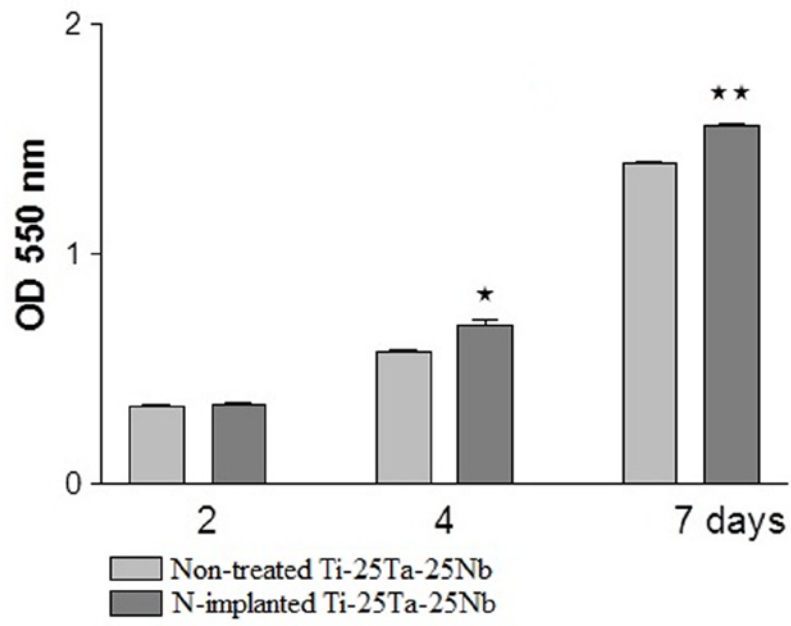


Fig. 11

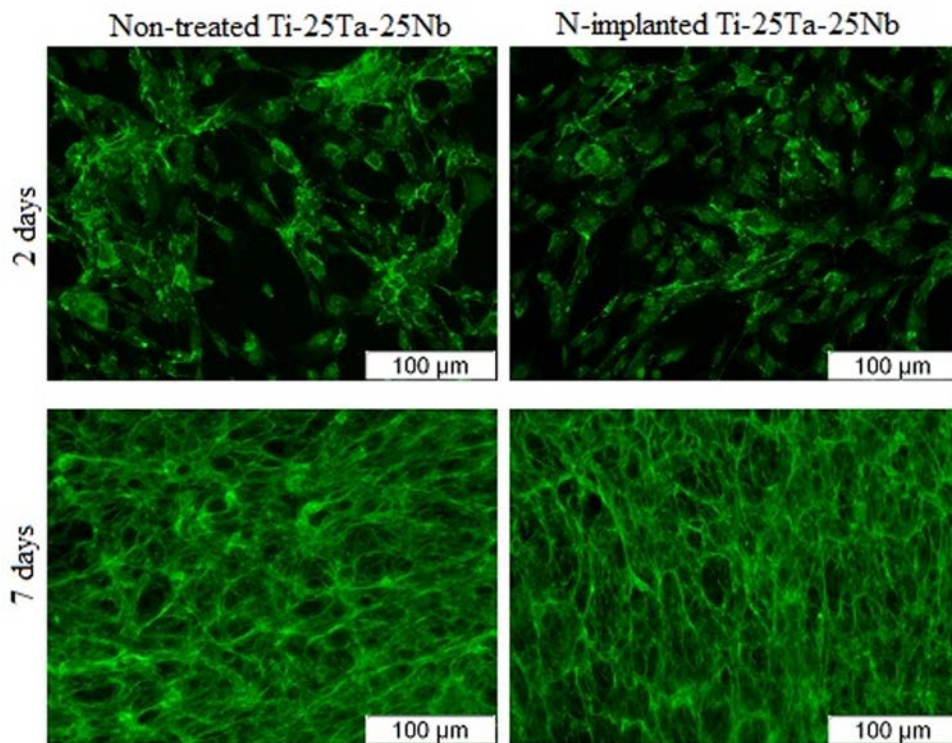


Fig. 12

Table 1

Alloy	E_{corr} (mV)	E_p (mV)	ΔE_p (mV)	$ E_{\text{corr}} - E_p $ (mV)	i_p ($\mu\text{A}/\text{cm}^2$)
Ringer pH = 2.33					
Non-treated	-369.4	-340	>600	29.4	8.2
N-implanted	-353.1	-330	>600	23.1	4.8
Ringer pH = 6.98					
Non-treated	-279.6	-240	>600	34.6	2.0
N-implanted	-182.2	-160	>600	22.2	1.3
Ringer pH = 9.1					
Non-treated	-276.7	-230	>600	46.7	7.9
N-implanted	-248.8	-220	>600	28.8	3.8
Ringer-Brown pH = 7.1					
Non-treated	-251.2	-220	>600	31.2	2.0
N-implanted	-153.5	-130	>600	23.5	1.2

Table 2

Alloy	i_{corr} ($\mu\text{A}/\text{cm}^2$)	V_{corr} ($\mu\text{m}/\text{Y}$)	Resistance class	R_p ($\text{k}\Omega \text{cm}^2$)	Ion release (ng/cm^2)
Ringer pH = 2.33					
Non-treated	0.02560	0.2270	PS	560	23.06
N-implanted	0.00607	0.0539	PS	738	5.470
Ringer pH = 6.98					
Non-treated	0.01397	0.1240	PS	1350	12.59
N-implanted	0.00225	0.0190	PS	1430	1.93
Ringer pH = 9.1					
Non-treated	0.01496	0.1339	PS	883	13.60
N-implanted	0.00363	0.0322	PS	1090	3.27
Ringer-Brown pH = 7.1					
Non-treated	0.00276	0.0245	PS	1573	2.48
N-implanted	0.00219	0.0195	PS	1850	1.98

PS – Perfect Stable

Table 3

Alloy	Time (h)	ΔE_{oc1} (mV)	ΔE_{oc2} (mV)	ΔE_{oc3} (mV)	ΔE_{oc4} (mV)
Non-treated	24	-145	+14	+159	-4
	500	-77	-4	+73	-29
	1000	-69	-56	+13	-35
	1500	+71	-39	-32	-20
	2000	-77	-36	-41	-5
N-implanted	24	-155	-26	+129	-15
	500	-154	-35	+119	-5
	1000	-149	-25	+124	-1
	1500	-71	-20	+125	+8
	2000	-69	-20	+129	+105