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1	Occurrence, distribution and fluxes of Benzotriazoles along
2	the German large river basins into the North Sea
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#### 18 **1** Abstract

19 Benzotriazole (BT) and tolyltriazole (TT) are high production volume chemicals which are used in various industrial and household applications. In this study, the 20 21 distribution of benzotriazoles in the estuaries of different rivers of central Europe and in the North Sea was analyzed by solid-phase extraction (SPE) and liquid 22 23 chromatography-mass spectrometry (HPLC-MS/MS). BT as well as TT were detected 24 in all water samples. The concentrations for total benzotriazoles (BTs) ranged from 25 1.7 to 40 ng/L in the North Sea in costal areas. Concentrations in rivers are from 200 to 1250 ng/L, respectively. The mass flux of total benzotriazoles from the major rivers 26 27 of central Europe into the North Sea was calculated to 78 t/a, dominated by the 28 Rhine with an individual flux of 57 t/a of BTs. The analysis of the distribution profile in 29 the North Sea showed that the decrease of the concentration was mostly caused by 30 dilution and that the benzotriazoles are poorly degradable in the North Sea. This 31 paper presents the first report of benzotriazoles in the marine environment.

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33 Keywords:

34 benzotriazole, tolyltriazole, river water, North Sea, HPLC-MS/MS

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### 36 2 Introduction

1H-benzotriazole (BT) and 1H-methyl-benzotriazole (tolyltriazole, TT, used as a technical mixture of 4- and 5-TT) are a class of high production volume chemicals (HPVC) with broad applications in various industrial processes as well as in households. They show metal complexing properties and are used as anticorrosive additives and flame retardants in aircraft de-icers and anti-ice fluids (ADAFs) (Gruden *et al.* 2001), in cooling and hydraulic fluids and for silver protection in dishwashing agents (Ort *et al.* 2005, Janna *et al.* 2011). The content of BTs in ADAFs varies

between 0.2 -1.7% (WIPO 2002). The annual production of BTs in the United States
was reported to be in the range of 9000 tons per year (Hem *et al.* 2003). The content
of Benzotriazoles in Dishwashing agents is reported with 0.09 mg per wash to 27.8
mg (Janna *et al.* 2011).

BTs are compounds with a low vapour pressure, high water solubility and a high polarity ( $\log K_{ow}$  1.44 for BT and 1.71 for TT, respectively) (US EPA 2010). Moreover, they are quite persistent against biological and photochemical degradation processes in the aquatic environment (Hart *et al.* 2004). Thus, they can be characterized as mobile in aquatic environment. They are classified as toxic to aquatic organisms as they can cause adverse long-term effects on the aquatic environments (Cancilla *et al.* 1998, Pillard *et al.* 2001, Cancilla *et al.* 2003, Hem *et al.* 2003).

In the environment benzotriazoles were firstly reported in subsurface water close to 55 56 airports as potential source in concentrations of 128 mg/L for BT, 17 mg/L for 4-TT 57 and 198 mg/L for total TT (Cancilla et al. 1998). Because of the broad applications, 58 BT and TT have high loads in waste waters treatment plants (WWTPs). The mean 59 concentrations in untreated wastewaters of WWTPs of Berlin were 12 µg/L, 1.3 µg/L and 2.1 µg/L for BT, 4-TT and 5-TT, respectively Weiss et al. (2006). The elimination 60 efficiency in WWTPs shows wide variations. In a study of 24 WWTPs of the Glatt 61 Valley, Switzerland elimination rates were reported between 13-62 % for BT and 23-62 74 % for the TT-isomers (Voutsa et al. 2006) while Weiss et al. (2006) determined 63 removals of 37 % for BT and 11% for 5-TT but no significant removal for the 4-TT-64 isomer. In sewage effluents of the UK were concentrations up to 3.6 µg/L for BT and 65 5.7 µg/L determinated (Janna et al. 2011). Several studies showed that the BT and 66 67 TT are widely distributed in surface waters (Giger et al. 2006, Reemtsma et al. 2006, Voutsa et al. 2006, Reemtsma et al. 2010). Loos et al. (2009) detected BT and TT in 68 94% and 81% of 122 river water samples distributed over the European Union (EU) 69

with mean concentrations of 493 ng/L and 617 ng/L for BT and TT, respectively. In an EU-wide study on groundwater, BT and TT were detected in mean concentrations of 1 and 4 ng/L, respectively (Loos *et al.* 2010a). Only one flux calculation is available for the river Danube into the Black Sea (Loos et al. 2010b). However, there are no data about concentration and distribution of BTs in the marine environment.

The objective of the present study was to investigate the occurrence and distribution of benzotriazoles in the German Bight and its major affluxes with respect to their spatial distribution as well as seasonal variations. Furthermore, the mass flux of BTs from the main rivers as sources of BTs into the North Sea was investigated. This study presents the first data on BTs in the marine environment and improves the understanding of the fate and the ecosystem risk assessment of BTs.

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82 **3 Experimental** 

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#### 84 **3.1 Chemicals**

1-H-benzotriazole (BT; 99%), 5-methyl-benzotriazole (5-tolyltriazole, 5-TT; 98%) and the deuterated 1H-Benzotriazole-d4 solution were purchased from Sigma-Aldrich (Steinheim, Germany). A tolyltriazole-isomer mixture was purchased from Dr. Ehrenstorfer (Augsburg, Germany). Acetone (Nanograde) and methanol (Picograde) were purchased from LGC Promochem (Wesel, Germany). Formic acid and hydrochloric acid were purchased from Merck (Darmstadt, Germany). Pure water was obtained from a Milli-Q system (Millipore, Billericia, MA, USA).

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#### 93 **3.2 Sample Collection**

Surface water samples (water depth <0.5m) were taken by stainless steel bucked in</li>
 pre-cleaned 1 L glass bottles on board of the research vessel *Ludwig Prandtl* along

the rivers Elbe and Weser during four campaigns in March, May, August and October 2010 with sampling intervals of 10 km. In addition, 15 spot samples were taken from the shore of the rivers Elbe, Weser, Ems, the Rhine-Meuse delta and the river Scheldt in August 2010. North Sea samples were taken on board of the research vessel *Heincke* during three campaigns in March, July and September 2010. The sampling stations are shown in Figure 1. Additional information on sampling are included in the supplementary material.

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105 Figure 1 – Map showing the sampling positions in the investigated rivers and the North Sea

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#### 108 **3.3 Sample Preparation**

109 Solid phase extraction (SPE) was used for compound extraction and enrichment from 110 the water samples. The unique river samples were filtered through glass fiber filters 111 (GF-C, Whatman) while seawater samples were not filtered. SPE was performed with 112 a self-designed glass based setup to prevent contaminations from plastic tubings. 113 500 – 700 mL surface water were acidified with hydrochloric acid to pH 2 and spiked 114 with 20 ng of the internal standard 1-H-benzotriazole-d4. Oasis HLB cartridges 115 (500mg, 60µm, Waters, Milfort, USA) were conditioned with 15mL acetone, followed 116 by 15mL methanol and 5 mL acidified Milli-Q-Water. The water samples were passed through the pre-conditioned cartridges at a flow rate of 1-2 mL min<sup>-1</sup> and afterwards 117 118 washed with 5 mL acified Milli-Q-Water. The cartridges were dried under vacuum for 119 5 min. The analytes were eluted with 15 mL methanol. The extracts were roti-120 evaporated to 5 mL and further concentrated to a volume of 150 µL under a gentle 121 stream of preheated nitrogen. The particulate phase on the filters was extracted by 122 ultrasonic with 20 mL methanol two times for 15 min. The evaporation procedures 123 were the same as described for the cartridges.

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125 **3.4 Instrumental analysis** 

The extracts were analysed using high performance liquid chromatography-tandem mass-spectrometry (HPLC-MS/MS) in electrospray positive ionisation mode. A HP 128 1100 HPLC by Agilent Technologies was used for separation equipped with a Phenomenex Synergi Hydro RP 80A column and a suitable guard column (Phenomenex Synergi 2 I Hydro RP Mercury 20 2mm). The injection volume was 10  $\mu$ L. The mobile phases were 1% formic acid as ionisation aid in Millipore water and methanol, respectively. The operating flow was 200  $\mu$ L/min in gradient mode, starting

with 90% of water as mobile phase decreasing to 0% over 25 minutes, then keptconstant for the next five minutes.

135 Detection and quantification of the Benzotriazoles were carried out by an API 3000 136 triple-guadrupole mass spectrometer (Applied Biosystems/MDS SCIEX) with an 137 electrospray ionization interface in positive ionization mode. The mass spectrometer 138 was operated in a multiple reaction monitoring mode (MRM) with a dwell time of 15 139 ms. The ion spray voltage was set to 5000 V and the temperature of the source block 140 was 400° C. For identification, the mass transitions 120.1 to 65.2 for BT, 134.1 to 141 77.0 for TT and 124.2 to 68.9 for the internal standard BT-d4 were monitored. The 142 chromatographical separation of the tolyltriazole isomers was not possible in this 143 study, so the isomers were as sum considered.

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#### 146 **3.5 Quality control**

Quality control and quality assurance included the use of a mass labelled internal standard, the determination of recovery rates, of method and instrumental blanks as well as breakthrough and reproducibility experiments. Method blanks were prepared using 500 mL pre-cleaned Millipore water at every set of 8 samples together with the samples. Blank levels were below the method quantification limits (MQL). The total analytical error is calculated by 7,6% over the method.

Recovery rates of the mass labelled internal standard 1H-Benzotriazole-d4 in river water samples ranged between  $53 \pm 11\%$  for the winter season samples and  $40 \pm$ 6% for the summer season. The recovery for the seawater samples was  $69 \pm 10\%$ . The breakthrough of BTs was found to be negligible with <1%. The recovery displacements were caused by matrix suppressions and losses during the solvent

evaporation. The MQLs were calculated based on a signal-to-noise ratio of ten. The
resulting MQLs are 1.2 ng/L for BT and 0.4 ng/L for TT, respectively.

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#### 161 **4 Results and Discussion**

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#### 164 **4.1** Occurrence in the rivers Elbe and Weser

Benzotriazole and tolyltriazole were detected in the dissolved phase of all 165 166 investigated surface water samples. The individual concentrations of the four sampling campaigns along the two rivers are shown in Table 1. In the Elbe the 167 concentrations ranged from 24 ng/L and 21 ng/L in the outer estuary of Elbe up to 168 169 304 ng/L and 322 ng/L in the harbour of Hamburg for BT and TT, respectively. The 170 concentrations of BT and TT in the Weser ranged from 28 ng/L in the sea water influenced area up to 219 ng/L (BT) and from 21 ng/L to up to 454 ng/L (TT), 171 172 respectively. The maximum concentration was found in the City of Bremen. The 173 observed concentrations are comparable to Reemtsma et al. (2010), who reported concentrations of 480 ng/L for BT and 570 ng/L for the TT-isomers, respectively, in 174 175 the Elbe downstream of the city of Hamburg (km 638) in January 2006.

In addition to the dissolved phase, the particulate phase of three samples taken in the turbidity maximum of the Elbe estuary (around km 689) was analyzed. The concentrations of BT and TT in the particulate phase were found to be less than 1% of the dissolved phase or even below the MQL. Based on these results, the filters of other stations were not analysed.

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Table 1: Concentrations of benzotriazoles in ng/L in Elbe (E) und Weser (W). (--) = not sampled; station names are stream kilometres; NE1-NE3 are positions in the North Sea; exact positions are included in the supplementary data 

Station	Ma	rch	Ма	ay	August		October	
	BT [ng/L]	TT [ng/L]						
E570	142	191	139	231				
E579	155	281	142	268				
E589	139	270	136	159				
E599	140	279	138	246				
E609	118	229	134	209	193	142	93	64
E619	142	213			209	214	92	49
E624	134	195	136	234	228	173	98	71
E629	157	239	152	231	254	196	91	72
E639	179	323	141	177	268	195	120	101
E649	135	243	139	123	304	222	164	138
E659	156	250	128	109	257	194	163	154
E669	139	231	134	130	212	195	162	169
E679	137	235	127	133	180	186	162	146
E689	114	223	121	103	155	125	160	152
E699	103	176	115	126	144	113	148	119
E709	115	228	102	81	110	122	120	84
E719	97	180					105	93
NE1	101	203	75	114			92	73
NE2	48	74	51	85			57	41
NE3	25	37	24	34			29	21
W2	125	208	215	454			169	136
W12	138	190	219	325			186	192
W22	128	193	215	517			168	159
W32	117	146	207	426			179	180
W42	114	175	199	381			198	155
W52	134	233	188	395			203	221
W62	113	195	156	211			182	232
W72	109	160	106	189			107	133
W90	70	128	69	158			63	54
W110	31	53	39	65			28	21
1	1		1		1		1	

#### **4.2** Mass fluxes of benzotriazoles in the Elbe

For an improved comparison of the occurrence of benzotriazoles among the different campaigns, riverine mass fluxes of BTs were calculated. They consider the water discharge variations caused by rain events and dry periods. Mass fluxes for the Elbe were calculated based on the sum concentrations of BT and TT and the daily water discharges with a correction factor for sea water influence. The calculation based on the following equation (1),

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$$f = \left(\frac{c}{(1 - \frac{S_{sample}}{S_{sea}})}\right)^* Q_{day}$$
(1)

198 where *f* is the mass flux [mg/s], c is the measured concentration [ng/L], S<sub>sample</sub> means 199 salinity at sampling point in PSU (Practical Salinity Units), Ssea means salinity in 200 seawater (35 PSU) and Q<sub>dav</sub> is the water discharge [m<sup>3</sup>/s] on the sampling day. The 201 water discharge data are supplied from the FFG ELBE. The salinity of the river Elbe 202 is negligible (<1 PSU) compared to the Salinity in the North Sea. Hence, the salinity 203 of the River was in the calculation not considered. The results are shown in Figure 2. 204 The mean mass fluxes were 450 mg/s, 194 mg/s, 232 mg/s and 474 mg/s in March, 205 May, August and October 2010, respectively. The high flux in March, which was constant along the entire sampling course along the river, was probably caused by 206 207 the first spring discharge after a strong winter season with high usage of 208 benzotriazoles containing de-icing agents during winter.

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In March the median concentrations were very similar to the concentrations in May
and August. The variations of the mass flux are caused by different water discharges,
a constant flux level is not observed.

213 An elevated input in the area of the city of Hamburg was observed in August and 214 especially in October. On this input based the high averages in October. A possible 215 reason for this increase could be a discontinuous emission by a WWTP or another 216 source located in the harbour area of Hamburg. In contrast, no influence of the city of 217 Hamburg was observed in March and May. In October, to the outer estuary the high 218 mass flows disappear. Probably there leave the sampling transect a waste water 219 plume, which is responsible for the high contamination downstream from Hamburg. 220 In all cruises slightly decreasing mass fluxes were observed downstream of the city 221 of Hamburg. This could be caused by dilution with fresh waters of inflows of tributary 222 streams and surface run off. Based on the study in part 4.6. degradation processes are implausible. The exchange with other compartments like air or sediment seems 223 not to be an important process for BTs, due to their physical-chemical properties. 224





227 Figure 2: Calculated benzotriazole mass fluxes [mg/s] of the river Elbe

#### **4.3** Occurrence of benzotriazoles in the tributary rivers of the North Sea

229 Spot samples taken in the estuaries of the tributary rivers were investigated. The 230 concentrations of BTs in the investigated affluxes of the North Sea are shown in 231 Table 2. The concentrations, which were not influenced by seawater, ranged from 232 172 ng/L for BT and 232 ng/L for TT in the river Elbe, and up to 390 ng/L (BT) and 233 1114 ng/L for TT in the river Scheldt. Especially upstream of the city of Antwerp, the 234 river Scheldt is in comparison to other lage rivers with 1500 ng/L for benzotriazoles 235 highly contaminated. In the Rhine and its anthropogenic modified delta the concentrations ranged from 155 – 369 ng/L for BT and 184 – 431 ng/L for TT. In the 236 237 rivers Ems and Weser, concentrations were within 278 – 291 ng/L for BT and 312 – 357 ng/L for TT, which are relatively comparable to those in the Elbe and Rhine. 238 239 Obviously the concentrations of BTs decreased towards the mouth from Ems-1 to 240 Ems-2 and Weser-W2 to Weser-W90, even under consideration of the seawater 241 influence. It is assumable that this is due to dilution processes from a diffuse inflow of 242 the rural surrounding areas and probably degradation processes occurring.

In comparison to other studies for BTs in surface waters, the results in this study showed relatively similar concentrations. In an European survey on more than 100 rivers from 27 countries *Loos et al. (2009)* found median concentrations of 226 ng/L for BT and 140 ng/L for TT, respectively. In the river Main, which is influenced by discharge of the Frankfurt airport, median concentrations were 132 ng/L for BT and 162 ng/L for TT, respectively.

Reemtsma et al. (2010) reported increasing concentrations along the rivers Elbe (by means of 5 samples) and Rhine (by means of 4 samples) over distances of 650 and 700 km, respectively. In the present study their results were affirmed. This increasing of concentration could be caused by an industrial input of the high industrial density along estuaries due to the logistical connection from rivers and the sea and a

accumulation along the rivers. This indicate a bad degradation of the investigatedcompounds.

256 Especially noticeable in the present study is that the concentration of TT in relation to 257 BT is elevated. In studies in Switzerland (Giger et al. 2006, Voutsa et al. 2006) as 258 well as in the European survey study on polar organic contaminations (Loos et al. 259 2009) higher median concentrations for BT than TT are reported. In the study on the 260 Rhine and Elbe (Reemtsma et al. 2010), a decrease of the BT/TT ratio along the 261 rivers was observed. Since 5-TT is known to be more degradable than BT and 4-TT (Weiss et al. 2006), this effect is likely not related to degradation processes. A 262 possible explanation could be different usages of benzotriazoles in the upper reaches 263 264 regions of the rivers than in the lower reaches regions.

Table 2: Concentrations of benzotriazoles of the North Sea feeder rivers in ng/L; exact
 positions are included in the supplementary data

River Sample station		BT	TT	Salinity	Pos	ition
			[ng/L]	[PSU]	E	N
Elbe	E670 - Glückstadt	172	232	0.6		
Libe	LUT9 - Gluckstaut	172	232	0,0	9,40	53,79
Elbe	E 712 - Cuxhaven	55	77	20,1	8.71	53.87
Weser	W 90 - Bremerhaven	105	126	19,8	0,71	50,01
Weser	W2 - Hemelingen	278	357	0,5	8,58	53,53
Emo	Ema2 Emdan	146	200	10.7	8,86	53,06
EIIIS	EIIIS2 - EIIIdell	140	200	10,7	7,31	53,32
Ems	Ems1 - Herbrum	291	312	0,6	7.32	53.03
Rhine - IJssel	R1 - Kampen	369	332	0,0	5.00	50,50
Rhine - North Sea Canal	R2 - Ijmuiden	155	184	10,0	5,92	52,50
Rhine - New Waterway	P3 - Maassluis	246	131	1 1	4,61	52,47
Rinne - New Waterway	10 - 1000351015	240	431	1,1	4,25	51,91
Rhine - Hollands Diep	R4 - Bovensluis	241	428	0,1	4.41	51.70
Rhine - GER/NED border	R5 - Lobith	262	378	0,0	6 10	E1 9E
Waal	R6 - Andelst	233	342	0,0	6,10	51,05
Mouso	P7 Mogon	307	417	0.1	5,74	51,88
Meuse	Kr - Megen	397	417	0,1	5,57	51,83
Scheldt	S2 - Schaar van Ouden Doel	180	374	16,7	4.21	51.40
Scheldt	S1 - Schelle	390	1114	3,6	4.00	51.10
					4,33	51,13

#### **4.4** Mass fluxes of European Rivers into the North Sea

271 The mass flux estimations are based on one sampling campaign in August 2010 272 only. Mass fluxes were calculated by equation (1) with the modification that the 273 average annual water discharges were used. The mass fluxes are shown in Figure 3. 274 The overall total input from all investigated rivers into the North Sea was calculated 275 with 30 t/a for BT and 47 t/a for TT, respectively. The dominate source is the Rhine with his 4 main inflows into the North Sea (North Sea Canal, New Waterway Canal, 276 277 lissel and Hollands Diep/Haringvliet) with approximately 22 t/a for BT and 35 t/a for TT, respectively. This accounted for 72% of the discharge of the investigated rivers. 278

Seasonal variations are not considered in this estimation. In a previous study, (Giger *et al.* 2006) showed no significant seasonal trends in the Rhine. The seasonal variations of the mass flux of BTs from the Elbe to the North Sea estimated in this study is correlated to the water discharge. For more specified conclusions of the seasonal trends is a details analyses with smaller time intervals required.



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Figure 3: Estimated mass fluxes in t/a of European rivers into the North Sea; Triangles show the water discharge of the river

#### **4.5 Occurrence in the North Sea**

BT and TT were detected above the MQL in all seawater samples from the German Bight. The results are shown in Figure 4. In the coastal area concentrations up to 21 ng/L and 37 ng/L were found for BT and TT, respectively. In the open sea in approximately 300 km distance to the coast, concentrations of BT and TT were 1.4 ng/L and 1.1 ng/L, respectively, which are just above the MQL.

294 The main current conditions in the German Bright are dominated on an eastern 295 current the "Continental Coastal Water". This transported water masses from the Dutch Coast into the German Bright and turned than to the north (Turrell 1992). The 296 297 benzotriazoles distribution in the North Sea indicates an input from the rivers Elbe and Weser from the south-east into the German Bright. The high contamination in 298 299 front of the East Frisian Islands can be traced back to the input of the rivers Rhine 300 and Scheldt. The contamination in the area of the North Frisian Islands is based of 301 the input over the Rhine and Scheld as well as from the Elbe and Weser. The 302 samples from the open sea were mainly influenced by Atlantic Ocean water 303 streamed along the Scottish Coast as central and south North Sea waters into this 304 area (Turrell 1992).

A slight seasonal variation was observed. In comparison to the summer in spring high concentrations were detected in the coastal area, with a strong decreasing trend towards the open sea. In summer the concentrations in the coastal area were slightly lower, but interestingly, the open sea stations showed higher concentrations in summer than in spring. This is against the theory that the highest photo- and biodegradation is in summer. In autumn the trend followed the summer pattern.

The contamination of Benzotriazoles in the North Sea is relatively high in comparison to other organic pollutants. For example, concentrations of polyfluoroalkyl compounds (PFCs) such as perfluorooctanoic acid (PFOA) and perfluorooctane

sulfonate (PFOS) are one order of magnitude lower (Ahrens *et al.* 2009),
concentrations of the endocrine disruptor Triclosan are three orders of magnitude
lower as benzotriazoles (Xie *et al.* 2008). The sum concentrations of
organophosphate flame retardants (OPE) showed comparable concentrations to
single concentrations of BT and TT (Andresen *et al.* 2007).

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321 Figure 4: Distribution of benzotriazoles in the North Sea in ng/L

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### 323 **4.6** Study of dilution and persistency

To investigate if decreasing concentrations are based on degradation or merely on dilution, a comparison with salinity is useful. The correlation of dilution versus salinity was assessed with utilization of the equations (2) and (3).

(2)

329 Normalized concentration (NC) = concentration sampling point / concentration freshwater (3)
 330

331 The salinity of sea water was assumed with 35 PSU. The freshwater concentration 332 was set to the last sampling point in the estuary with a salinity < 1 PSU.

Figure 5 displays the dilution factor (DF) at each sampling point with sea water influence (salinity > 1 PSU) of the March and September/October sampling in the Elbe and the North Sea in comparison to the normalized concentration (NC) for BT and TT.

337 The mostly linear observed relationship signifies that the decrease of the 338 concentration is mainly due to dilution with seawater. Only for TT in autum are, a 339 slight degradation can be ascertained in autumn, because the most points are under 340 the line. This can be explained by higher degradability of the 5-TT isomer than BT 341 and the 4-TT isomer (Weiss et al. 2006). In both graphs the sampling station a point 342 E719 for TT is located in the front of the harbour of Cuxhaven and has assumedly a 343 local source of tolyltriazole. Based on this analysis, the seasonal variations which are described in chapter 4.5 were caused by seawater currents and dilution processes 344 rather than degradation. 345



Figure 5: Correlation of dilution factor and normalized concentrations for benzotriazole and
 tolyltriazole in the river Elbe and the North Sea

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### 351 **5** Conclusions

Benzotriazoles are poorly degradable polar organic pollutants which are present in the anthropogenic water cycle. This study has shown that approximately 80 tonnes per year of these chemicals are discharged via the investigated rivers into the North Sea, mainly via the river Rhine. Further it is shown that the decrease of the concentration in the German Bight is mostly attributed to dilution.

On the one hand the toxicity of these substances is reported to be moderate, the bioaccumulation potential is low and the detected concentration are two magnitude lower as the chronic predicted no effect concentrations. But on the other hand the benzotriazoles are observed concentrations of benzotriazoles in the aquatic environment rank among the highest in the group of polar organic pollutants in the water cycle. Thus the widespread distribution in European lakes, groundwater, rivers and the North Sea in addition to the shown persistence in aquatic samples has to be noted. Further studies have to focus on the occurrence and degradation of the
 benzotriazoles in the marine water cycle, especially separate investigation of the
 tolyltriazole isomers.

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# 456 Supplementary data

# 

# 458 Table S1: Additional data of North Sea sampling campaigns

Station		Pos	ition	Marc	:h camp	aign	July	y campa	lign	S	eptembo ampaig	er n
	Acronym	Longitud e [E°]	Latitude [N°]	Date	Time	Salinity [PSU]	Date	Time	Salinity [PSU]	Date	Time	Salinity [PSU]
Tiefe Rinne	N1	7,900	54,142	04.03. 2010	17:30	31,80	13.07. 2010	16:39	32,5	-	-	-
Eiderstetdt	N2	8,250	54,227	04.03. 2010	19:40	29,47	-	-	-	18.09. 2010	09:52	29,4
Deutsche Bucht	N3	7,504	54,225	-	-	-	-	-	-	19.09. 2010	09:17	31,3
Amrum	N4	8,113	54,680	08.03. 2010	10:30	30,75	15.07. 2010	10:28	30,9	18.09. 2010	14:07	30,4
List	N5	8,250	55,050	08.03. 2010	14:20	31,25	-	-	-	18.09. 2010	17:32	30,7
Langeoog	N6	7,500	53,817	05.03. 2010	23:30	31,84	15.07. 2010	21:49	31,4	19.09. 2010	12:57	31,44
Borkum	N7	6,600	53,702	05.03. 2010	18:18	30,82	-	-	-	19.09. 2010	19:50	31,43
Emssüd	N8	6,360	54,225	05.03. 2010	12:30	33,95	16.07. 2010	13:17	33,3	-	-	-
Fino 1	N9	6,600	54,000	-	-	-	16.07. 2010	09:08	35,2	19.09. 2010	22:08	33,3
NSB 2	N10	6,335	55,005	06.03. 2010	18:10	34,64	16.07. 2010	21:44	34,1	-	-	-
NSB 3	N11	6,732	54,680	-	-	-	19.07. 2010	06:44	33,5	20.09. 2010	07:38	33,4
Nordsee 1	N12	6,317	55,402	09.03. 2010	01:15	34,42	17.07. 2010	12:42	34,1	-	-	-
Nordsee 2	N13	7,500	54,680	06.03. 2010	09:00	33,92	15.07. 2010	14:24	31	19.09. 2010	04:06	32,3
Nordsee 3	N14	6,360	54,680	06.03. 2010	21:20	34,30	14.07. 2010	08:41	33,7	20.09. 2010	05:21	34
Nordsee 4	N15	7,500	55,150	06.03. 2010	13:00	34,63	19.07. 2010	15:18	32,4	18.09. 2010	22:14	32,3
Nordsee 6	N16	5,033	55,167	09.03. 2010	08:00	34,54	17.07. 2010	20:40	34,5	-	-	-
Enten- schnabel	N17	4,418	55,462	09.03. 2010	12:30	34,55	18.07. 2010	04:18	34,7	-	-	-

Table S2: Additionally data of estuary sampling campaigns

462	
463	

May (	campaig	ut	August	campa	ign	Octobe	r campa	lign
te	Time	PSU*	Date	Time	PSU	Date	Time	PSU
2010	12:22	ı	ı	ı	ı	I	ı	ı
2010	11:29	ı	ı	ı	ı	ı	ı	I
2010	9:28	ı	I	ı	ı	I	ı	ı
2010	8:50	ı	ı	ı	ı	ı	ı	ı
2010	8:00	,	10.08.2010	16:06	ı	06.10.2010	10:19	ı
2010	20:54	ı	10.08.2010	15:20	ı	06.10.2010	11:01	ı
2010	20:39	ı	10.08.2010	15:06	ı	06.10.2010	11:29	ı
2010	20:07	ı	10.08.2010	14:47	ı	06.10.2010	11:52	ı
2010	19:50	ı	10.08.2010	14:18	1	06.10.2010	12:45	ı
2010	19:30	ı	10.08.2010	13:50	ı	06.10.2010	13:30	ı
2010	19:02	ı	10.08.2010	13:22	ı	06.10.2010	14:11	ı
2010	18:28	ı	10.08.2010	12:54	ı	06.10.2010	15:01	ı
2010	18:01	0,61	10.08.2010	12:27	0,53	06.10.2010	15:37	0,62
2010	17:28	1,02	10.08.2010	12:00	1,11	06.10.2010	16:13	1,28
2010	17:00	2,45	10.08.2010	11:34	3,49	06.10.2010	16:39	3,47
2010	16:32	5,38	10.08.2010	11:04	7,44	06.10.2010	17:15	8,09
2010	16:01	8,87	10.08.2010	10:31	11,47	06.10.2010	17:40	13,8
2010	14:55	13,20	I	ı	I	07.10.2010	ı	14,60
2010	14:07	20,86	I	ı	I	07.10.2010	ı	23,20
2010	12:51	29,84	I	ı	ı	07.10.2010	ı	29,90
							ı	
2010	09:26	ı	I	ı	I	07.10.2010	ı	0,50
2010	10:08		I	ı	I	07.10.2010	ı	0,30
2010	10:42	ı	I	ı	I	07.10.2010	ı	0,30
2010	11:14	ı	I	ı	ı	07.10.2010	ı	0,30
2010	11:43	0,07	I	ı	I	07.10.2010	ı	0,70
2010	12:13	0,75	I	ı	I	07.10.2010	ı	3,00
2010	12:50	2,08	I	ı	I	07.10.2010	ı	7,70
2010	10:19	11,64	I	ı	ı	07.10.2010	ı	19,20

5:12

53,60 53,66

9,51

5:51

53,56

E639 E644 E649 E659

53,57

9,66 9,59

9,88 9,73 0,72 0,94 3,27 11,48

05.05.2010 05.05.2010 05.05.2010 05.05.2010

> 23,75 30,45

> > 11:14

11,61

22.03.2010 22.03.2010 22.03.2010

53,914

8,671 8,437

NE1 NE2 NE3

53,976

53,926

8,081

05.05.2010 05.05.2010 05.05.2010 05.05.2010 05.05.2010

2:25

24.03.2010 24.03.2010 24.03.2010 24.03.2010 24.03.2010 24.03.2010 24.03.2010

53,87 53,87 53,84 53,84

9,43 9,36 9,25 8,95 8,81

E669 E679 E679 E689 E699 E709 E719

0,74

0,77

3:55 3:08 1:35 0:49 09:53 13:35 12:37

53,73

53,81

4:32

Station	Posi	ition
	[]	[_N]
E570	10,54	53,37
E579	10,43	53,40
E589	10,30	53,43
E599	10,18	53,40
E609	10,07	53,46
E624	9,94	53,54
E629	9,88	53,54

06.05.2010 06.05.2010 06.05.2010 06.05.2010 06.05.2010 05.05.2010 05.05.2010 05.05.2010 05.05.2010 05.05.2010 05.05.2010

12:35 11:48

25.03.2010

Date

10:08

25.03.2010 25.03.2010 09:27 17:54

25.03.2010 24.03.2010 24.03.2010 24.03.2010 24.03.2010 24.03.2010 24.03.2010 24.03.2010

17:10 6:32 6:46

Date

March campaign late Time PSU\*

464 465

30,10

24,40

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> ı ı.

19,02 11,64

26,30

07.10.2010 07.10.2010 07.10.2010 07.10.2010 07.10.2010

12:13 12:50 10:19 11:07 12:08

03.05.2010

11:55

2:31

11:21

53,258

W32 W42 W52 W62 W72

53,344 53,433 53,517

03.05.2010

13:11

08:41

22.03.2010 22.03.2010

53,587 53,704

03.05.2010 03.05.2010 03.05.2010 03.05.2010 03.05.2010

0,58

09:52

22.03.2010

53,081 53,127 53,183

8,790

0,55

0:28 11:07

22.03.2010

8,655 8,555 8,484 8,500 8,493 8,545 8,525 8,332

W2 W12 W22

0,52 0,52 0,47 0,49 0,53 8,55

22.03.2010 22.03.2010 22.03.2010 22.03.2010 22.03.2010

05.05.2010

05.05.2010 05.05.2010

15,52

09:22 10:19

26,01

22.03.2010

824

53.

8,127

W110

06M

# Table S3: Additionally data of river sampling campaigns

River	Sample station	Acronym	Latitude [°E]	Longitude [°N]	Date	Local time	Hď	Tw[°C]	Salinity [PSU]	Discharge [m³/s]
Elbe	Glückstadt	E1	9,40	53,79	02.08. 2010	14:16	7,55	23,0	0,6	
Elbe	Cuxhaven	E719	8,71	53,87	02.08. 2010	16:30	8,08	19,3	20,1	712
Weser	Bremerhaven	W90	8,58	53,53	02.08. 2010	17:15	7,92	20,8	19,8	390
Weser	Hemelingen	W2	8,86	53,06	02.08. 2010	18:30	8,35	21,9	0,5	
Ems	Emden	Ems1	7,31	53,32	03.08. 2010	10:50	7,70	20,8	10,7	80
Ems	Herbrum	Ems2	7,32	53,03	03.08. 2010	12:03	7,53	21,2	0,6	
Rhine - IJssel	Kampen	R1	5,92	52,56	03.08. 2010	14:45	7,02	21,4	0,0	340
Rhine - North Sea Canal	ljmuiden	R2	4,61	52,47	03.08. 2010	17:30	7,92	22,4	10,0	95
Rhine - New Waterway	Maassluis	R3	4,25	51,91	03.08. 2010	19:20	7,92	21,3	1,1	1335
Rhine - Hollands Diep	Bovensluis	R4	4,41	51,70	04.08. 2010	09:35	7,96	21,8	0,1	940
Rhein - GER/NED Border	Lobith	R5	6,10	51,85	04.08. 2010	17:20	8,18	21,4	0,0	
Waal	Andelst	R6	5,74	51,88	04.08. 2010	15:35	8,20	21,2	0,0	
Meuse	Megen	R7	5,57	51,83	04.08. 2010	14:30	8,00	21,8	0,1	
Scheldt	Schaar van Ouden Doel	S1	4,21	51,40	04.08. 2010	10:45	8,01	20,2	16,7	120
Scheldt	Schelle	S2	4,33	51,13	04.08. 2010	12:10	7,90	21,9	3,6	

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# 473 Table S4: Instrumental parameters474

## HLPC-MS/MS Parameter

Autosampler Modell:		Agilent 1100 Autosampler
Syringe Size:		100 μL
Injection Volume:		10 µL
Needle Level:		5 mm
Eject Speed:		200 µl/min
Dump Modell:	İ	Agilant 1100 LC Binary Dump
		Aglient 1100 LC binary Fullip
Equalisation time:		5 min
Gradient ramp		
Time		Solvent Ration
	0	90/10
	3	90/10
	5,5	50/50
	8	30/70
	12	10/90
	16	0/100
	24	0/100
	24,5	90/10
	25	90/10

## Mass Spectrometer

Туре	API 3000 Applied Biosystems/MDS Sciex
Scan Type	MRM
Polarity	positiv
Ion soure	Electronen spray ionization
MRM detection window	240 sec
Nebulizer Gas	10
Curtain Gas™	6
IonSpray Voltage	5000 V
Source Temperature	400°C

## **Compound-Dependent Parameters:**

Analyts	BT	TT	BT-d4
Q1 Mass	120.20	134.13	124.21
Q3 Mass	65.20	77.00	68.90
Retention time	14.00	15.21	14.04
Declustering Potential	6	11	11
Focusing Potential	190	190	180
Entrance Potential	33	35	35
Collision Energy	2	4	4

Table S5: Relative recovery of target compounds in preliminary tests; Samples were spiked with 100ng/L of each target compound and 20 ng/ of internal standard

Sample	BT	TT
Milli-Q Water	•	
Sample A	108%	94%
Sample B	96%	92%
Sample C	113%	105%
Avarage	105%	97%
Sea Water (A	arctic)	
Sample D	99%	89%
Sample E	112%	115%
Sample F	95%	119%
Avarage	102%	108%